

Québec Education Program Progression of Learning

Science and Technology Secondary III 2021-2022 School Year

Learning to Be Prioritized for the 2021-2022 School Year in the Context of the Pandemic

This document is identical to the one produced for the 2020-2021 school year.





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Progression of Learning in Secondary School

The progression of learning in secondary school constitutes a complement to each school subject, providing further information on the knowledge that the students must acquire and be able to use in each year of secondary school. This tool is intended to assist teachers in planning both their teaching and the learning that their students are to acquire.

The role of knowledge in learning

The knowledge that young people acquire enables them to better understand the world in which they live. From a very early age, within their families and through contact with the media and with friends, they accumulate and learn to use an increasingly greater body of knowledge. The role of the school should be to progressively broaden, deepen and structure this knowledge.

Knowledge and competencies must mutually reinforce each other. On the one hand, knowledge becomes consolidated when it is used and, on the other hand, the exercise of competencies entails the acquisition of new knowledge. Helping young people acquire knowledge raises the challenging question of how to make this knowledge useful and durable, and thus evokes the notion of competency. For example, we can never be really assured that a grammar rule has been assimilated until it is used appropriately in a variety of texts and contexts that go beyond the confines of a repetitive, targeted exercise.

Intervention by the teacher

The role of the teacher in knowledge acquisition and competency development is essential, and he or she must intervene throughout the learning process. In effect, the *Education Act* confers on the teacher the right to "select methods of instruction corresponding to the requirements and objectives fixed for each group or for each student entrusted to his care." It is therefore the teacher's responsibility to adapt his or her instruction and to base it on a variety of strategies, whether this involves lecture-based teaching for the entire class, individualized instruction for a student or a small group of students, a series of exercises to be done, a team activity or a particular project to be carried out.

In order to meet the needs of students with learning difficulties, teachers should encourage their participation in the activities designed for the whole class, although support measures should also be provided, when necessary. These might involve more targeted teaching of certain key elements of knowledge, or they might take the form of other specialized interventions.

As for the evaluation of learning, it serves two essential functions. Firstly, it enables us to look at the students' learning in order to guide and support them effectively. Secondly, it enables us to verify the extent to which the students have acquired the expected learning. Whatever its function, in accordance with the *Policy on the Evaluation of Learning*, evaluation should focus on the acquisition of knowledge and the students' ability to use this knowledge effectively in contexts that draw upon their competencies.

Structure

The progression of learning is presented in the form of tables that organize the elements of knowledge similarly to the way they are organized in the subject-specific programs. In mathematics, for example, learning is presented in fields: arithmetic, geometry, etc. For subjects that continue on from elementary school, the *Progression of Learning in Secondary School* has been harmonized with the *Progression of Learning in Elementary School*. Every element of learning indicated is associated with one or more years of secondary school during which it is formally taught.

A uniform legend is used for all subjects. The legend employs three symbols: an arrow, a star and a shaded box. What is expected of the student is described as follows:



An **arrow** indicates that teaching must be planned in a way that enables students to begin acquiring knowledge during the school year and continue or conclude this process in the following year, with ongoing systematic intervention from the teacher.

A **star** indicates that the teacher must plan for the majority of students to have acquired this knowledge by the end of the school year.

A **shaded box** indicates that the teacher must plan to ensure that this knowledge will be applied during the school year.

General Education Path

Introduction

This document provides additional information about the learning prescribed in the compulsory secondary-level Science and Technology programs and its progression from year to year and from cycle to cycle. This document is intended to help teachers with their lesson planning.

To progress in their learning, students need to do more than merely acquire knowledge. They must also learn to apply their knowledge in a variety of increasingly complex situations. By appropriately using the knowledge, techniques and strategies listed in this document, they will develop the competencies outlined in the Science and Technology programs. By applying these competencies, they will acquire new knowledge which, in turn, will help them further develop their competencies.

In order to seek answers or solutions to scientific and technological problems (Competency 1), students must become familiar with strategies and acquire conceptual and technical knowledge that will enable them to define a problem, explore it and then justify their methodological choices and results. Similarly, the appropriate scientific or technological concepts and principles can help them understand phenomena, explain the operation of objects or form an opinion and, consequently, make the most of their scientific and technological knowledge (Competency 2). Finally, in order to communicate in the languages used in science and technology (Competency 3), they must have knowledge that will enable them to interpret and convey messages using the languages and types of representation associated with science and technology.

In elementary school, students became familiar with science and technology and explored knowledge involving simple and usually observable phenomena in their immediate environment. In secondary school, they further develop their scientific and technological literacy and continue to do so throughout their lives. In Cycle One, students learn about natural phenomena and man-made objects that interest them. In Cycle Two, the compulsory concepts are organized around two themes: *The Human Organism* in Secondary III and *The Environment* in Secondary IV. In the optional Environmental Science and Technology program, the knowledge to be acquired is organized around three environmental issues, two of which are new. Successful completion of this program will make it easier to enroll in the optional Physics and Chemistry programs offered in Secondary V.

The tables in this document outline the knowledge related to each of the four areas of the programs: The Material World, The Living World, The Earth and Space, and The Technological World. Each table is preceded by a text explaining how this knowledge contributes to students' learning in science and technology. Each section begins with a short text describing the related knowledge that was acquired at the elementary level. Two other tables provide information about the appropriate techniques and strategies for students to use.

The concepts are further clarified by a list of statements indicating the degree of complexity of the subject matter targeted and explanations of the progression of learning from one year to the next. In some cases, specifications about the extent of the knowledge to be addressed appear in parentheses.

Elementary school teachers can choose themes from among those listed in the program. It is therefore possible that
some students may not have studied certain concepts, even though the concepts mentioned here should have been
addressed at the elementary level. The table of <u>strategies</u> includes a column devoted to learning acquired in elementary
school.

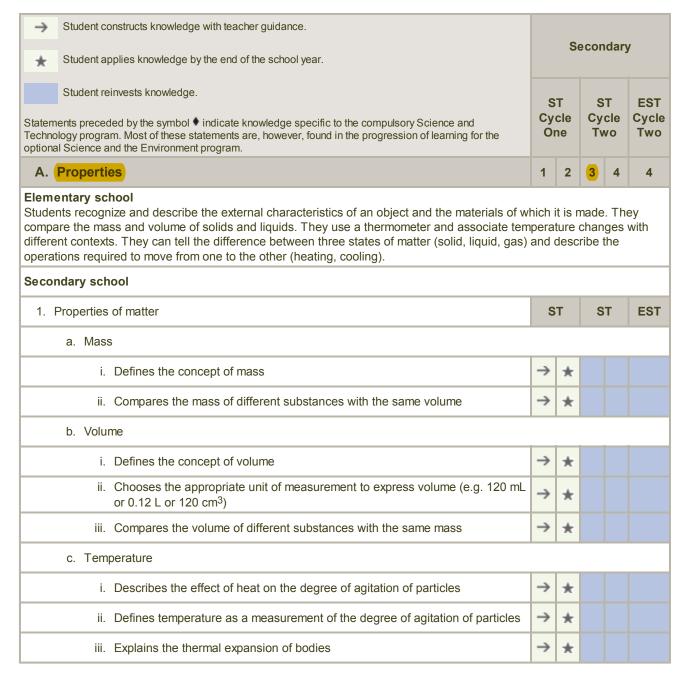
General Education Path

The Material World

In The Material World, students acquire scientific and technological knowledge pertaining to the organization of the world around us, the elements that compose it and the forces that govern it.

In secondary school, students explore increasingly complex phenomena and technical objects and seek answers and solutions to a variety of problems. They acquire scientific knowledge about The Material World that helps them understand and explain the factors at play in different scientific issues and in the operation of technological objects, systems and processes. This knowledge, along with the knowledge they acquire in other areas of the program, enables them to understand scientific models, theories and laws. Students refine their understanding of the concepts related to The Material World by using the experimental method, technological analysis and the technological design process.

In Secondary III, knowledge in this area is organized around the problem-solving process and the analysis and design of technical objects pertaining to the human organism, which enables students to make connections with knowledge from The Living World. In Secondary IV, students continue to construct their scientific and technological knowledge by exploring laws and models related to environmental issues.



d.	States of matter				
	 Names the different phase changes of matter (vaporization, condensation, freezing, melting, deposition, sublimation)¹ 	\rightarrow	*		
	ii. Interprets the phase change diagram for a pure substance	\rightarrow	*		
e.	Acidity/alkalinity				
	 Determines the observable properties of acidic, basic or neutral solutions (e.g. reaction to litmus, reactivity with metals) 	\rightarrow	*		
	ii. Determines the acidity or alkalinity of common substances (e.g. water, lemon juice, vinegar, soft drinks, milk of magnesia, cleaners)	\rightarrow	*		
f.	Characteristic properties				
	 Defines a characteristic property as a property that aids in the identification of a substance or group of substances 	\rightarrow	*		
	 ii. Identifies groups of substances based on their common characteristic properties (e.g. acids turn litmus red) 	\rightarrow	*		
	iii. Associates a characteristic property of a substance or material with its use (e.g. metal is used to make pots because it is a good conductor of heat)	\rightarrow	*		
2. Cha	acteristic physical properties	S	Т	ST	EST
a.	Melting point				
	i. Identifies a substance by its melting point using a reference document			*	
b.	Boiling point				
	i. Identifies a substance by its boiling point using a reference document			*	
C.	Density				
	i. Explains the concept of density			*	
	ii. Determines the density of different substances			*	
	iii. Identifies liquid and solid substances by their density using a reference document			*	
d.	Solubility				
+	i. Defines the concept of solubility			*	
*	ii. Describes the effect of variations in temperature on the solubility of a substance			*	
3. Prop	erties of solutions	S	Т	ST	ES1
a.	Solutions ²				
	 Describes the properties of an aqueous solution (e.g. only one visible phase, translucent) 	\rightarrow	*		
b.	Solute				
•	i. Recognizes the solute in a given aqueous solution			*	
C.	Solvent				
	i. Recognizes the solvent in a given aqueous solution (e.g. lymph, tears, cell			*	
*	plasma, urine)				
	plasma, urine) Concentration				

	solution's concentration	_				
	(iii. Determines the concentration of an aqueous solution (g/L or percentage)			*		
*	iv. Determines the concentration of an aqueous solution (g/L, percentage or ppm)		L		*	
	 Determines the concentration of an aqueous solution (g/L, percentage, pp mol/L) 	m,				*
e.	Electrolytes					
+	i. Defines the concept of electrolyte				*	
f.	Strength of electrolytes					
	 Qualitatively speaking, associates the strength of an electrolyte with its degree of dissociation 					k
g.	pH scale ³					
+	 Describes the pH scale (acidity, alkalinity, neutrality, increasing and decreasing values) 				*	
+	ii. Determines the pH of a few common substances (e.g. distilled water, rainwater, saliva, lemon juice, cleaners)				*	
h.	Electrolytic dissociation					
+	i. Describes electrolytic dissociation				*	
i.	lons					
•	i. Defines the concept of ion				*	
j.	Electrical conductivity					
+	 Describes the mechanism that allows aqueous solutions to conduct electri (electrolytic dissolution of a solute, formation of mobile ions) 	city			*	
4. Char	acteristic chemical properties		ST	S	Т	ES
a.	Reaction to indicators					
	 i. (Recognizes a substance by its characteristic chemical properties (e.g. star turns blue in the presence of an iodine solution, acidic solutions turn bromothymol blue yellow) 	<mark>ch</mark>		*		
B. Cha		1	2	3	4	4
D. One	1900					
tudents d ushing), matter is	ry school lemonstrate that the properties of matter do not change during physical changes but that they do change during chemical changes (e.g. cooking, combustion). The conserved during physical changes (e.g. mass of a piece of chalk whether whole h how certain household products are made (e.g. soap, paper, maple syrup).	ey recog	nize 1	hat t	he qu	ianti
econdar	y school					
1. Char	nges in matter		ST	S	Т	ES
a.	Conservation of matter					
	 Demonstrates that matter is conserved during a chemical change (e.g. conservation of mass in a precipitation reaction) 	\rightarrow	*			
	Mixtures					
b.						
b.	Describes the properties of a mixture (e.g. made up of several substances one or more phases)	· -	*			

