

Québec Education Program Progression of Learning

Applied Science and Technology Secondary IV 2021-2022 School Year

Learning to Be Prioritized for the 2021-2022 School Year in the Context of the Pandemic

This document is identical to the one produced for the 2020-2021 school year.





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Progression of Learning in Secondary School

The progression of learning in secondary school constitutes a complement to each school subject, providing further information on the knowledge that the students must acquire and be able to use in each year of secondary school. This tool is intended to assist teachers in planning both their teaching and the learning that their students are to acquire.

The role of knowledge in learning

The knowledge that young people acquire enables them to better understand the world in which they live. From a very early age, within their families and through contact with the media and with friends, they accumulate and learn to use an increasingly greater body of knowledge. The role of the school should be to progressively broaden, deepen and structure this knowledge.

Knowledge and competencies must mutually reinforce each other. On the one hand, knowledge becomes consolidated when it is used and, on the other hand, the exercise of competencies entails the acquisition of new knowledge. Helping young people acquire knowledge raises the challenging question of how to make this knowledge useful and durable, and thus evokes the notion of competency. For example, we can never be really assured that a grammar rule has been assimilated until it is used appropriately in a variety of texts and contexts that go beyond the confines of a repetitive, targeted exercise.

Intervention by the teacher

The role of the teacher in knowledge acquisition and competency development is essential, and he or she must intervene throughout the learning process. In effect, the *Education Act* confers on the teacher the right to "select methods of instruction corresponding to the requirements and objectives fixed for each group or for each student entrusted to his care." It is therefore the teacher's responsibility to adapt his or her instruction and to base it on a variety of strategies, whether this involves lecture-based teaching for the entire class, individualized instruction for a student or a small group of students, a series of exercises to be done, a team activity or a particular project to be carried out.

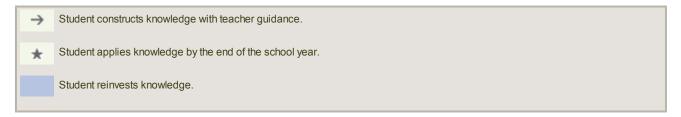
In order to meet the needs of students with learning difficulties, teachers should encourage their participation in the activities designed for the whole class, although support measures should also be provided, when necessary. These might involve more targeted teaching of certain key elements of knowledge, or they might take the form of other specialized interventions.

As for the evaluation of learning, it serves two essential functions. Firstly, it enables us to look at the students' learning in order to guide and support them effectively. Secondly, it enables us to verify the extent to which the students have acquired the expected learning. Whatever its function, in accordance with the *Policy on the Evaluation of Learning*, evaluation should focus on the acquisition of knowledge and the students' ability to use this knowledge effectively in contexts that draw upon their competencies.

Structure

The progression of learning is presented in the form of tables that organize the elements of knowledge similarly to the way they are organized in the subject-specific programs. In mathematics, for example, learning is presented in fields: arithmetic, geometry, etc. For subjects that continue on from elementary school, the *Progression of Learning in Secondary School* has been harmonized with the *Progression of Learning in Elementary School*. Every element of learning indicated is associated with one or more years of secondary school during which it is formally taught.

A uniform legend is used for all subjects. The legend employs three symbols: an arrow, a star and a shaded box. What is expected of the student is described as follows:



An **arrow** indicates that teaching must be planned in a way that enables students to begin acquiring knowledge during the school year and continue or conclude this process in the following year, with ongoing systematic intervention from the teacher.

A **star** indicates that the teacher must plan for the majority of students to have acquired this knowledge by the end of the school year.

A **shaded box** indicates that the teacher must plan to ensure that this knowledge will be applied during the school year.

Applied General Education Path

Introduction

This document provides additional information about the learning prescribed in the compulsory secondary-level Science and Technology programs and its progression from year to year and from cycle to cycle. This document is intended to help teachers with their lesson planning.

To progress in their learning, students need to do more than merely acquire knowledge. They must also learn to apply their knowledge in a variety of increasingly complex situations. By appropriately using the knowledge, techniques and strategies listed in this document, they will develop the competencies outlined in the Science and Technology programs. By applying these competencies, they will acquire new knowledge which, in turn, will help them further develop their competencies.