C.	Solutions ⁴					
d.	Separation of mixtures					
	i. Associates a separation technique with the type of mixture to be separated ⁵	\rightarrow	*			
	 Describes the steps involved in separating a complex mixture (e.g. sedimentation, decantation and evaporation to separate salt water and sand) 	\rightarrow	*			
e.	Particle model					
	i. Defines the particle model as a means of representing the behaviour of matter			*		
	ii. Describes the particle model in terms of the qualities and limitations of a scientific model			*		
2. Phy	sical changes	S	Т	S	Т	EST
a.	Physical changes					
	 Describes the characteristics of a physical change (e.g. substance retains its properties, molecules remain intact) 	\rightarrow	*			
	ii. Recognizes different physical changes (e.g. phase changes, preparation or separation of a mixture)	\rightarrow	*			
b.	Dissolution					
+	i. Explains dissolution using the particle model			*		
C.	(Dilution)					
•	i. Explains dilution in terms of concentration and volume			*		
*	ii. Determines the final volume or concentration of an aqueous solution after dilution (e.g. when the volume of solvent is doubled, the concentration of the solution decreases by half)			*		
d.	Phase changes					
+	 Compares the arrangement of particles in a substance in a solid, liquid or gaseous form 			*		
+	ii. Explains phase changes using the particle model	L		*		
3. Che	mical changes	S	Т	S	Т	EST
a.	Chemical changes					
	 Describes the indicators of a chemical change (formation of a precipitate, effervescence, colour change, heat, light) 	\rightarrow	*			
	ii. Explains a chemical change based on the changes in the properties of the substances involved	\rightarrow	*			
	iii. Names different types of chemical changes (e.g. decomposition, oxidation)	\rightarrow	*			
	iv. Names chemical changes that occur in the human body (e.g. respiration, digestion)			*		
b.	Decomposition and synthesis					
+	i. Represents a decomposition or synthesis reaction using the particle model			*		
*	ii. Associates known chemical reactions with decomposition or synthesis reactions (e.g. respiration, photosynthesis, combustion, digestion)			*		
C.	Oxidation					

	II.	Associates known chemical reactions with oxidation reactions (e.g. combustion, corrosion)		*		
	iii.	Associates a chemical equation in which oxygen is one of the reactants with one of the possible cases of an oxidation reaction				7
d.	Prec	ipitation				
+	i.	Describes the visible manifestation of precipitation (formation of a solid deposit when two aqueous solutions are mixed)		*		
+	ii.	Represents a precipitation reaction using the particle model		*		
e.	Com	bustion				
	i.	Describes the perceivable manifestations of rapid combustion (e.g. heat, light)			*	
	ii.	Explains a combustion reaction using the fire triangle			*	
f.	Phot	osynthesis and respiration ⁶				
g.	Acid-	-base neutralization reaction				
*	i.	Gives examples of acid-base neutralization reactions (e.g. adding lime to neutralize the acidity of a lake)			*	
+	ii.	Names the products formed during acid-base neutralization (salt and water)			*	
+	iii.	Recognizes an acid-base neutralization from its equation			*	
h.	Salts	3				
	i.	Determines the molecular formula of the salt produced by the neutralization of a given acid and a given base				
i.	Туре	es of bonds				
	i.	Covalent				
		Defines a covalent bond as a bond resulting from a sharing of electrons				
		Makes a schematic representation of a covalent bond				
		 Identifies molecules that feature a covalent bond (e.g. N₂, CO₂) 				
	ii.	lonic				
		 Defines an ionic bond as a bond resulting from the gain or loss of electrons 				
		Makes a schematic representation of an ionic bond				
		 Identifies molecules that feature an ionic bond (e.g. NaCl, NH₄OH) 				
		Associates an ionic bond with an electrolytic substance				
j.	Law	of conservation of mass				
+	i.	Explains the law of conservation of mass during a chemical reaction			*	
+	ii.	Represents the conservation of mass using the particle model			*	
k.	Bala	ncing chemical equations				
	i.	Balances chemical equations			*	
+			 			
* L	Stoic	chiometry				

	i. Distinguishes an endothermic reaction from an exothermic reaction according					*
	to perceptible signs (e.g. temperature variations, emission of light)					
	ii. Distinguishes an endothermic reaction from an exothermic reaction according to the position of the energy term in the chemical equation					*
. Nucl	ear changes ⁷	S	Т	S	Т	ES
a.	Nuclear stability					
	 Explains nuclear stability as the case where the nucleus of the atom is held together by an optimal number of neutrons 					*
b.	Radioactivity					
	 Defines radioactivity as the emission of particles or energy by the nuclei of atoms following nuclear transformations 					*
	ii. Associates the use of radioactivity with technological applications (e.g. radiotherapy, dating)					*
C.	Fission and fusion					
	i. Distinguishes nuclear fission from nuclear fusion					*
. Tran	sformation of energy ⁸	S	Т	S	Т	ES
a.	Forms of energy					
	i. Describes different forms of energy (chemical, thermal, mechanical, radiation))		*		
	ii. Identifies the forms of energy involved in a transformation (e.g. electrical to thermal in a toaster, electrical to radiation in an infrared lamp)			*		
+	iii. Defines joule as the unit of measurement for energy			*		
b.	Law of conservation of energy					
	i. Explains qualitatively the law of conservation of energy				*	
	ii. Applies the law of conservation of energy in different contexts				*	
C.	Energy efficiency					
C.	Defines the energy efficiency of a device or system as the proportion of energy consumed that is transformed into effective work (amount of useful energy / amount of energy consumed x100)				*	
C.	Defines the energy efficiency of a device or system as the proportion of energy consumed that is transformed into effective work (amount of useful				*	
	 Defines the energy efficiency of a device or system as the proportion of energy consumed that is transformed into effective work (amount of useful energy / amount of energy consumed x100) 					
	i. Defines the energy efficiency of a device or system as the proportion of energy consumed that is transformed into effective work (amount of useful energy / amount of energy consumed x100) ii. Explains how to improve the energy efficiency of an electrical appliance					
	i. Defines the energy efficiency of a device or system as the proportion of energy consumed that is transformed into effective work (amount of useful energy / amount of energy consumed x100) ii. Explains how to improve the energy efficiency of an electrical appliance Distinction between heat and temperature ⁹				*	
d.	i. Defines the energy efficiency of a device or system as the proportion of energy consumed that is transformed into effective work (amount of useful energy / amount of energy consumed x100) ii. Explains how to improve the energy efficiency of an electrical appliance Distinction between heat and temperature9 i. Describes heat as a manifestation of energy	e varia	tion		*	
d.	i. Defines the energy efficiency of a device or system as the proportion of energy consumed that is transformed into effective work (amount of useful energy / amount of energy consumed x100) ii. Explains how to improve the energy efficiency of an electrical appliance Distinction between heat and temperature9 i. Describes heat as a manifestation of energy ii. Describes the relationship between heat and temperature		tion		*	*
d.	 i. Defines the energy efficiency of a device or system as the proportion of energy consumed that is transformed into effective work (amount of useful energy / amount of energy consumed x100) ii. Explains how to improve the energy efficiency of an electrical appliance Distinction between heat and temperature⁹ ii. Describes heat as a manifestation of energy iii. Describes the relationship between heat and temperature Relationship between thermal energy, specific heat capacity, mass and temperature i. Describes qualitatively the relationship between the change in thermal energy (quantity of heat) of a substance, its mass, its specific heat capacity and the 	/	ttion		*	*
d.	 i. Defines the energy efficiency of a device or system as the proportion of energy consumed that is transformed into effective work (amount of useful energy / amount of energy consumed x100) ii. Explains how to improve the energy efficiency of an electrical appliance Distinction between heat and temperature⁹ i. Describes heat as a manifestation of energy ii. Describes the relationship between heat and temperature Relationship between thermal energy, specific heat capacity, mass and temperature i. Describes qualitatively the relationship between the change in thermal energy (quantity of heat) of a substance, its mass, its specific heat capacity and the variations in temperature to which it is exposed ii. Applies the mathematical relationship between thermal energy, mass, specific 	/	tion		*	*

	 Determines graphically the magnitude of the effective force in a given situation 					*
g.	Relationship between work, force and distance travelled					
	Describes qualitatively the relationship between the work done, the force applied on a body and the distance travelled by the body					*
	 ii. Applies the mathematical relationship between work, effective force and distance travelled (W = F∆d) 					*
h.	Relationship between mass and weight					
	i. Describes qualitatively the relationship between mass and weight					*
	ii. Applies the mathematical relationship between mass and weight ($F_g = mg$)					*
i.	Relationship between potential energy, mass, acceleration and distance travelled					
	 Describes qualitatively the relationship between the potential energy of a body, its mass, its gravitational acceleration and the distance it travels 					*
	ii. Applies the mathematical relationship between potential energy, mass, gravitational acceleration and the distance travelled (E_p = mgh)					*
j.	Relationship between kinetic energy, mass and speed					
	 Describes qualitatively the relationship between the kinetic energy of a body, its mass and its speed 					*
	ii. Applies the mathematical relationship between kinetic energy, mass and speed (E $_{k}$ = $1\!\!/_{2}\text{mv}^{2})$					*
k.	Relationship between work and energy ¹⁰					
	 Describes qualitatively the relationship between the work done on a body and the variation in energy within that body 					*
	ii. Applies the mathematical relationship between work and energy (W = ΔE)					*
C. Org	anization	1	2	3	4	4
characteris	ry school classify objects or substances based on their properties, and living organisms based o stics. They use the common names for certain substances (e.g. water, carbon dioxide, nesis and respiration.					dying
Secondar	y school					
1. Struc	cture of matter	S	Т	S	ST	EST
a.	Atom					
	i. Describes Dalton's atomic model	\rightarrow	*			
	ii. Defines the atom as the basic unit of the molecule	\rightarrow	*			
b.	Molecule					
	Describes a molecule using Dalton's atomic model (combination of atoms linked by chemical bonds)	\rightarrow	*			
	ii. Represents the formation of a molecule using Dalton's atomic model	\rightarrow	*			
C.	Element					
	 Defines an element as a pure substance made of a single type of atom (e.g. Fe, N₂) 	\rightarrow	*			
d.	Periodic table					
	Describes the periodic table as a structured classification of elements	\rightarrow	*			

	i. Defines a pure substance as a substance made up of a single type of atom or molecule			*	
	ii. Distinguishes between elements (e.g. iron, dioxygen, sodium) and compounds (e.g. water, carbon dioxide, glucose)			*	
f.	Homogeneous and heterogeneous mixtures ¹¹				
	i. Describes homogeneous and heterogeneous mixtures in the human body (e.g. lymph, blood, urine)			*	
g.	Groups and periods				
+	i. Locates the groups and periods in the periodic table			*	
*	 Describes the common characteristics of a group (e.g. number of valence electrons, chemical reactivity) 			*	
+	iii. Associates the number of electron shells in an element with the number of its period			*	
h.	Rutherford-Bohr atomic model			_	
•	i. Describes the Rutherford-Bohr atomic model			*	
+	ii. Represents atoms using the Rutherford-Bohr model			*	
i.	Neutron				
	i. Describes the position and electrical charge of the neutron in an atom				
j.	Simplified atomic model				
	i. Represents an atom of a given element using the simplified atomic model				
k.	Lewis notation				
+	i. Determines the number of valence electrons in an element			*	
+	ii. Represents atoms using Lewis notation			*	
l.	Nomenclature and notation rules				
	 i. Applies nomenclature and notation rules to name the molecule or write the molecular formula of binary compounds 				
m.	Polyatomic ions				
	i. Recognizes the common polyatomic ions (e.g. NH ₄ ⁺ , OH ⁻ , NO ₃ ⁻ , CO ₃ ² -, SO ₄ ² -, PO ₄ ³ -) by their name, their formula or their composition				
n.	Concept of the mole				
	i. Defines the mole as the unit of measure of the amount of a substance				
	ii. Expresses an amount of a substance in moles		П		
0.	Avogadro's number				
	i. Expresses a quantity of particles using Avogadro's number				
Perio	dic classification	S	ЭТ	ST	E
a.	Atomic number				
	i. Associates the atomic number of an element with the number of protons it has	Π			

b. Isotopes ¹²					
 Defines isotopes as atoms of the same element whose nuclei have different numbers of neutrons and therefore different atomic masses 					*
ii. Defines a radioactive isotope as an isotope whose atomic nucleus is unstable					*
c. Relative atomic mass					
i. Explains qualitatively the concept of relative atomic mass					*
d. Periodicity of properties					
 Describes the periodicity of certain properties of elements (e.g. chemical reactivity, atomic radius, electronegativity) 					*
D. Fluids	1	2	3	4	4
Elementary school Students distinguish between three states of matter: solid, liquid and gas.					
Secondary school					
a. Pressure					
 Defines pressure as the force exerted by particles when they collide with a constricting surface 			*		
 Describes qualitatively the main factors that affect the pressure exerted by a fluid 			*		
b. Compressible and incompressible fluids					
i. Distinguishes between compressible and incompressible fluids			*		
ii. Names compressible fluids (e.g. air) and incompressible fluids (e.g. blood) in the human body			*		
iii. Explains how fluids move around in the human body, using the concept of pressure			*		
c. Relationship between pressure and volume					
 Qualitatively describes the relationship between the pressure and volume of a gas (e.g. inhaling and exhaling, bicycle pump) 			*		
E. Waves	1	2	3	4	4
Elementary school Students associate sunlight with a source of energy.					
Secondary school					
a. Frequency					
i. Defines the frequency of a wave as the number of cycles per second (Hz)			*		
 ii. Associates the frequency of a sound wave with the pitch of the sound (e.g. a low-frequency wave produces a low-pitched sound) 			*		
b. Wavelength					
 Defines wavelength as the distance between two identical points on a wave at a given time (e.g. distance between crests) 			*		
ii. Describes the relationship between wavelength and energy (e.g. X-rays, which are high-energy waves, have a short wavelength)			*		
c. Amplitude					
i. Defines the amplitude of a sound wave as the loudness of the sound			*		
d. Decibel scale					

i.	Locates on the decibel scale levels dangerous to the human ear based on duration or frequency of exposure			*		
e. Elec	tromagnetic spectrum					
i.	Locates different areas on the electromagnetic spectrum (e.g. radio waves, visible light, X-rays)			*		
ii.	Describes different applications of electromagnetic waves in the health care sector (e.g. X-rays, infrared optical imaging)			*		
f. Devi	ation of light waves					
i.	Describes how light rays are deviated by a plane reflective surface			*		
ii.	Determines the angle of reflection of a light ray on the surface of a plane mirror			*		
iii.	Describes how light rays are deviated when they pass through the surface of a translucent substance			*		
g. Foca	al point of a lens					
i.	Determines the focal point of concave and convex lenses			*		
ii.	Describes the relationship between the focal point of a lens and the degree of deviation of light rays in different situations (e.g. accommodation of the crystalline lens, choice of corrective lenses)			*		
F. Electrici	ty and electromagnetism	1	2	3	4	4
	hool the components of a simple electrical circuit (wire, power source, bulb, switch) a ecognize the effects of magnetism in magnets (attraction and repulsion).	nd d	escri	be th	eir	
Secondary sch	nool					
1. Electricity		S	Т	S	т	EST
a. Elec	trical charge					
i.	Associates elementary particles with their electrical charge				*	
ii.	Describes the behaviour of electrical charges of opposite signs or of the same sign when close together				*	
b. Stati	c electricity					
i.	Describes static electricity as the transfer of electrons from one body to another				*	
c. Ohm	's law					
i.	Describes qualitatively the relationship between voltage, resistance and current intensity in an electrical circuit				*	
ii.	Applies the mathematical relationship between voltage, resistance and current intensity in an electrical circuit (V = RI)				*	
d. Elec	trical circuits					
	tireal circuits					
i.	Describes the function of different elements of an electrical circuit (e.g. the wires transmit electrons along the circuit, resistors transform electrical energy into another form of energy) ¹³				*	
	Describes the function of different elements of an electrical circuit (e.g. the wires transmit electrons along the circuit, resistors transform electrical energy				*	
ii.	Describes the function of different elements of an electrical circuit (e.g. the wires transmit electrons along the circuit, resistors transform electrical energy into another form of energy) ¹³					
ii.	Describes the function of different elements of an electrical circuit (e.g. the wires transmit electrons along the circuit, resistors transform electrical energy into another form of energy) ¹³ Describes the two types of connections in electrical circuits (series, parallel)				*	
ii. iii. iv.	Describes the function of different elements of an electrical circuit (e.g. the wires transmit electrons along the circuit, resistors transform electrical energy into another form of energy) ¹³ Describes the two types of connections in electrical circuits (series, parallel) Distinguishes between alternating and direct current Represents a simple electrical circuit using a diagram				*	
ii. iii. iv. e. Rela	Describes the function of different elements of an electrical circuit (e.g. the wires transmit electrons along the circuit, resistors transform electrical energy into another form of energy) ¹³ Describes the two types of connections in electrical circuits (series, parallel) Distinguishes between alternating and direct current				*	

	 Describes qualitatively the relationship between the power of an electrical appliance, the electrical energy it consumes and the amount of time it is in operation 			*	
	iii. Applies the mathematical relationship between electrical energy consumed, the power of an electrical appliance and the amount of time it is in operation (E = $P\Delta t$)			*	
f.	Kirchhoff's laws				
	 Describes the distribution of current in various components of an electrical circuit 				*
	 Determines the value of the current flowing in various components of a series or parallel circuit 				*
	iii. Describes the distribution of the voltage across various components of an electrical circuit				*
	 iv. Determines the value of the voltage across various components of a series or parallel circuit 				*
	 Determines the value of the equivalent resistance of a series or parallel circui using Ohm's law and Kirchhoff's laws 	t			*
g.	Electrical field				
	 Describes qualitatively the effect of an electrical field on electrically charged particles 				k
h.	Coulomb's law				
	i. Applies the mathematical relationship between the electrical force, the magnitude of the electrical charges and the distance separating these charges (F = kq_1q_2/r^2)				*
2. Elect					
	tromagnetism	5	ST	ST	ES
	Magnetic field of a live wire	S	ST	ST	ES
	<u> </u>		БТ	ST *	ES
	Magnetic field of a live wire i. Describes the magnetic field produced by a current-carrying wire (right-hand		ST		ES
a.	Magnetic field of a live wire i. Describes the magnetic field produced by a current-carrying wire (right-hand rule or left-hand rule) ii. Names ways of modifying the intensity of the magnetic field produced by a		ST	*	ES
a.	Magnetic field of a live wire i. Describes the magnetic field produced by a current-carrying wire (right-hand rule or left-hand rule) ii. Names ways of modifying the intensity of the magnetic field produced by a current-carrying wire (type of wire, current intensity)		ST	*	ES
a.	i. Describes the magnetic field produced by a current-carrying wire (right-hand rule or left-hand rule) ii. Names ways of modifying the intensity of the magnetic field produced by a current-carrying wire (type of wire, current intensity) Forces of attraction and repulsion i. Compares the behaviour of a compass in the magnetic field of a magnet with		ST	*	ES
a.	i. Describes the magnetic field produced by a current-carrying wire (right-hand rule or left-hand rule) ii. Names ways of modifying the intensity of the magnetic field produced by a current-carrying wire (type of wire, current intensity) Forces of attraction and repulsion i. Compares the behaviour of a compass in the magnetic field of a magnet with the magnetic field created by a current-carrying wire		BT	*	
a.	Magnetic field of a live wire i. Describes the magnetic field produced by a current-carrying wire (right-hand rule or left-hand rule) ii. Names ways of modifying the intensity of the magnetic field produced by a current-carrying wire (type of wire, current intensity) Forces of attraction and repulsion i. Compares the behaviour of a compass in the magnetic field of a magnet with the magnetic field created by a current-carrying wire Magnetic field of a solenoid i. Describes the magnetic field produced by a solenoid (right-hand rule or		BT	*	ES

- 1. See The Material World, Changes, Physical changes, Phase changes (MW, B, 2, d).
- 2. In Cycle One, this concept appears under Changes.
- 3. See The Material World, Properties, Properties of matter, Acidity/alkalinity (MW, A, 1, e).
- 4. See The Material World, Properties, Properties of solutions (MW, A, 3).
- 5. See Techniques, Science, Separating mixtures (Techniques, Science, b).
- 6. See The Living World, Life-sustaining processes, Photosynthesis and respiration (LW, B, f).
- 7. See Organisation, Isotopes (MW, C, 2, b).
- 8. See The Technological World, Mechanical engineering, Technological systems, Energy transformations (<u>TW</u>, <u>B</u>, <u>2</u>, <u>c</u>).

- 9. The definition of temperature was covered in Cycle One. See *Properties* above (MW, A, 1 c).
- 10. See Law of conservation of energy (MW, B, 5, b).
- 11. See The Material World, Changes, Changes in matter, Mixtures (MW, B, 1, b).
- 12. See The Material World, Nuclear changes (MW, B, 4).
- 13. See The Technological World, Electrical engineering (TW, C).

General Education Path

The Living World

In The Living World, students acquire scientific and technological knowledge pertaining to life as it relates to molecules, cells, organisms and ecosystems.

In secondary school, students explore increasingly complex phenomena and technical objects and seek answers and solutions to a variety of problems. They acquire scientific knowledge about The Living World that helps them understand and explain the factors at play in different scientific issues. This knowledge, along with the knowledge they acquire in other areas of the program, particularly The Earth and Space, enables them to understand the complexity of the relationships between living organisms and their planet. Students refine their understanding of the concepts related to The Living World by using the experimental method, the observation method and modelling.

In Secondary III, the theme *The Human Organism* places The Living World at the heart of the program. In Secondary IV, students continue constructing their knowledge in this area by exploring environmental issues. They examine the influence of living organisms on their environment and discover how the knowledge they acquired can help them understand and improve the world around them.

* S	Student a	onstructs knowledge with teacher guidance. pplies knowledge by the end of the school year.		S	econ	econdary	
Statemen Technolog	nts prece	ded by the symbol • indicate knowledge specific to the compulsory Science and am. Most of these statements are, however, found in the progression of learning for the and the Environment program.	Су	T cle ne	S Cy Tv		EST Cycle Two
·		y of life forms	1	2	3	4	4
amphibia	s classif ans). Th onment	fy animals by kingdom and, in the case of vertebrates, by class (mammals, reptiled the physical and behavioural characteristics that demonstrate that (e.g. fins, claws, colour, ability to burrow into the ocean floor, migration).				adap	oted to
1. Ec			S	T	S	Т	EST
6	a. Hab	itat					
	i.	Names the characteristics that define a habitat (e.g. geographic location, climate, flora, fauna, proximity of man-made constructions)	\rightarrow	*			
	ii.	Describes the habitat of certain species	\rightarrow	*			
ı	b. Ecol	ogical niche					
	i.	Names the characteristics that define an ecological niche (e.g. habitat, diet, daily rhythms)	\rightarrow	*			
	ii.	Describes the ecological niche of an animal species	\rightarrow	*			
(c. Spe	cies					
	i.	Names the characteristics that define a species (common physical characteristics; natural, viable and fertile reproduction)	\rightarrow	*			
(d. Pop	ulation					
	i.	Distinguishes between a population and a species	\rightarrow	*			
	ii.	Calculates the number of individuals of a species in a given territory	\rightarrow	*			

♦ i.	Describes a given population (density, distribution, biological cycles)	*	
• ii.	Describes the influence of biotic or abiotic factors on the biological cycles of a population (natality, mortality, immigration, emigration)	*	
♦ iii.	Explains how the availability of resources in the environment affects reproduction and survival	*	
♦ iv.	Defines a community as a group of populations that interact	*	
V.	Defines an ecosystem as the relationships between the individuals in a community and abiotic factors in the environment	*	
f. Dyna	mics of communities		
i.	Biodiversity		
*	Defines the biodiversity of a community as the relative abundance of species it comprises	*	
+	Explains factors that affect the biodiversity of a given community	*	
ii.	Disturbances		
	Defines a disturbance in a community	*	
	Explains the effects of certain factors that disturb the ecological balance (e.g. human activity, natural disasters)	*	
g. Dyna	mics of ecosystems		
i.	Trophic relationships		
	Describes the trophic levels (producers, consumers, decomposers)	*	
	Explains the relationships between the trophic levels of a food web	*	
ii.	Primary productivity		
	Defines primary productivity as the quantity of organic matter produced by plants in a given territory	*	
	 Explains the effects of certain factors on primary productivity (e.g. bees help pollinate fruit trees, pathogenic microorganisms hinder plant growth) 	*	
iii.	Material and energy flow		
	Describes material and energy flow in an ecosystem	*	
iv.	Chemical recycling		
	 Describes certain processes underlying chemical recycling (e.g. action of microorganisms and decomposers, erosion) 	*	
h. Ecolo	ogical footprint		
i.	Explains the concept of ecological footprint		Γ
i. Ecoto	oxicology		
i.	Contaminant ¹		
	Defines a contaminant as an agent that causes changes in the physical, chemical or biological properties of an environment or an organism		
ii.	Bioaccumulation		_
	Defines bioaccumulation as the process by which a contaminant from the environment or food supply accumulates in an organism		
	Explains bioaccumulation in food chains (biomagnification)		T

iii. Bio	concentration					
	 Defines bioconcentration as a special case of bioaccumulation by which an organism accumulates a contaminant through direct contact with its environment (from sources other than food) 					*
iv. To	cicity threshold					
	 Defines the toxicity threshold of a substance as the minimum concentration of a substance that produces a significant harmful effect in an organism (mg/kg of the organism's mass) 					*
	 Describes factors that influence the toxicity of a contaminant (e.g. concentration, characteristics of the environment into which it is released, nature of the organisms with which it is in contact, duration of exposure) 					*
2. Diversity of life	forms	S	Т	S	Т	ES
a. Physical	and behavioural adaptation					
the	scribes physical adaptations that enable animals and plants to improve ir chances of survival (e.g. coat colour matched to the environment, shape eaves)	\rightarrow	*			
	scribes behavioural adaptations that enable animals and plants to improve ir chances of survival (e.g. movement in groups, phototropism)	\rightarrow	*			
b. Evolution						
i. Des	scribes the stages in the evolution of living organisms	\rightarrow	*			
ii. Exp	plains the natural selection process	\rightarrow	*			
c. Taxonom	у					
	ines taxonomy as a system for classifying living organisms based for the st part on their anatomical and genetic characteristics	\rightarrow	*			
ii. Ide	ntifies a species using a taxonomic key	\rightarrow	*			
d. Genes ar	nd chromosomes ²					
i. Loc	eates chromosomes in the cell	\rightarrow	*			
ii. Def	înes a gene as part of a chromosome	\rightarrow	*			
iii. Des	scribes the role of genes (transmission of hereditary characteristics)	\rightarrow	*			
3. Genetics		S	Т	S	т	ES
a. Heredity						
i. Def	înes heredity					*
b. Gene						
	ines a gene as being, in most cases, a DNA segment that carries the code synthesizing one or more proteins					*
	scribes the composition (nitrogen bases, sugar, phosphate) and the overall acture (bonding of bases on the double helix) of a DNA molecule					*
c. Characte	r trait					
i. Def	ines what an hereditary trait is					*
	mes hereditary traits in an individual or population					