In order to seek answers or solutions to scientific and technological problems (Competency 1), students must become familiar with strategies and acquire conceptual and technical knowledge that will enable them to define a problem, explore it and then justify their methodological choices and results. Similarly, the appropriate scientific or technological concepts and principles can help them understand phenomena, explain the operation of objects or form an opinion and, consequently, make the most of their scientific and technological knowledge (Competency 2). Finally, in order to communicate in the languages used in science and technology (Competency 3), they must have knowledge that will enable them to interpret and convey messages using the languages and types of representation associated with science and technology.

In elementary school, students became familiar with science and technology and explored knowledge involving simple and usually observable phenomena in their immediate environment. In secondary school, they further develop their scientific and technological literacy and continue to do so throughout their lives. In Cycle One, students learn about natural phenomena and man-made objects that interest them. In Cycle Two, the concepts are organized around applications related to seven technological fields: medical, agricultural and agri-food, energy, information and communications, transportation, manufacturing and construction technologies. In the optional Science and the Environment program, the knowledge to be acquired is organized around two environmental issues about which the students will be required to form their own opinions, thus developing a new aspect of their subject-specific competencies. Successful completion of this program will make it easier to enroll in the optional Physics and Chemistry programs offered in Secondary V.

The tables in this document outline the knowledge related to each of the four areas of the programs: The Material World, The Living World, The Earth and Space, and The Technological World. Each table is preceded by a text explaining how this knowledge contributes to students' learning in science and technology. Each section begins with a short text describing the related knowledge that was acquired at the elementary level. Two other tables provide information about the appropriate techniques and strategies for students to use.

The concepts are further clarified by a list of statements indicating the degree of complexity of the subject matter targeted and explanations of the progression of learning from one year to the next. In some cases, specifications about the extent of the knowledge to be addressed appear in parentheses.

Elementary school teachers can choose themes from among those listed in the program. It is therefore possible that
some students may not have studied certain concepts, even though the concepts mentioned here should have been
addressed at the elementary level. The table of <u>strategies</u> includes a column devoted to learning acquired in elementary
school.

Applied General Education Path

The Material World

In The Material World, students acquire scientific and technological knowledge pertaining to the organization of the world around us, the elements that compose it and the forces that govern it.

In secondary school, students explore increasingly complex phenomena and technical objects and seek answers and solutions to a variety of problems. They acquire scientific knowledge about The Material World that helps them understand and explain the factors at play in different scientific issues and in the operation of technological objects, systems and processes. This knowledge, along with the knowledge they acquire in other areas of the program, enables them to understand scientific models, theories and laws. Students refine their understanding of the concepts related to The Material World by using the experimental method, technological analysis and the technological design process.

In Secondary III, students explore applications related to the seven technological fields, which enables them to make connections between human beings and The Material World, and provides them with an opportunity to integrate knowledge related to The Living World. In Secondary IV, they continue to construct and apply their knowledge about The Material World by analyzing and designing a variety of applications related to the same technological fields. Thus they acquire a better understanding of the omnipresence of science and technology in the world around us. In the optional Science and the Environment program, students consolidate their knowledge and form their own opinions regarding two environmental issues they will be asked to examine.

	•								
\rightarrow	Student constructs knowledge with teacher guidance.								
*	Student applies knowledge by the end of the school year.		Secondary						
	Student reinvests knowledge.			_			0.5		
Techno	ents preceded by the symbol indicate knowledge specific to the compulsory Applied Science ogy program. Most of these statements are, however, found in the progression of learning for th Environmental Science and Technology program.		Су	T cle ne	Су	ST cle vo	SE Cycle Two		
A.	Properties		1	2	3	4	4		
Stude compa differe	ntary school Its recognize and describe the external characteristics of an object and the materia The the mass and volume of solids and liquids. They use a thermometer and associate contexts. They can tell the difference between three states of matter (solid, liquid ons required to move from one to the other (heating, cooling).	ate tem	pera	ture (chan	ges			
Seco	dary school								
1.	roperties of matter		S	Т	AS	ST	SE		
	a. Mass								
	i. Defines the concept of mass		\rightarrow	*					
	ii. Compares the mass of different substances with the same volume		\rightarrow	*					
	b. Volume								
	i. Defines the concept of volume		\rightarrow	*					
	 ii. Chooses the appropriate unit of measurement to express volume (e.g. 1 or 0.12 L or 120 cm³) 	20 mL	\rightarrow	*					
	iii. Compares the volume of different substances with the same mass		\rightarrow	*					
	c. Temperature								
	i. Describes the effect of heat on the degree of agitation of particles		\rightarrow	*					
	ii. Defines temperature as a measurement of the degree of agitation of par	ticles	\rightarrow	*					
	iii. Explains the thermal expansion of bodies		\rightarrow	*					
	during the 2021 2022 school year only in the contact of the mandamic								