d.	Allele					
	i. Defines an allele as a possible form of a gene					7
e.	Homozygotes and heterozygotes					
	Defines a homozygote as an individual with two identical alleles for a particular character trait					,
	 Defines a heterozygote as an individual with two different alleles for a particular character trait 					7
f.	Dominant and recessive					
	i. Describes the phenomena of dominant and recessive character traits					1
g.	Genotype and phenotype					
	i. Defines genotype					1
	ii. Defines phenotype					1
	 Describes an individual's genotype and phenotype for a character trait (e.g. a bean with a Yellow phenotype may have a Yellow-Yellow genotype or a Yellow-Green genotype) 					1
h.	Protein synthesis					
	i. Describes the role of DNA in protein synthesis					1
	ii. Explains the phenomena of transcription and translation of a strand of DNA					
i.	Crossbreeding					
	Explains the relationship between the crossbreeding carried out by humans on animals and plants and the desired traits obtained					
B. Life	e-sustaining processes	1	2	3	4	ı
Students e (transforma	ry school explain the essential needs of living organisms (e.g. food, respiration) and describe meation of energy, growth, maintenance of systems and body temperature). They describes the sis, which they distinguish from respiration.					
Secondar	y school					
a.	Characteristics of living things	_				_
	 Describes certain characteristics common to all living things (nutrition, relationships, adaptation, reproduction) 	\rightarrow	*			
b.	Plant and animal cells					
	i. Defines the cell as the structural unit of life	\rightarrow	*			
	ii. Names vital functions carried out by cells	\rightarrow	*			
	iii. Distinguishes between animal and plant cells	\rightarrow	*			
C.	Cellular components visible under a microscope					
	 i. Identifies the main cellular components visible under a microscope (cell membrane, cytoplasm, nucleus, vacuoles) 	\rightarrow	*			
	ii. Describes the role of the main cellular components visible under a microscope	\rightarrow	*			
d.	Inputs and outputs (energy, nutrients, waste)					
	i. Names cellular inputs	\rightarrow	*			
	ii. Names cellular outputs	\rightarrow	*			

e. Osmosis and diffusion					
i. Distinguishes between osmosis and diffusion	\rightarrow	*			
f. Photosynthesis and respiration ³					
i. Names the inputs and outputs involved in photosynthesis	\rightarrow	*			
ii. Represents the photosynthesis reaction in a balanced equation				*	
iii. Names the inputs and outputs involved in respiration	\rightarrow	*			
iv. Represents the respiration reaction in a balanced equation	П			*	
C. Tissues, organs and systems	1	2	3	4	4
Elementary school Students associate parts of animal anatomy and systems with their main function.					
Secondary school					
a. Tissues					
 i. Defines a tissue as a set of identical or different cells that work together to perform a common function in an organism 			*		
b. Organs		_		_	
i. Defines an organ as a differentiated part of an organism that performs one or more specific functions i. Defines an organ as a differentiated part of an organism that performs one or more specific functions.			*		
c. Systems					
 i. Defines a biological system as a set of cells, tissues or organs that perform one or more common functions 			*		
 ii. Describes the main functions performed by the human body (nutrition, relationships, reproduction) 			*		
D. Systems	1	2	3	4	
Elementary school Students describe the functions of certain parts of their anatomy (limbs, head). They associate pand systems with their main functions.	oarts	of ar	nimal	ana	tomy
Secondary school					
Nutrition					
1. (Digestive system)	S	Т	S	Т	EST
a. Digestive tract					
i. (Names the main parts of the digestive tract (mouth, esophagus, stomach, small intestine, large intestine, anus)			*		
ii. Explains the role of the digestive tract (decomposition of food, absorption of nutrients and water, elimination of waste)			*		
iii. Describes the functions of the main organs that make up the digestive tract (mouth, stomach, small intestine, large intestine)			*		
b. Digestive glands					
 i. Names the main digestive glands (salivary glands, gastric glands, pancreas, liver, intestinal glands) 			*		
ii. Describes the function of the main digestive glands (e.g. secretion of saliva, gastric enzymes, digestive juices, bile)			*		

c. (Types of foods)			
i. Describes the main biological functions of the different food constituents (water, proteins, carbohydrates, fats, vitamins, minerals)		*	
ii. Associates food constituents with their main sources (e.g. proteins with meat and meat substitutes)		*	
d. Energy value of different foods			
i. Describes the main biological functions of the different food constituents (water, proteins, carbohydrates, fats, vitamins, minerals) ii. Associates food constituents with their main sources (e.g. proteins with meat) and meat substitutes) d. (Energy value of different foods) i. (Evaluates the energy and nutritional value of different foods) e. (Transformation of food) i. (Describes the two types of transformation of food that take place in the digestive system (mechanical, chemical)) iii. Associates the organs in the digestive tract with the type of transformation they perform (e.g. mechanical action of teeth, chemical action of glands) 2. (Respiratory system) i. (Names the main parts of the respiratory system (nasal cavity, pharynx) trachea, bronchi, lungs) ii. (Explains the role of the respiratory system (gaseous exchanges between the blood and the surrounding air) iii. (Describes the function of the nasal cavity and lungs) b. (Functions of blood constituents) i. (Describes the main function of plasma (transportation of the blood's soluble) and formed elements) ii. (Names the formed elements of the blood (red blood cells, white blood cells) platelets) iii. (Describes the main function of the formed elements of the blood) c. (Compatibility of blood types) i. (Determines the compatibility or incompatibility of blood types (e.g., an individual with type A- blood can only receive type Q- or type A- blood) d. (Circulatory system) i. (Names the main parts of the circulatory system (heart, types of blood vessels) pulmonary and systemic circulation) ii. (Explains the role of the irrupatic system (transportation and exchange of gases, nutrients and waste) ii. (Describes the function of the main parts of the circulatory system (heart, types of blood vessels) pulmonary and systemic circulation) ii. (Explains the role of the irrupatic system (transportation and exchange of gases, nutrients and waste) iii. (Describes two ways of acquiring active immunity (production of antibodies) iii. (Describes			
e. (Transformation of food)			
		*	
		*	
2. Respiratory and circulatory systems	ST	ST	EST
a. (Respiratory system)			
		*	
		*	
iii. Describes the function of the nasal cavity and lungs		*	
b. Functions of blood constituents			
		*	
		*	
iii. Describes the main function of the formed elements of the blood		*	
c. Compatibility of blood types			
		*	
d. Circulatory system			
		*	
		*	
		*	
e. (Lymphatic system)			
i. Names the main parts of the lymphatic system (lymph, antibodies)		*	
		*	
		*	
3. (Excretory system)	ST	ST	EST
a. (Urinary system)			
		*	
ii. Explains the role of the excretory system (filtration of the blood, elimination of cellular waste)		*	
The same designs the 2004-2009 selections with the contest of the condensity			

iii. Describes the function of the kidneys and bladder		7	k	
b. Components of urine				
i. Names the main components of urine (water, mineral salts, urea)		2	k	
c. Maintaining a balanced metabolism				
i. Explains the role of the kidneys, lungs and sweat glands in maintaining a balanced metabolism		2	k	
Relationships				
1. Nervous and musculoskeletal systems	S	Т	ST	EST
a. Central nervous system				
i. (Identifies the parts of the central nervous system (brain, spinal cord)		9	k	
ii. Explains the role of the central nervous system (e.g. to manage complex behaviours and process sensory information and the related responses)		7	k	
iii. Describes the functions of the brain and the spinal cord		6	k	
b. (Peripheral nervous system)				
i. (Neuron)				
 Names the main parts of a neuron (synapse, axon, dendrites) 		7	k	
 Explains the role of the peripheral nervous system (transportation of nerve impulses from the senses to the brain and from the brain to the muscles) 		4	k	
ii. Neural inflow				
 Associates nerves with the transmission of nerve impulses 		7	k	
 Distinguishes between voluntary acts and reflex arcs 		-	k	
c. Sensory receptors				
i. <mark>Eye</mark>				
 Names the parts of the eye involved in vision (iris, cornea, crystalline lens, retina) 		7	k	
 Describes the function of the main parts of the eye 			k	
ii. <mark>Ear</mark>				
 Names the main parts of the ear involved in hearing and balance (auditory canal, ear drum, ossicles, cochlea, semicircular canals) 		2	k	
 Describes the function of the main parts of the ear involved in hearing 		9	k	
 Describes the role of the semicircular canals in maintaining balance 		1	k	
iii. <mark>Tongue</mark>				
 Describes the function of the taste buds on the tongue (transformation and transmission of flavours: sweet, salty, sour, bitter, umami) 		7	k	
iv. Nose				
 Names the parts of the nose involved in smelling (nasal cavity, olfactory bulb) 		9	k	
 Describes the function of the olfactory bulb 		9	k	
v. <mark>Skin</mark>				

and transmission of feelings of pressure, temperature and pain)			*		
and transmission of feelings of pressure, temperature and pain) d. (Musculoskeletal system) i. (Function of bones, joints and muscles) - (Names the main parts of the skeleton (head, thorax, spinal column, upper and lower limbs)) - (Describes the functions of the main parts of the skeleton (e.g. the spinal) column protects the spinal cord and allows the trunk to move) - (Explains the role of the musculoskeletal system) - (Describes how pairs of antagonistic muscles work (e.g. biceps, triceps)) - (Describes how joints work (linking bone to bone, mobility)) ii. (Types of muscles) - (Associates the different types of muscles (smooth, skeletal, heart) with the tissues in which they are found) iii. (Types of joint movement) - (Describes types of joint movement (e.g. flexion, rotation)) E. Survival of species 1 2 3 4 4 **Immentary school** deents describe the growth and reproduction of flowering plants and different animals. condary school					
i. Function of bones, joints and muscles					
			*		
			*		
Explains the role of the musculoskeletal system			*		
Describes how pairs of antagonistic muscles work (e.g. biceps, triceps)			*		
 Describes how joints work (linking bone to bone, mobility) 			*		
ii. <mark>Types of muscles</mark>					
			*		
iii. (Types of joint movement)					
 Describes types of joint movement (e.g. flexion, rotation) 			*		
E. Survival of species	1	2	3	4	4
Elementary school Students describe the growth and reproduction of flowering plants and different animals.					
Secondary school					
1. Reproduction	S	Т	S	Т	EST
a. Asexual and sexual reproduction					
	\rightarrow	*			
b. Reproductive mechanisms in plants					
i. Describes asexual reproductive mechanisms in plants (e.g. cutting, layering)	_				
comment of comment in plants (cig. comment)	7	*			
ii. Describes the sexual reproductive mechanism in plants (flowering plants)					
ii. Describes the sexual reproductive mechanism in plants (flowering plants) c. Reproductive mechanisms in animals i. Describes the roles of the male and female in the reproduction of certain	→	*			
ii. Describes the sexual reproductive mechanism in plants (flowering plants) c. Reproductive mechanisms in animals i. Describes the roles of the male and female in the reproduction of certain types of animals (e.g. birds, fish, mammals)	→	*			
ii. Describes the sexual reproductive mechanism in plants (flowering plants) c. Reproductive mechanisms in animals i. Describes the roles of the male and female in the reproduction of certain types of animals (e.g. birds, fish, mammals) d. Reproductive organs i. Names the main male and female reproductive organs (penis, testicles,	→	*			
ii. Describes the sexual reproductive mechanism in plants (flowering plants) c. Reproductive mechanisms in animals i. Describes the roles of the male and female in the reproduction of certain types of animals (e.g. birds, fish, mammals) d. Reproductive organs i. Names the main male and female reproductive organs (penis, testicles, vagina, ovaries, Fallopian tubes, uterus)	→	*			
ii. Describes the sexual reproductive mechanism in plants (flowering plants) c. Reproductive mechanisms in animals i. Describes the roles of the male and female in the reproduction of certain types of animals (e.g. birds, fish, mammals) d. Reproductive organs i. Names the main male and female reproductive organs (penis, testicles, vagina, ovaries, Fallopian tubes, uterus) e. Gametes	→	*			
ii. Describes the sexual reproductive mechanism in plants (flowering plants) c. Reproductive mechanisms in animals i. Describes the roles of the male and female in the reproduction of certain types of animals (e.g. birds, fish, mammals) d. Reproductive organs i. Names the main male and female reproductive organs (penis, testicles, vagina, ovaries, Fallopian tubes, uterus) e. Gametes i. Names the male and female gametes	 → → 	* *			
ii. Describes the sexual reproductive mechanism in plants (flowering plants) c. Reproductive mechanisms in animals i. Describes the roles of the male and female in the reproduction of certain types of animals (e.g. birds, fish, mammals) d. Reproductive organs i. Names the main male and female reproductive organs (penis, testicles, vagina, ovaries, Fallopian tubes, uterus) e. Gametes i. Names the male and female gametes ii. Describes the role of gametes in reproduction	 → → 	* *			
ii. Describes the sexual reproductive mechanism in plants (flowering plants) c. Reproductive mechanisms in animals i. Describes the roles of the male and female in the reproduction of certain types of animals (e.g. birds, fish, mammals) d. Reproductive organs i. Names the main male and female reproductive organs (penis, testicles, vagina, ovaries, Fallopian tubes, uterus) e. Gametes i. Names the male and female gametes ii. Describes the role of gametes in reproduction f. Fertilization	 → → → 	* *			
ii. Describes the sexual reproductive mechanism in plants (flowering plants) c. Reproductive mechanisms in animals i. Describes the roles of the male and female in the reproduction of certain types of animals (e.g. birds, fish, mammals) d. Reproductive organs i. Names the main male and female reproductive organs (penis, testicles, vagina, ovaries, Fallopian tubes, uterus) e. Gametes i. Names the male and female gametes ii. Describes the role of gametes in reproduction f. Fertilization i. Describes fertilization in humans	 → → → 	* *			

h.	Stages of human development				
	 Describes the stages of human development (childhood, adolescence, adulthood) 	\rightarrow	*		
i.	Contraception				
	Describes contraceptive methods (e.g. condom, ovulation suppression agents)	\rightarrow	*		
	 Describes the advantages and disadvantages of different contraceptive methods 	\rightarrow	*		
j.	Methods of preventing the implantation of the zygote in the uterus				
	Names methods of preventing the implantation of the zygote in the uterus (intrauterine device, day-after pill)	\rightarrow	*		
k.	Sexually transmitted and blood-borne diseases ⁴				
	i. Names sexually transmitted and blood-borne diseases	\rightarrow	*		
	ii. Describes behaviours to prevent contracting sexually transmitted and blood-borne diseases (e.g. wearing a condom)	\rightarrow	*		
	iii. Describes responsible behaviours to adopt after being diagnosed with a sexually transmitted or blood-borne disease (e.g. informing one's partner)	\rightarrow	*		
2. Cell o	livision ⁵	S	Т	ST	EST
a.	DNA				
+	i. Describes the shape of DNA (double helix)			*	
+	ii. Explains the role of DNA (a molecule bearing the organism's genetic code)			*	
b.	Mitosis				
	i. Describes the functions of mitosis (reproduction, growth, regeneration)			*	
C.	Meiosis and sexual development (meiosis, fertilization)				
	i. Describes the function of meiosis (production of gametes)			*	
*	ii. Indicates the advantages of sexual development (e.g. blending of genes from both parents, difference between descendants and their parents)			*	
d.	Functions of cell division				
+	i. Distinguishes between mitosis and meiosis based on their functions			*	
e.	Genetic diversity				
	i. Associates genetic diversity with sexual reproduction			*	
	Reproduction				
1. Repr	oductive system	S	Т	ST	EST
a.	Puberty (male and female)				
	 Describes physical and psychological changes that occur at puberty (e.g. appearance of body hair, voice change, ability to procreate, need for independence) 			*	
b.	Hormone regulation in men				
	i. Spermatogenesis				
	 Names the hormones responsible for the formation of spermatozoa (follicle stimulating hormone [FSH], luteinizing hormone [LH], testosterone) 			*	

ii. (Erection)	
Describes the physiology of erection	*
iii. (Ejaculation)	
Explains the function of ejaculation in reproduction	*
c. (Hormone regulation in women)	
i. <mark>Oogenesis</mark>	
 Names the hormones responsible for the maturation of the ovarian follicle (FSH, LH, estrogen, progesterone) 	*
ii. Ovarian cycle	
Describes the hormone changes that occur during the menstrual cycle	*
iii. Menstrual cycle	
 Describes the main stages in the menstrual cycle (e.g. menstruation, endometrium development, ovulation) 	*

- 1. See The Earth and Space, Characteristics of the Earth, Contamination (ES, A, 2, o; ES, A, 3, f; ES, A, 4, f).
- 2. See The Living World, Survival of species, Cell division (LW, E, 2).
- 3. The Secondary IV concepts related to photosynthesis and respiration are presented in the program under *The Material World, Changes*, Chemichal changes.
- 4. This replaces the term "sexually transmitted diseases" used in the program.
- 5. For concepts related to *Genes and chromosomes* addressed in Cycle One, see *The Living World, Diversity of life forms* above (<u>LW</u>, A, 2, <u>d</u>).

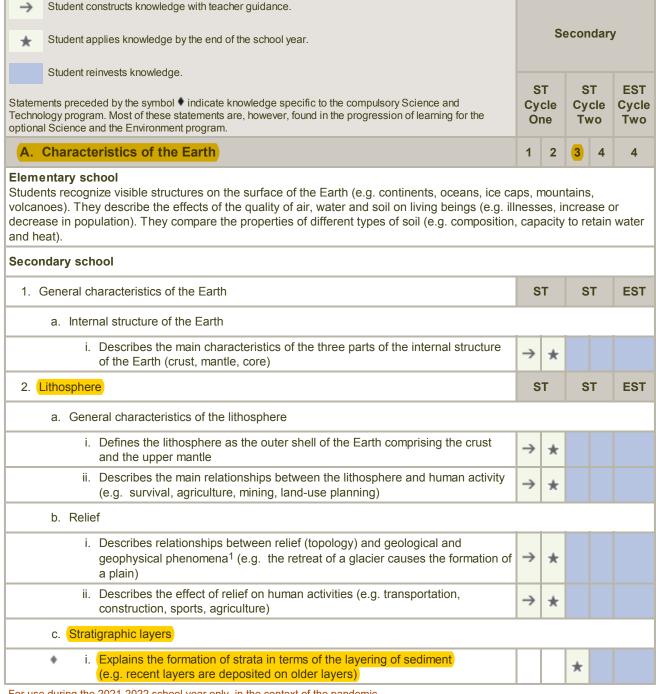
General Education Path

The Earth and Space

In The Earth and Space, students acquire scientific and technological knowledge pertaining to space and interactions in the biosphere.

In secondary school, students explore different phenomena that occur on Earth and in space and related technical objects. They seek answers and solutions to a variety of problems. They acquire knowledge about The Earth and Space that helps them explain the factors at play in different scientific issues. This knowledge, along with the knowledge they acquire in other areas of the program, enables them to understand scientific models, theories and laws. Students refine their understanding of the concepts related to The Earth and Space by using the experimental method, technological analysis and modelling.

In Secondary III, analyzing geological and geophysical phenomena and studying related technical objects enable students to locate living organisms on the geological time scale. In Secondary IV, students continue constructing their knowledge by exploring environmental issues. They do a more in-depth study of the impact of certain natural phenomena and human activities on the biosphere.



		Ogical time scale ²					
*		Places in order the main divisions of the geological time scale (Precambrian, Paleozoic, Mesozoic, Cenozoic)			*		
*	ii.	Describes events associated with the main divisions of the geological time scale (e.g. formation of oceans in the Precambrian Era, reign of the reptiles and dinosaurs in the Mesozoic Era)			*		
e.	Majo	or stages in the history of life on Earth					
+	i.	Locates the appearance or evolution of living organisms (e.g. bacteria, plants, fish, hominids) on the geological time scale			*		
f.	Extin	ctions					
*	i.	Locates periods of mass extinction of certain species on the geological time scale (e.g. disappearance of a large portion of marine life in the Paleozoic Era)			*		
g.	Foss	i <mark>ils</mark>)					
+	i.	Defines fossils as traces of organisms preserved for the most part in sedimentary rock			*		
*	ii.	Explains the usefulness of fossils in dating stratigraphic layers			*		
h.	Туре	es of rocks					
	i.	Describes the formation of three types of rock: igneous, metamorphic, sedimentary	\rightarrow	*			
	ii.	Classifies rocks by method of formation (e.g. granite is an igneous rock, lime is a sedimentary rock, slate is a metamorphic rock)	\rightarrow	*			
	iii.	Distinguishes between rocks and minerals	\rightarrow	*			
i.	Mine	rals					
	i.	Names basic minerals based on their properties (e.g. colour, hardness, magnetism)	\rightarrow	*			
	ii.	Distinguishes between minerals and ore				*	
	iii.	Describes some of the environmental impacts of mining or of the transformation of minerals				*	
j.	Туре	es of soil					
	i.	Classifies soils based on their composition (e.g. sand, clay, organic material)	\rightarrow	*			
k.	Soil	profile (horizons)					
+	i.	Describes the structure of a soil (superimposition of layers of different compositions and thicknesses)				*	
*	ii.	Explains the chemical and biological reactivity of a soil based on its composition (e.g. oxidation, acid-base neutralization, decomposition)				*	
I.	Perm	nafrost					
+	i.	Defines the permafrost as a layer of permanently frozen soil				*	
*	ii.	Explains some of the consequences of a rise in temperature in the permafrost (e.g. landslides, methane emissions)				*	
m.	Soil	depletion					
	i.	Explains how human activities contribute to soil depletion					
		ering capacity of the soil					

	ii. Explains the advantages of a good soil buffering capacity				7
0.	Contamination				
	i. Names soil contaminants ³				7
p.	Biogeochemical cycles				
	i. Carbon cycle				
*	 Describes transformations related to the circulation of carbon (e.g. photosynthesis, plant decomposition, dissolution in water, combustion of fossil fuels) 			*	
	ii. Nitrogen cycle				
*	 Describes transformations related to the circulation of nitrogen (e.g. nitrogen fixation, nitrification, denitrification) 			*	
	iii. Phosphorous cycle				
	 Describes transformations related to the circulation of phosphorous (e.g. erosion of rocks, breakdown of fertilizers, metabolism of algae) 				7
B. Hydr	osphere	S	т	ST	ES
a.	General characteristics of the hydrosphere				
	i. Describes the distribution of fresh water and salt water on the Earth's surface (e.g. glaciers contain inaccessible fresh water)	\rightarrow	*		
	 Describes the main interactions between the hydrosphere and the atmosphere (e.g. heat exchanges, climate regulation, meteorological phenomena) 	\rightarrow	*		
b.	Catchment area				
	i. Defines a catchment area as a territory surrounding a waterway			*	
	 Describes some of the impacts of human activity on the waterways in a catchment area 			*	
C.	Salinity				
+	i. Defines salinity as a measure of the quantity of salt in a solution			*	
+	ii. Describes the influence of salinity on the density of a solution			*	
d.	Oceanic circulation				
+	 Describes factors that affect the circulation of surface currents and deep currents (e.g. wind, the Earth's rotation, temperature, salinity, density) 			*	
*	 ii. Describes the role of thermohaline circulation on global climate regulation (e.g. effect of the Gulf Stream on the climate of the east coast of North America) 			*	
e.	Glacier and ice floe				
+	i. Distinguishes between glaciers and ice floes			*	
+	ii. Describes some of the impacts of the melting of glaciers and ice floes (e.g. increase in sea level, disturbance of thermohaline circulation)			*	
f.	Contamination				
	i. Names water contaminants ⁴				1
g.	Eutrophication				

		natural water					
4. A	Atmosphe	ere	S	Т	S	T	E
	a. Ger	neral characteristics of the atmosphere					
	i	. Locates the main layers of the atmosphere (troposphere, stratosphere, mesosphere, thermosphere)	\rightarrow	*			
	ii	. Describes the composition of pure air at sea level (nitrogen, oxygen, carbon dioxide, water vapour)	\rightarrow	*			
	iii	. Describes the relationships between the atmosphere and certain human activities (e.g. recreation, transportation, energy consumption)	\rightarrow	*			
	b. Gre	enhouse effect					
	+ i	. Describes the greenhouse effect				*	
	• ii	Explains some of the consequences of a higher concentration of greenhouse gases (e.g. global warming that could result in higher sea levels, disturbances in ecosystems or the melting of glaciers)				*	
	c. Air	mass					
	i	. Describes the properties of an air mass (temperature, humidity, pressure)				*	
	ii	. Explains the formation of clouds when two different air masses meet				*	
	d. Atm	ospheric circulation					
	+ i	. Describes the main factors responsible for atmospheric circulation (e.g. pressure variations, uneven heating of the Earth's surface)				*	
	ii	. Describes the effect of prevailing winds on the dispersal of air pollutants in a given region					7
	e. Cyc	clone and anticyclone					
	i	. Explains the formation of cyclones (low-pressure areas) and anticyclones (high-pressure areas)				*	
	f. Cor	ntamination					
	i	. Names air contaminants ⁵					1
5. C	Climate z	one	S	Т	S	Т	Ε
	a. Fac	ctors that influence the distribution of biomes					
	i	. Describes the geographical and climatic factors that affect the distribution of biomes (e.g. latitude, humidity, temperature, salinity)				*	
	b. Ter	restrial biomes					
	• i	. Describes different terrestrial biomes (e.g. fauna, flora, climate, type of soil)				*	
	c. Mai	rine biomes					
	+ i	. Describes different marine biomes (e.g. fauna, flora, temperature, salinity)				*	
		ical and geophysical phenomena	1	2	3	4	

Elementary school

Students explain the water cycle (evaporation, condensation, precipitation, runoff, infiltration) and describe different types of precipitation (rain, snow, hail, freezing rain). Concepts related to energy play an important role in the elementary school program. Students explain that sunlight, moving water and wind are renewable energy resources. They differentiate them from nonrenewable energy resources such as fossil fuels (e.g. gasoline, propane, butane, oil, natural gas). They describe technologies used to convert renewable energy into electricity (hydroelectric dams, wind turbines, solar panels).