d. S	ates of matter					
	 Names the different phase changes of matter (vaporization, condensation, freezing, melting, deposition, sublimation) 	\rightarrow	*			
	ii. Interprets the phase change diagram for a pure substance	\rightarrow	*			
e. A	cidity/alkalinity					
	 Determines the observable properties of acidic, basic or neutral solutions (e.g. reaction to litmus, reactivity with metals) 	\rightarrow	*			
	ii. Determines the acidity or alkalinity of common substances (e.g. water, lemon juice, vinegar, soft drinks, milk of magnesia, cleaners)	\rightarrow	*			
f. C	haracteristic properties					
	i. Defines a characteristic property as a property that aids in the identification of a substance or group of substances	\rightarrow	*			
	ii. Identifies groups of substances based on their common characteristic properties (e.g. acids turn litmus red)	\rightarrow	*			
	iii. Associates a characteristic property of a substance or material with its use (e.g. metal is used to make pots because it is a good conductor of heat)	\rightarrow	*			
2. Charac	teristic physical properties	S	Т	AS	ST	SI
a. M	elting point					
	i. Identifies a substance by its melting point using a reference document			*		
b. B	piling point					
	i. Identifies a substance by its boiling point using a reference document			*		
c. D	ensity					
	i. Explains the concept of density			*		
	ii. Determines the density of different substances			*		
	iii. Identifies liquid and solid substances by their density using a reference document			*		
d. S	olubility					
	i. Defines the concept of solubility					*
	ii. Describes the effect of variations in temperature on the solubility of a substance					*
3. Propert	ies of solutions	S	Т	AS	ST	SI
a. S	olutions					
•	i. Recognizes the solute and the solvent in a homogeneous mixture			*		
	ii. Describes the effect of variations in the quantity of solute or solvent on a solution's concentration			*		
	iii. Determines the concentration of an aqueous solution (g/L or percentage)			*		
	 Determines the concentration of an aqueous solution (g/L, percentage, ppm, mol/L) 					*
b. E	ectrolytes					
	i. Defines the concept of electrolyte					*
	ii. Qualitatively speaking, associates the strength of an electrolyte with its degree of dissociation					*
	··· U · · · · · · · · · · · · · · · · ·					