Secondary scho	pol					
a. Tecto	nic plate					
	Describes the main elements of the theory of tectonic plates (e.g. plate, subduction zone, mid-oceanic ridge)	\rightarrow	*			
b. Oroge	enesis					
	Describes the formation of mountains, folding and breaks (tectonic plate movements)	\rightarrow	*			
c. Volcar	no					
i.	Describes a volcanic eruption	\rightarrow	*			
ii.	Describes the geographical distribution of volcanoes	\rightarrow	*			
d. Eartho	quake					
	Describes the processes that cause earthquakes (e.g. tectonic plate movements, slides)	\rightarrow	*			
e. Erosio	on					
	Describes different types of erosion (e.g. soils dried by the wind, fragmentation of rocks caused by water freezing and thawing)	\rightarrow	*			
f. Winds						
	Names the main factors responsible for wind (e.g. convection movements, movement of air masses)	\rightarrow	*			
g. Water	cycle					
i.	Explains the water cycle (phase changes, energy exchanges)	\rightarrow	*			
h. Natura	al energy sources					
	Describes the role of solar energy as a natural energy source (e.g. wind, tornadoes, hurricanes, storms)	\rightarrow	*			
i. Renev	wable and nonrenewable energy resources					
	Distinguishes between renewable and nonrenewable energy resources (e.g. Sun, molten rock, moving water, oil)	\rightarrow	*			
	Describes technologies used to produce electricity using the energy resources in the lithosphere, hydrosphere and atmosphere				*	
	Describes the main impact of the use of energy resources in the lithosphere, hydrosphere and atmosphere				*	
C. Astronom	nical phenomena	1	2	3	4	4
nd moons in our	at the cycle of day and night is related to the Earth's rotation. They distinguish solar system. They describe seasonal changes (e.g. temperature variations, leparticular the apparent position of the Sun and its influence on the length of share	umin	osity			nets
	elated to astronomy	S	T	S	Т	ES.
	rsal Gravitation					
	Defines gravitation as a force of mutual attraction between bodies	\rightarrow	*			
b. Earth-	-Moon system					
i. 1	Describes the tides in terms of the gravitational effect of the Earth-Moon				*	

<u> </u>	Light				
	i. Defines light as a form of radiant energy ⁶	\rightarrow	*		
	ii. Describes properties of light (propagation in a straight line, diffuse reflection by surfaces)	\rightarrow	*		
	iii. Explains different phenomena using the properties of light (cycles of day and night, seasons, phases of the Moon, eclipses)	\rightarrow	*		
d.	Solar energy flow				
	 Describes the main factors that affect the quantity of solar energy that reaches the Earth's surface (e.g. reflection and absorption of solar energy by the atmosphere or surfaces) 			*	
2. Sola	system	ST		ST	ES
a.	Characteristics of the solar system				
	 i. Compares some of the characteristics of the planets in our solar system (e.g. distances, relative size, composition) 	\rightarrow	*		
b.	Cycles of day and night				
	i. Explains the alternation of day and night in terms of the Earth's rotation	\rightarrow	*		
C.	Phases of the Moon				
	i. Explains the phases of the lunar cycle	\rightarrow	*		
d.	Eclipses				
	i. Explains a lunar or solar eclipse	\rightarrow	*		
e.	Seasons				
	 i. Explains the phenomenon of seasons in terms of the position of the Earth with respect to the Sun (tilt, revolution) 	\rightarrow	*		
f.	Comets				
	 Describes the main parts of a comet (core of ice and rock, tail of gas, and tail of dust) 	\rightarrow	*		
g.	Aurora borealis (northern lights)				
	 Locates the geographic regions where the aurora borealis occurs (polar regions) 	\rightarrow	*		
	ii. Identifies the atmospheric layer in which the aurora borealis occurs	\rightarrow	*		
h.	Meteoroid impact				
	 i. Identifies traces left by meteoroid impacts in Québec (e.g. craters, astroblemes) 	\rightarrow	*		
3. Spac	e	S	Т	ST	ES
a.	Scale of the universe				
	i. Astronomical unit				
+	 Defines an astronomical unit as the unit of length corresponding to the average distance between the Earth and the Sun 			*	
	ii. Light year				
	Defines light year as a unit of length corresponding to the distance			*	

	iii. Location of the Earth in the universe		
+	 Compares the relative distance between different celestial bodies (e.g. stars, nebulae, galaxies) 	*	
b.	Conditions conducive to the development of life		
*	 Describes conditions conducive to the development or maintenance of life (e.g. presence of a gaseous atmosphere, water, energy source) 	*	

- 1. See The Earth and Space, Geological and geophysical phenomena below (ES, B).
- 2. According to the scale established by the International Commission on Stratigraphy (2009).
- 3. See Living World, Ecotoxicology, Contaminant (LW, A, 1, i, i).
- 4. See Living World, Ecotoxicology, Contaminant (<u>LW</u>, A, 1, i, i).
- 5. See Living World, Ecotoxicology, Contaminant (<u>LW</u>, A, 1, i, i).
- 6. See The Material World, Changes, Transformation of energy, Forms of energy (MW, B, 5, a).

General Education Path

The Technological World

In The Technological World, students acquire and apply scientific and technological knowledge.

In secondary school, students analyze and design increasingly complex technical objects and seek solutions to increasingly sophisticated technological problems. The technical and technological knowledge they acquire helps them understand the objects and factors at play in different scientific issues as well as evaluate possible technological solutions. It also helps them apply knowledge they acquire in other areas of the program, in particular The Material World.

In Secondary III, analyzing and designing technical objects, processes and technological systems associated with the human organism allow students to integrate knowledge associated with The Living World. In Secondary IV, students continue constructing their knowledge by exploring environmental issues. They study the influence of technology with respect to these issues and discover how it can help them understand and improve the world around them.

	onstructs knowledge with teacher guidance. pplies knowledge by the end of the school year.	Secondary				
	einvests knowledge. ded by the symbol • indicate knowledge specific to the compulsory Science and	ST Cycle One		cle Cycle		EST Cycle Two
Technology progra	im. Most of these statements are, however, found in the progression of learning for the and the Environment program.					
A. Graphic	al language ¹	1	2	3	4	4
Elementary sc Students learn s Secondary sch	symbols associated with motion and parts and use them to produce or interpret of	liagra	ams (or dra	awin	gs.
	ram of principles (design plan)					
i.	Defines a diagram of principles as a representation used to effectively explain the operation of a technical object	\rightarrow	*			
ii.	Associates the functional elements of a technical object with the appropriate diagram of principles	\rightarrow	*			
iii.	Explains the operation of a simple technical object by drawing a diagram illustrating the active forces and the resulting motion	\rightarrow	*			
iv.	Names the subassemblies and parts essential to the operation of a technical object	\rightarrow	*			
V.	Indicates certain principles of simple machines illustrated in a technical object (e.g. a lever in a wheelbarrow, a wedge in an axe)	\rightarrow	*			
b. Cons	struction diagram (technical diagram)					
i.	Defines a construction diagram as a representation used to effectively explain the construction and assembly of a technical object	\rightarrow	*			
ii.	Associates the shape and arrangement of parts of technical objects with the appropriate construction diagram	\rightarrow	*			
iii.	Explains the construction of a simple technical object by drawing a diagram illustrating the assembly and arrangement of parts	\rightarrow	*			
iv.	Names the components of a simple technical object	\rightarrow	*			
V.	Indicates the links and guiding controls on a construction diagram	\rightarrow	*			
c. Stan	dards and representations					
i.	Chooses the appropriate type of diagram for a given representation (e.g. uses a construction diagram to represent assembly solutions, a diagram of principles to represent the operation of an object)			*		

	fferent types of motion related to the operation of an object ropriate symbols (rectilinear translation, rotation, helical)	*	
d. Geometric lines			
	drawing with a combination of geometric lines (e.g. the drawing) corner of a table is an arc joined to two sides of a right angle)	*	
e. Basic lines			
i. Names basic li extension, dim	ines in a drawing (visible contour, hidden contour, centre, lension lines)	*	
ii. Associates the simple part	e basic lines in a drawing with the contours and details of a	*	
f. Orthogonal projection	ons		
i. Associates the projections)	e types of projection with their use (multiview and isometric	*	
ii. (Interprets drav	vings representing parts in multiview orthogonal projection	*	
iii. Represents sir	mple shapes in multiview orthogonal projection	*	
iv. Represents sir	mple shapes in isometric projection	*	
v. Interprets asse number of part	embly drawings of technical objects consisting of a small ts	П	*
g. <mark>Scales</mark> ²			
i. Associates sca enlargement o	ales with their use (actual-size representation, reduction or of an object)	*	
ii. Chooses a sim	nple scale for a drawing (e.g. 1:1,1:2,5:1)	*	
iii. Takes the sca	le into account when interpreting drawings	*	
h. Forms of representa	ation		
i. Defines perspo	ective drawing, oblique projection and axonometric projection	*	
ii. (Sketches simp	ole objects freehand using different forms of representation	*	
i. Axonometric projecti	ion: exploded view (reading)		
i. Names the cha	aracteristics of an exploded view		*
	urpose of exploded views (projection accompanying the ructions or specifications for an object)		*
j. Cross-sectional view	ws)		
i. Describes the	purpose of cross-sectional views in technical drafting	*	
ii. Interprets a ted	chnical drawing with cross-sectional views	*	
iii. (Represents a	simple shape in a cross-sectional view	*	
k. (Dimensioning)			
	main dimensioning rules (e.g. to make a drawing easy to read, g dimensioning lines)	*	
ii. (Interprets tech manufacturing	nical drawings including the dimensions required for purposes	*	
I. Dimensional tolerand	ces		
	nce as the required manufacturing precision (dimensions ne drawing, along with allowances)		*

B. Mechanical engineering Elementary school Students describe the characteristics of motion (direction, speed). They describe the effect of a force on an object and on certain materials or structures. They become familiar with simple machines. They identify mechanical parts (e.g. gear assemblies, cams, springs), distinguish between translation and rotation and describe a simple sequence of mechanical parts in motion (e.g. in a door lock, the lever rotates and the motion of the bolt is rectilinear translation). Secondary school 1. Forces and motion ST ST **EST** a. Types of motion i. Identifies parts that move in a specific way in a technical object (rectilinear \rightarrow translation, rotation, helical) b. Effects of a force Explains the effects of a force in a technical object (change in the motion of \rightarrow an object, distortion of a material) c. Simple machines i. Identifies wheels, inclined planes and levers in simple technical objects * (e.g. a wheelbarrow is made up of a second-class lever and a wheel) ii. Describes qualitatively the mechanical advantages of different types of levers \rightarrow * (first-class, second-class, third-class) in different applications ST **EST** Technological systems ST a. System i. Identifies a system (set of connected elements that interact with each other) in \rightarrow * a technical object or technological application \rightarrow ii. Describes the overall function of a technological system * \rightarrow iii. Names the inputs and outputs of a technological system * \rightarrow * iv. Names the processes and control elements of a technological system b. Components of a system i. Describes the role of the components of a technological system (e.g. explains \rightarrow * the role of the parts of a lighting system) c. Energy transformations³ \rightarrow i. Associates energy with radiation, heat or motion * ii. Defines energy transformations \rightarrow * \rightarrow * iii. Identifies energy transformations in a technical object or technological system 3. Engineering ST **EST** a. Basic mechanical functions (links, guiding control) i. Describes the role of links and guiding controls in a technical object * ii. Identifies a guiding control in a technical object, as well as the related links \rightarrow * (e.g. a pizza wheel is guided by a pivot, which links it to the handle) b. Typical mechanical links i. Describes the advantages and disadvantages of different types of links ii. Names the types of links used in a technical object (e.g. the spiral link) * between a jar and its lid)