oH scale					
 Describes the pH scale (acidity, alkalinity, neutrality, increasing and decreasing values) 					*
ii. Determines the pH of a few common substances (e.g. distilled water, rainwater, saliva, lemon juice, cleaners)					*
ons					
i. Defines the concept of ion					*
Electrical conductivity					
 Describes the mechanism that allows aqueous solutions to conduct electricity (electrolytic dissolution of a solute, formation of mobile ions) 					*
cteristic chemical properties	S	Т	AS	ST	SE
Reaction to indicators					
 Recognizes a substance by its characteristic chemical properties (e.g. starch turns blue in the presence of an iodine solution, acidic solutions turn bromothymol blue yellow) 			*		
iges	1	2	3	4	
school		-) T	0.5
les in matter	S	T	A	ST	
					SE
Conservation of matter					SE
i. Demonstrates that matter is conserved during a chemical change (e.g. conservation of mass in a precipitation reaction)	→	*			36
i. Demonstrates that matter is conserved during a chemical change	→	*			3E
Demonstrates that matter is conserved during a chemical change (e.g. conservation of mass in a precipitation reaction)	→	*			3E
i. Demonstrates that matter is conserved during a chemical change (e.g. conservation of mass in a precipitation reaction) Mixtures i. Describes the properties of a mixture (e.g. made up of several substances,		_			SE
 i. Demonstrates that matter is conserved during a chemical change (e.g. conservation of mass in a precipitation reaction) Mixtures i. Describes the properties of a mixture (e.g. made up of several substances, one or more phases) ii. Distinguishes between a solution or homogenous mixture (e.g. drinking water, 	→	*			SE
 i. Demonstrates that matter is conserved during a chemical change (e.g. conservation of mass in a precipitation reaction) Mixtures i. Describes the properties of a mixture (e.g. made up of several substances, one or more phases) ii. Distinguishes between a solution or homogenous mixture (e.g. drinking water, air, alloy) and a heterogeneous mixture (e.g. tomato juice, smog, rock) 	→	*			SE
 i. Demonstrates that matter is conserved during a chemical change (e.g. conservation of mass in a precipitation reaction) Mixtures i. Describes the properties of a mixture (e.g. made up of several substances, one or more phases) ii. Distinguishes between a solution or homogenous mixture (e.g. drinking water, air, alloy) and a heterogeneous mixture (e.g. tomato juice, smog, rock) Solutions i. Describes the properties of an aqueous solution (e.g. only one visible phase, 	→	*			SE
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 i. Demonstrates that matter is conserved during a chemical change (e.g. conservation of mass in a precipitation reaction) Mixtures i. Describes the properties of a mixture (e.g. made up of several substances, one or more phases) ii. Distinguishes between a solution or homogenous mixture (e.g. drinking water, air, alloy) and a heterogeneous mixture (e.g. tomato juice, smog, rock) Solutions i. Describes the properties of an aqueous solution (e.g. only one visible phase, translucent) Separation of mixtures i. Associates a separation technique with the type of mixture to be separated¹ ii. Describes the steps involved in separating a complex mixture (e.g. sedimentation, decantation and evaporation to separate salt water and sand) 	 → → 	* *	*		SE
F	decreasing values) ii. Determines the pH of a few common substances (e.g. distilled water, rainwater, saliva, lemon juice, cleaners) lons i. Defines the concept of ion Electrical conductivity i. Describes the mechanism that allows aqueous solutions to conduct electricity (electrolytic dissolution of a solute, formation of mobile ions) icteristic chemical properties Reaction to indicators i. Recognizes a substance by its characteristic chemical properties (e.g. starch turns blue in the presence of an iodine solution, acidic solutions turn bromothymol blue yellow) inges y school emonstrate that the properties of matter do not change during physical changes (e.g. but that they do change during chemical changes (e.g. cooking, combustion). They recognize the properties of the properties o	ii. Determines the pH of a few common substances (e.g. distilled water, rainwater, saliva, lemon juice, cleaners) lons i. Defines the concept of ion Electrical conductivity i. Describes the mechanism that allows aqueous solutions to conduct electricity (electrolytic dissolution of a solute, formation of mobile ions) cteristic chemical properties Sereaction to indicators i. Recognizes a substance by its characteristic chemical properties (e.g. starch turns blue in the presence of an iodine solution, acidic solutions turn bromothymol blue yellow) seresol y school emonstrate that the properties of matter do not change during physical changes (e.g. distort that they do change during chemical changes (e.g. cooking, combustion). They recognizes conserved during physical changes (e.g. mass of a piece of chalk whether whole or crust in how certain household products are made (e.g. soap, paper, maple syrup).	ii. Determines the pH of a few common substances (e.g. distilled water, rainwater, saliva, lemon juice, cleaners) lons i. Defines the concept of ion Electrical conductivity i. Describes the mechanism that allows aqueous solutions to conduct electricity (electrolytic dissolution of a solute, formation of mobile ions) cteristic chemical properties Fraction to indicators i. Recognizes a substance by its characteristic chemical properties (e.g. starch turns blue in the presence of an iodine solution, acidic solutions turn bromothymol blue yellow) Inges 1 2 y school emonstrate that the properties of matter do not change during physical changes (e.g. distortion out that they do change during chemical changes (e.g. cooking, combustion). They recognize to conserved during physical changes (e.g. mass of a piece of chalk whether whole or crushed), thow certain household products are made (e.g. soap, paper, maple syrup).	decreasing values) ii. Determines the pH of a few common substances (e.g. distilled water, rainwater, saliva, lemon juice, cleaners) lons i. Defines the concept of ion Electrical conductivity i. Describes the mechanism that allows aqueous solutions to conduct electricity (electrolytic dissolution of a solute, formation of mobile ions) cteristic chemical properties ST Assection to indicators i. Recognizes a substance by its characteristic chemical properties (e.g. starch turns blue in the presence of an iodine solution, acidic solutions turn bromothymol blue yellow) nges 1 2 3 y school emonstrate that the properties of matter do not change during physical changes (e.g. distortion, bre but that they do change during chemical changes (e.g. cooking, combustion). They recognize that the conserved during physical changes (e.g. mass of a piece of chalk whether whole or crushed). The phow certain household products are made (e.g. soap, paper, maple syrup).	decreasing values) ii. Determines the pH of a few common substances (e.g. distilled water, rainwater, saliva, lemon juice, cleaners) lons i. Defines the concept of ion Electrical conductivity i. Describes the mechanism that allows aqueous solutions to conduct electricity (electrolytic dissolution of a solute, formation of mobile ions) cteristic chemical properties ST AST Reaction to indicators i. Recognizes a substance by its characteristic chemical properties (e.g. starch turns blue in the presence of an iodine solution, acidic solutions turn bromothymol blue yellow) nges 1 2 3 4 y school emonstrate that the properties of matter do not change during physical changes (e.g. distortion, breaking ut that they do change during chemical changes (e.g. cooking, combustion). They recognize that the quenserved during physical changes (e.g. mass of a piece of chalk whether whole or crushed). They be a how certain household products are made (e.g. soap, paper, maple syrup).