· · ·	Linking of mechanical parts					
	 Describes the characteristics of the links in a technical object (direct or indirect, rigid or flexible, removable or permanent, partial or complete) 				*	
	 Determines the desirable characteristics of links in the design of a technical object 				*	
	iii. Judges the choice of assembly solutions in a technical object				*	
d.	Degree of freedom of a part					
	 i. Explains the purpose of limiting motion (degree of freedom) in a technical object (e.g. some hinges limit how far a cupboard door can open, preventing from hitting the wall) 	it				
e.	Typical functions	_				
	i. Defines the typical functions (linking, guiding, sealing, lubricating)			*		
	ii. Associates a typical function with certain parts of a technical object			*		
	iii. Explains the choice of a type of link in a technical object (e.g. using a screw makes it possible to attach and remove a battery case)			\rightarrow	*	
f.	Guiding controls					
	 i. Explains the choice of a type of guiding control in a technical object (e.g. the slide guides a drawer and reduces friction) 				*	
g.	Adhesion and friction of parts					
	 Describes the advantages and disadvantages of the adhesion and friction of parts in a technical object 					
h.	Motion transmission systems					
	i. Identifies motion transmission systems in technical objects	\rightarrow	*			
i.	Function, components and use of motion transmission systems					
	 Names motion transmission systems in technical objects (friction gears, pulleys and belt, gear assembly, sprocket wheels and chain, wheel and worr gear) 	n		*		
	ii. Describes the functions of the components of a motion transmission system (e.g. in a bicycle, the gear assembly on the crankset is the driving unit, the sprocket wheel on the rear wheel is the receiving unit, and the chain is the intermediate unit)			*		
	iii. Describes the speed changes or reversibility of a motion transmission system (e.g. a sprocket wheel that is replaced by a smaller wheel or a wheel with fewer teeth increases rotation speed)	n)		*		
j.	Construction and characteristics of motion transmission systems					
	 i. Explains the choice of a motion transmission system in a technical object (e.g. using a gear assembly rather than friction gears to get better engine torque and avoid slipping) 				*	
k.	Motion transformation systems					
	i. Identifies motion transformation systems in technical objects	\rightarrow	*			
	Function, components and use of motion transformation systems					
l.	i. Names motion transformation systems in technical objects (e.g. screw gear)			*		
I.	system, cam and roller, connecting rod and crank, rack and pinion)			^		

i	ii. (Describes speed changes or the reversibility of a motion transformation system (e.g. the cam and roller is a nonreversible motion transformation system)			*		
m. Co	nstruction and characteristics of motion transformation systems					
	i. Explains the choice of a motion transformation system (screw gear system, cams, connecting rods, cranks, slides, rotating slider crank mechanisms, rack-and-pinion drive) in a technical object (e.g. most car jacks use a screw gear system rather than a rack-and-pinion system, because the force of the arm on the small crank provides more thrust and because, given that it is nonreversible, the system is safer)				*	
	 Explains the choice of a motion transformation system (screw gear, cams, connecting rods, cranks, slider-crank mechanism, rack-and-pinion drive, eccentric) in a technical object 					*
n. Sp	eed changes					
	i. Uses systems that allow for speed changes in the design of technical objects				*	
C. Electri	cal engineering	1	2	3	4	4
	cribe energy transformations and recognize them in different devices. They describe energy into electricity (e.g. wind turbines transform wind energy into electricity).	e wa	ays o	of tran	nsfor	ming
a. Po	wer supply					
	Defines power supply as the ability to generate electrical current				*	
	ii. Determines the source of current in technical objects with an electrical circuit (e.g. chemical battery, solar cell, alternator, thermocouple, piezoelectric)				*	
b. Co	nduction, insulation and protection					
	i. Defines conduction as the ability to conduct electricity				*	
	ii. Distinguishes between electrical conductors and insulators in a technical object				*	
i	ii. Describes the role of a protective device in a circuit (fuse, breaker)				*	
i	v. Analyzes the factors that affect electrical conductivity (section, length, nature, temperature of conductor)				*	
	v. Uses the colour code to determine the electrical resistance of a resistor					*
	i. Describes the operation of a printed circuit					*
c. Co	ntrol					
	i. Defines control as the ability to control the travel of electrical current				*	
	 Describes different types of switches (lever, pushbutton, flip-flop, magnetic control) 				*	
i	ii. Distinguishes between unipolar and bipolar switches					*
i	v. Distinguishes between unidirectional and bidirectional switches					*
d. Tr	ansformation of energy (electricity and light, heat, vibration, magnetism)					
	i. Associates the transformation of energy with different components of a circuit (e.g. bulbs transform electrical energy into light and heat)				*	
	ii. Describes the energy transformations that take place in electrical or electronic appliances (e.g. in a cell phone, electricity is transformed into light for the display and vibrations for the sound)				*	

i. Describes the function of certain electronic components (condenser, dio	de)					*	
D. (Materials)	1	Ì	2	3	4	4	
ementary school							
udents describe the physical properties of certain materials.							
econdary school							
Material resources		ST		S	Т	ES	
a. Raw materials		_					
 i. Associates raw materials with the unprocessed materials used in an indu (e.g. bauxite is the raw material used in aluminum smelters) 	stry	•	*				
b. Materials				_			
 Names the materials present in a technical object (e.g. a cooking pot is composed of two materials: a metal used to make the container and plas used to coat the handle) 	tic →	•	*				
ii. Determines the origins of the materials present in a technical object (animplant, mineral, wood)	mal, →		*				
c. Equipment							
 Defines tools and equipment as the elements needed to manufacture an object (machining, control, assembly) 	\rightarrow	•	*				
2. Mechanical properties of materials		ST ST		ST ST		Т	ES
a. (Constraints)							
 Describes the constraints to which different technical objects are subject tension, compression, torsion (e.g. the top of a beam is subject to compression) 	:)			*			
 Describes the constraints to which different technical objects are subject tension, compression, torsion, deflection, shearing (e.g. a diving board is subject to deflection) 					*		
b. (Mechanical properties)							
 i. (Describes the mechanical properties of different materials (hardness, due lasticity, malleability, corrosion resistance) 	ctility,	T		*			
c. Characteristics of mechanical properties							
 i. Explains the choice of a material based on its properties (e.g. the mallea of aluminum makes it useful for making thin-walled containers) 	bility				*		
d. Types and properties							
i. Associates the use of different types of materials with their respective pro-	perties						
 Ferrous alloys (cast iron is harder than steel) 		T		*			
 Nonferrous metals and alloys (the wire used in a dental appliance be made of a nickel and titanium alloy, which has shape memory) 	can			*			
 Wood and modified wood (e.g. oak is used for flooring because it in hard wood that is shock and wear resistant) 	s a			*			
 Plastics: thermoplastics and thermosetting plastics (e.g. thermopla are used for prostheses because of their corrosion resistance and lightness; Bakelite, a thermosetting plastic, is used to mould electr parts because it is a good electrical insulator) 					*		
 Ceramics (e.g. ceramics are used in ovens because they are very and heat and wear resistant) 	hard	T			*		
·		_					

 Composites (e.g. carbon fibre is used for hock hardness, resilience and lightness) 	ey sticks because of its				*	
e. Modification of properties						
 Describes different treatments to prevent degradatio plating, antirust treatments, painting) 	n of materials (e.g. metal				*	
f. Heat treatments						
 Defines heat treatments as ways of changing the pro (e.g. quenching increases hardness but fragility as w 					Ì	*
E. Manufacturing		1	2	3	4	4
Elementary school Students are introduced to the design and construction of instruments systems (e.g. water filtration), models (e.g. glider) and simple electrical different materials using the appropriate tools. They use a variety of a round-head fasteners, nuts) and tools to obtain an aesthetic finish.	al circuits. They trace parts a	ind	cut t	hem	out c	
Secondary school						
a. Specifications						
 Defines specifications as a set of constraints associated technical object 	ited with the design of a	>	*			
 ii. Evaluates a prototype or technical object based on t described in the specifications (human, technical, inc physical, environmental) 		→	*			
b. Manufacturing process sheet						
 Defines a manufacturing process sheet as a set of s the parts that make up a technical object 	eps to follow to machine	→	*			
ii. Follows a process and assembly sheet to construct few components or to construct part of that object	an object consisting of	→	*			
c. Shaping						
i. Machines and tools						
 Associates shaping processes with the types of the control of the co						*
 Determines the appropriate shaping technique observation of technical objects (e.g. some tab- lathe) 						*
d. Manufacturing						
i. Characteristics of laying out, drilling, tapping and thr	eading					
 Associates laying out (marking) with saving matechniques and the types of materials to be sh 					Ì	*
 Describes the characteristics of the tools need be machined (e.g. the tip of a metal drill is confident of the confident of the characteristics of the tools need be machined (e.g. the tip of a metal drill is double fluted) 						*
e. Measurement						
i. Direct measurement						
 Explains the purpose of direct measurement (under the machining of a part 	sing a ruler) to control					*
 Explains the choice of the direct measurement (a vernier caliper is more precise than a ruler) 	instrument used					*

F. Biotechnology	1	2	3	4	4
Elementary school Students do not address any concepts associated with biotechnology.					
Secondary school					
a. Processes					
i. Pasteurization					
 Describes the pasteurization process 			*		
 Describes the purpose of pasteurization (preservation of food and its nutritional properties) 			*		
ii. Manufacture of vaccines					
Describes the process for manufacturing vaccines			*		
iii. Assisted reproduction					
Describes different assisted-reproduction processes			*		
 Describes the purpose of artificial insemination (animal reproduction, an answer to human infertility, preservation of the gene pool) 			*		
iv. Cell cultures					
 Names parameters to be controlled in the case of cultured cells (sources of mother cells, growth, preservation, characteristics of cell media, ethical standards) 			*		
v. Genetic transformation (genetically modified organisms)					
 Names the main advantages and disadvantages of genetic transformation 			*		
vi. Cloning					
 Defines cloning as a reproductive process that results in an identical copy of an organism, a tissue or a cell, whether genetically modified or not 					*
Describes the main advantages and disadvantages of cloning					*
vii. Wastewater treatment					
Describes treatments used to decontaminate wastewater					*
viii. Biodegradation of pollutants					
 Describes ways to promote biodegradation of pollutants (e.g. phytoremediation) 					*

^{1.} See Techniques, Technology, Graphic communication (<u>Techniques - Technology, 1</u>).

^{2.} See Techniques, Technology, Graphic communication, Using scales (Techniques - Technology, 1, d).

^{3.} For Cycle Two concepts related to *Energy transformations*, see *The Material World*, *Changes, Transformation of energy* (MW, B, 5).

Science and Technology

General Education Path

Techniques

The techniques listed below are divided into three categories, depending on whether they apply to science or technology or both. Many of them require the use of instruments and tools or chemicals. Safety in the workshop and laboratory should be a constant concern.