. Pily:	sical cl	nanges	S	Т	A	ST	
a.	Phys	ical changes					
	i.	Describes the characteristics of a physical change (e.g. substance retains its properties, molecules remain intact)	\rightarrow	*			
	ii.	Recognizes different physical changes (e.g. phase changes, preparation or separation of a mixture)	\rightarrow	*			
•	iii.	Describes a few physical changes (e.g. dissolution, dilution, phase changes)			*		
+	iv.	Illustrates physical changes using the particle model			*		
b.	Disso	plution					Ī
	i.	Explains dissolution using the particle model					
C.	Diluti	on					
	i.	Explains dilution in terms of concentration and volume					
	ii.	Determines the final volume or concentration of an aqueous solution after dilution (e.g. the concentration of a solution decreases by half when the volume of solvent is doubled)					
. Che	mical o	changes	S	Т	A	ST	
a.	Cher	nical changes					Ī
	i.	Describes the indicators of a chemical change (formation of a precipitate, effervescence, colour change, heat, light)	\rightarrow	*			
	ii.	Explains a chemical change based on the changes in the properties of the substances involved	\rightarrow	*			
	iii.	Names different types of chemical changes (e.g. decomposition, oxidation)	\rightarrow	*			
	iv.	Names chemical changes that occur in the human body (e.g. respiration, digestion)			*		
b.	Deco	emposition and synthesis					
	i.	Associates known chemical reactions with decomposition or synthesis reactions (e.g. respiration, photosynthesis, combustion, digestion)					
C.	Oxida	ation at the second					
	i.	Represents an oxidation reaction using the particle model				*	
	ii.	Associates known chemical reactions with oxidation reactions (e.g. combustion, corrosion)				*	
	iii.	Associates a chemical equation in which oxygen is one of the reactants with one of the possible cases of an oxidation reaction					
d.	Preci	pitation					
	i.	Describes the visible manifestation of precipitation (formation of a solid deposit after two aqueous solutions are mixed)					
	ii.	Represents a precipitation reaction using the particle model					
e.	Com	bustion					
	i.	Describes the perceivable manifestations of rapid combustion (e.g. heat, light)				*	
	ii	Explains a combustion reaction using the fire triangle				*	

 ii. Names the products formed during acid-base neutralization (salt and water) iii. Recognizes an acid-base neutralization from its equation h. Salts i. Determines the molecular formula of the salt produced by the neutralization of a given acid and a given base i. Types of bonds i. Covalent – Defines a covalent bond as a bond resulting from a sharing of electrons – Makes a schematic representation of a covalent bond – Identifies molecules that feature a covalent bond (e.g. N₂, CO₂) ii. Ionic – Defines an ionic bond as a bond resulting from the gain or loss of electrons 					
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Defines an ionic bond as a bond resulting from the gain or loss of					
ologiono e e e e e e e e e e e e e e e e e e					
Makes a schematic representation of an ionic bond					
 Identifies molecules that present an ionic bond (e.g. NaCl, NH4OH) 					
Associates an ionic bond with an electrolytic substance					
j. Law of conservation of mass					
i. Explains the law of conservation of mass during a chemical reaction					
ii. Represents the conservation of mass using the particle model					
k. Balancing chemical equations					
i. Balances chemical equations					
I. Stoichiometry					
 Determines the quantities of reactants or products using stoichiometric calculations (gram or mole) 					
m. Endothermic and exothermic reactions					
 Distinguishes an endothermic reaction from an exothermic reaction according to perceptible signs (e.g. temperature variations, emission of light) 					
 Distinguishes an endothermic reaction from an exothermic reaction according to the position of the energy term in the chemical equation 					
Transformation of energy ³	3	ST	A	ST	
a. Forms of energy					
i. Describes different forms of energy (chemical, thermal, mechanical, radiation)			*		

b. I	aw of conservation of energy					
	i. Explain qualitatively the law of conservation of energy				*	
	ii. Applies the law of conservation of energy in different contexts				*	
c. I	Energy efficiency					
	i. Defines the energy efficiency of a device or system as the proportion of energy consumed that is transformed into effective work (amount of useful energy / amount of energy consumed x100)				*	
	ii. Explains how to improve the energy efficiency of an electrical appliance				*	
d. I	Distinction between heat and temperature ⁴					
	i. Describes heat as a manifestation of energy				*	
	ii. Describes the relationship between heat and temperature				*	
e. I	Relationship between thermal energy, specific heat capacity and temperature variation	n5				
	Describes qualitatively the relationship between the change in thermal energy (quantity of heat) of a substance, its mass, its specific heat capacity and the variations in temperature to which it is exposed					*
	ii. Applies the mathematical relationship between thermal energy, mass, specific heat capacity and temperature variation ($\Delta E = Q = mc\Delta T$)					*
f. I	Relationship between potential energy, mass, acceleration and distance travelled					
	Describes qualitatively the relationship between the potential energy of a body, its mass, its gravitational acceleration and the distance it travels					*
	ii. Applies the mathematical relationship between potential energy, mass, gravitational acceleration and the distance travelled (E _p = mgh)					*
g. I	Relationship between kinetic energy, mass and speed					
	Describes qualitatively the relationship between the kinetic energy of a body, its mass and its speed					*
	ii. Applies the mathematical relationship between kinetic energy, mass and speed (E _k = ½mv²)					*
h. I	Relationship between work and energy ⁶					
	Describes qualitatively the relationship between the work done on a body and the energy change within that body					*
	ii. Applies the mathematical relationship between work and energy (W = Δ E)					*
C. Orga	nization	1	2	3	4	4
characterist	r school ssify objects or substances based on their properties, and living organisms based on cs. They use the common names for certain substances (e.g. water, carbon dioxide, sis and respiration.					dying
Secondary						
a. <i>i</i>	Atom					
	i. Describes Dalton's atomic model	\rightarrow	*			
	ii. Defines the atom as the basic unit of the molecule	\rightarrow	*			
b. 1	Nolecule					
	Describes a molecule using Dalton's atomic model (combination of atoms linked by chemical bonds)	\rightarrow	*			
	ii. Represents the formation of a molecule using Dalton's atomic model	\rightarrow	*			