 → Student constructs knowledge with teacher guidance. ★ Student applies knowledge by the end of the school year. 		,			
Student reinvests knowledge.					
Statements preceded by the symbol indicate knowledge specific to the compulsory Science and Technology program. Most of these statements are, however, found in the progression of learning for the optional Science and the Environment program.	Су	cle ne	Су	T cle vo	EST Cycle Two
A. Technology	1	2	3	4	4
Elementary school Students use some symbols associated with motion and electrical and mechanical parts. They i simple drawings containing symbols. By designing technical objects, they become familiar with t machines (e.g. lever, inclined plane, pulley, wheel). They trace parts and cut them out of differe introduced to the safe use of tools (e.g. pliers, screwdriver, hammer, wrench, template) and difference. Screws, glue, nails, round-head fasteners, nuts). They pay attention to finishing.	he u	se of ateria	simp ls. T	ole hey a	are
Secondary school					
1. Graphic communication ¹	S	T	S	Т	EST
a. Doing a technical drawing					
i. Chooses the best view for an elevation drawing of a technical object	\rightarrow	*			
ii. Represents the visible edges using solid lines	\rightarrow	*			
iii. Represents the hidden edges using dotted lines	\rightarrow	*			
iv. Indicates the overall external dimensions of an object on a drawing	\rightarrow	*			
b. Reading plans					
i. Associates views with the sides of a technical object	\rightarrow	*			
ii. Associates lines with the edges of a technical object	\rightarrow	*			
c. Drawing diagrams ²					
i. Chooses the best view to describe a technical object	\rightarrow	\rightarrow	\rightarrow	*	
ii. Uses different colours for each part of a technical object	\rightarrow	*			
 Indicates all the information needed to explain the operation or construction of an object 	\rightarrow	\rightarrow	\rightarrow	*	
d. Using scales ³					
i. Associates real measurements with each of the dimensions in a drawing	\rightarrow	*			
ii. Reduces or multiplies the dimensions of a technical object based on the scale	\rightarrow	*			
iii. Dimensions multiview orthogonal projections in accordance with the main dimensioning rules			*		

i.	Uses drawing instruments (e.g. ruler, square) to make diagrams	\rightarrow	*			
f. Con	structing a graph using instruments					
	Uses instruments to construct a graph (e.g. multiview orthogonal projection, isometric representation, perspective drawing)			\rightarrow	*	
. Manufactu		S	Т	S	Т	ES
a. Safe	ely using machines and tools ⁴					
i.	Uses tools safely (e.g. retractable utility knife, hammer, screwdriver, pliers)	\rightarrow	*			
ii.	Uses machine tools safely (band saw, drill, sander)			\rightarrow	*	
b. Mea	suring and laying out					
i.	Identifies the unit of measurement on the instrument	\rightarrow	*			
ii.	Positions the measuring instrument to obtain reliable reference points	\rightarrow	*			
iii.	Adopts the appropriate position for reading an instrument	\rightarrow	*			
iv.	Marks the materials to be shaped using a pencil or punch	\rightarrow	*			
c. Mac	hining and forming					
i.	Chooses the appropriate materials, tools, techniques and processes	\rightarrow	*			
ii.	Draws the necessary reference lines	\rightarrow	*			
iii.	Immobilizes the part to be formed	\rightarrow	*			
iv.	Forms the part in accordance with the steps in the following machining processes: sawing, drilling, sanding, filing	\rightarrow	*			
V.	Forms the part in accordance with the steps in the following machining processes: stripping, splicing, soldering			\rightarrow	*	
d. Finis	shing					
i.	Sands the sides or deburrs the edges of each part after forming	\rightarrow	*			
ii.	Uses the appropriate finish (stain, paint)	\rightarrow	*			
iii.	Grinds, polishes, hammers or chisels metal parts			\rightarrow	*	
e. Asse	embling					
i.	Marks the references (holes, points or guidelines)	\rightarrow	*			
ii.	Immobilizes parts during gluing	\rightarrow	*			
iii.	Drills to the diameter of the screws, nails or rivets used	\rightarrow	*			
iv.	Countersinks the openings for countersunk screws	\rightarrow	*			
f. Asse	embling and disassembling					
i.	Identifies and gathers the parts and hardware	\rightarrow	*			
ii.	Chooses the appropriate tools	\rightarrow	*			
iii.	For disassembly, numbers and records the location of the parts	\rightarrow	*			
iv.	In the case of electrical circuits, identifies and gathers the electrical components				*	
		-				

	. In the case of electronic circuits, identifies and gathers the electronic components					*
vi	. Chooses and places the electrical components in sequence based on the circuit diagram				*	
Vii	. Chooses and arranges the electronic components based on the circuit diagram					*
viii	. Connects the components using wire, connectors or solders				*	
g. Per	forming verification and control tasks					
i	. Evaluates the dimensions of a part during and after construction using a ruler			\rightarrow	*	
ii	. Compares the real dimensions of a part with the specifications (e.g. draft, drawing, technical sheet)			\rightarrow	*	
iii	. Uses a template to verify the conformity of a part			\rightarrow	*	
iv	. Evaluates the dimensions of a part during and after construction using vernier callipers				*	
h. Ma	king a part					
	. Makes a part using the appropriate techniques			\rightarrow	*	
B. Science		1	2	3	4	4
	me familiar with the use of observational instruments (magnifying glass, stereomic ng instruments (ruler, eyedropper, graduated cylinder, balance, thermometer, chromoter, chromoter).				cular	s) and
	ely using laboratory materials and equipment ⁵					
	Uses laboratory materials and equipment safely (e.g. allows hotplate to cool, uses beaker tongs)	\rightarrow	\rightarrow	\rightarrow	\rightarrow	*
i	. Handles chemicals safely (e.g. uses a spatula and pipette filler)	\rightarrow	\rightarrow	\rightarrow	\rightarrow	*
b. Sej	parating mixtures					
i	. Separates heterogeneous mixtures using sedimentation and decantation	\rightarrow	*			
	Separates heterogeneous mixtures using sedimentation and decantation Separates heterogeneous mixtures using filtration	\rightarrow	*			
ii	· · · · · · · · · · · · · · · · · · ·					
ii	. Separates heterogeneous mixtures using filtration	\rightarrow	*			
ii c. Des	. Separates heterogeneous mixtures using filtration . Separates different aqueous solutions using evaporation or distillation	\rightarrow	*			
ii c. Des	Separates heterogeneous mixtures using filtration Separates different aqueous solutions using evaporation or distillation signing and creating an environment Uses environmental design and construction techniques that respect the	→	*			
ii c. Des i d. Usi	Separates heterogeneous mixtures using filtration Separates different aqueous solutions using evaporation or distillation signing and creating an environment Uses environmental design and construction techniques that respect the characteristics of the habitat (e.g. terrarium, aquarium, composting medium)	→	*			
ii c. Des i d. Usi	Separates heterogeneous mixtures using filtration Separates different aqueous solutions using evaporation or distillation signing and creating an environment Uses environmental design and construction techniques that respect the characteristics of the habitat (e.g. terrarium, aquarium, composting medium) In measuring instruments	→	*			
ii c. Des i d. Usi ii	Separates heterogeneous mixtures using filtration Separates different aqueous solutions using evaporation or distillation signing and creating an environment Uses environmental design and construction techniques that respect the characteristics of the habitat (e.g. terrarium, aquarium, composting medium) In measuring instruments Adopts the appropriate position for reading an instrument	 → → → 	* *			
ii c. Des i d. Usi ii	Separates heterogeneous mixtures using filtration Separates different aqueous solutions using evaporation or distillation signing and creating an environment Uses environmental design and construction techniques that respect the characteristics of the habitat (e.g. terrarium, aquarium, composting medium) and measuring instruments Adopts the appropriate position for reading an instrument Measures the mass of a substance using a balance	 → → → 	* * *			
ii c. Des d. Usi ii iii	Separates heterogeneous mixtures using filtration Separates different aqueous solutions using evaporation or distillation signing and creating an environment Uses environmental design and construction techniques that respect the characteristics of the habitat (e.g. terrarium, aquarium, composting medium) and measuring instruments Adopts the appropriate position for reading an instrument Measures the mass of a substance using a balance Measures the volume of a liquid using the appropriate graduated cylinder	 → → → → → 	* * *			
ii c. Des i d. Usi ii iii	Separates heterogeneous mixtures using filtration Separates different aqueous solutions using evaporation or distillation signing and creating an environment Uses environmental design and construction techniques that respect the characteristics of the habitat (e.g. terrarium, aquarium, composting medium) and measuring instruments Adopts the appropriate position for reading an instrument Measures the mass of a substance using a balance Measures the volume of a liquid using the appropriate graduated cylinder Measures the volume of an insoluble solid using water displacement	 → → → → → 	* * * * *	→	*	

	i Lloca chearrational instruments appropriately (a.g. magnifying glass					
	 i. Uses observational instruments appropriately (e.g. magnifying glass, stereomicroscope, binoculars, microscope) 	\rightarrow	*			
f.	Preparing solutions					
	i. Prepares an aqueous solution of a specific concentration given a solid solute			*		
	 ii. Prepares an aqueous solution of a specific concentration given a concentrated aqueous solution 			*		
g.	Collecting samples					
	 Collects samples appropriately (e.g. sterilizes the container, uses a spatula, refrigerates the sample) 			\rightarrow	*	
. Tec	chniques common to Science and Technology	1	2	3	4	
a.	Verifying the repeatability, accuracy and sensitivity of measuring instruments					
	 Takes the same measurement several times to check the repeatability of the instrument used 					
	 ii. Carries out the required operations to ensure the accuracy of a measuring instrument (e.g. cleans and calibrates a balance, dries out a graduated cylinder, rinses and calibrates a pH-meter) 					
	-,···, ····					
	iii. Chooses a measuring instrument by taking into account the sensitivity of the instrument (e.g. uses a 25-mL graduated cylinder rather than a 100-mL one t measure 18 mL of water)	0				
b.	iii. Chooses a measuring instrument by taking into account the sensitivity of the instrument (e.g. uses a 25-mL graduated cylinder rather than a 100-mL one t	0				
b.	 iii. Chooses a measuring instrument by taking into account the sensitivity of the instrument (e.g. uses a 25-mL graduated cylinder rather than a 100-mL one t measure 18 mL of water) 					
b.	 iii. Chooses a measuring instrument by taking into account the sensitivity of the instrument (e.g. uses a 25-mL graduated cylinder rather than a 100-mL one to measure 18 mL of water) Interpreting the results of measurement i. Determines the error attributable to a measuring instrument (e.g. the error in measurement made using a graduated cylinder is provided by the 					

- 1. See The Technological World, Graphical language (TW, A).
- 2. The progression of learning associated with these techniques is characterized by the increasing complexity of the objects to be represented.
- 3. See The Technological World, Graphical language, Scales (TW, A, g).
- 4. When the teacher introduces a new technique, he or she should explain the related safety rules and repeat them often. After several practice sessions, students should apply the rules without being reminded.
- 5. When the teacher introduces a new technique, he or she should explain the related safety rules and repeat them often. After several practice sessions, students should apply the rules without being reminded.

Science and Technology

General Education Path

Strategies

The strategies listed below are fundamental to the approaches used in science and technology. They can be applied in a variety of increasingly complex contexts and are therefore inclusive. Thus, students build on the strategies they learned in elementary school. New strategies are added, including analytical strategies, which are adapted to students' level of cognitive development.

\rightarrow	Student constructs knowledge with teacher guidance.			Se	econ	dary	,
*	Student applies knowledge by the end of the school year.	tary					
	Student reinvests knowledge.	Elementary	S		S	-	EST Cycle
E:Th	be letter E indicates that students were introduced to this strategy in elementary school.		Óı		Tv		Two
A.	Exploration strategies		1	2	3	4	4
1.	Studying a problem or a phenomenon from different points of view (e.g. social, environmental, historical, economic)	Е					
2.	Distinguishing between the different types of information useful for solving the problem	E					
3.	Referring to similar problems that have already been solved	E					
4.	Becoming aware of his or her previous representations	E					
5.	Drawing a diagram for the problem or illustrating it	E					
6.	Formulating questions	E					
7.	Putting forward hypotheses (e.g. individually, in teams, as a class)	Е					
8.	Exploring various ways of solving the problem	Е					
9.	Anticipating the results of his or her approach	Е					
10.	Imagining solutions to a problem in light of his or her explanations	Е					
11.	Taking into account the constraints involved in solving a problem or making an object (e.g. specifications, available resources, time allotted)	Е					
12.	Examining his or her mistakes in order to identify their source	Е					
13.	Using different types of reasoning (e.g. induction, deduction, inference, comparison, classification)	Е					
14.	Using empirical approaches (e.g. trial and error, analysis, exploration using one's senses)	Е					
15.	Ensuring that the procedure is appropriate and safe and making the necessary adjustments		\rightarrow	*			
16.	Collecting as much scientific, technological and contextual information as possible to define a problem or predict patterns				\rightarrow	*	
17.	Generalizing on the basis of several structurally similar cases				\rightarrow	*	
18.	Developing various scenarios				\rightarrow	*	
19.	Considering various points of view on scientific or technological issues				\rightarrow	*	

B. Instrumentation strategies		1	2	3	4	4
 Using different sources of information (e.g. books, newspapers, Web sites, magazines, experts) 	Е					
2. Validating sources of information	Е					
 Using technical design to illustrate a solution (e.g. diagrams, sketches, technical drawings) 	Е					
 Using different tools for recording information (e.g. diagrams, notes, graphs, procedures, logbook) 	Е					
5. Using a variety of observational techniques and tools	Е					
6. Selecting suitable techniques or tools for observation		\rightarrow	*			
C. Analytical strategies		1	2	3	4	4
 Identifying the constraints and important elements related to the problem-solving situation 		\rightarrow	*			
2. Dividing a complex problem into simpler subproblems		\rightarrow	*			
 Using different types of reasoning (e.g. inductive and deductive reasoning, comparison, classification, prioritization) in order to process information 		\rightarrow	*			
Reasoning by analogy in order to process information and adapt scientific and technological knowledge				\rightarrow	*	
 Selecting relevant criteria to help him or her determine where he or she stands on a scientific or technological issue 				\rightarrow	*	
D. Communication strategies		1	2	3	4	4
 Using different means of communication to propose explanations or solutions (e.g. oral presentation, written presentation, procedure) 	Е					
2. Organizing information for a presentation (e.g. tables, diagrams, graphs)	Е					
Exchanging information	Е					
 Comparing different possible explanations for or solutions to a problem in order to assess their relevance (e.g. full-group discussion) 	Е					
Using tools to display information in various formats (e.g. data tables, graphs, diagrams)		\rightarrow	*			