c. Element					
 Defines an element as a pure substance made of a single type of atom (e.g. Fe, N₂) 	\rightarrow	*			
d. Periodic table					
i. Describes the periodic table as a structured classification of elements	\rightarrow	*			
e. Pure substance					
 Defines a pure substance as a substance made up of a single type of atom or molecule 			*		
ii. Distinguishes between elements (e.g. iron, dioxygen, sodium) and compounds (e.g. water, carbon dioxide, glucose)			*		
f. Homogeneous and heterogeneous mixtures ⁷					
 Describes homogeneous and heterogeneous mixtures in the human body (e.g. lymph, blood, urine) 			*		
g. Elementary particles					
 Describes the position and electrical charge of the elementary particles in an atom (proton, electron, neutron) 					*
h. Simplified atomic model					
i. Represents an atom of a given element using the simplified atomic model					*
i. Lewis notation					
i. Determines the number of valence electrons in an element					*
ii. Represents atoms using Lewis notation					*
j. Nomenclature and notation rules					
 i. Applies nomenclature and notation rules to name the molecule or write the molecular formula of binary compounds 					*
k. Polyatomic ions					
i. Recognizes the common polyatomic ions (e.g. NH ₄ +, OH-, NO ₃ -, CO ₃ ² -, SO ₄ ² -, PO ₄ ³ -) by their name, their formula or their composition					*
I. Concept of the mole					
i. Defines the mole as the unit of measure of the amount of a substance					*
ii. Expresses an amount of a substance in moles					*
m. Relative atomic mass and isotopes					
 Defines isotopes as atoms of the same element whose nuclei have different numbers of neutrons and therefore different atomic masses 					*
ii. Explains qualitatively the concept of relative atomic mass					*
D. Fluids	1	2	3	4	4
Elementary school Students distinguish between three states of matter: solid, liquid and gas.					

a.	Pressure					
	Defines pressure as the force exerted by particles when they collide with a constricting surface			*		
	ii. Qualitatively describes the main factors that affect the pressure exerted by a fluid			*		
b.	Compressible and incompressible fluids					
	i. Distinguishes between compressible and incompressible fluids			*		
	ii. Names compressible fluids (e.g. air) and incompressible fluids (e.g. blood) in the human body			*		
	iii. Explains how fluids move around in the human body, using the concept of pressure			*		
C.	Relationship between pressure and volume					
	 i. Qualitatively describes the relationship between the pressure and volume of a gas (e.g. inhaling and exhaling, bicycle pump) 			*		
d.	Archimedes' principle					
+	i. Describes the relationship between the weight of the water displaced by an immersed body and the upward acting force				*	
+	ii. Explains the buoyancy of a body in terms of Archimedes' principle				*	
e.	Pascal's law					
+	i. Recognizes technical objects or technological systems whose operation is based on Pascal's principle (e.g. hydraulic systems, pneumatic systems)				*	
f.	Bernoulli's principle					
+	i. Describes the relationship between the velocity of a fluid and its pressure				*	
+	ii. Explains the concept of lift in terms of Bernoulli's principle				*	
. Wa	es	1	2	3	4	4
	y school ssociate sunlight with a source of energy.					
condar	v school					
a.	Fraguenay					
	Frequency					
	i. Defines the frequency of a wave as the number of cycles per second (Hz)			*		
	· ·			*		
b.	i. Defines the frequency of a wave as the number of cycles per second (Hz) ii. Associates the frequency of a sound wave with the pitch of the sound					
b.	i. Defines the frequency of a wave as the number of cycles per second (Hz) ii. Associates the frequency of a sound wave with the pitch of the sound (e.g. a low-frequency wave produces a low-pitched sound)					
b.	i. Defines the frequency of a wave as the number of cycles per second (Hz) ii. Associates the frequency of a sound wave with the pitch of the sound (e.g. a low-frequency wave produces a low-pitched sound) Wavelength i. Defines wavelength as the distance between two identical points on a wave			*		
	 i. Defines the frequency of a wave as the number of cycles per second (Hz) ii. Associates the frequency of a sound wave with the pitch of the sound (e.g. a low-frequency wave produces a low-pitched sound) Wavelength i. Defines wavelength as the distance between two identical points on a wave at a given time (e.g. distance between crests) ii. Describes the relationship between wavelength and energy (e.g. high-energy 			*		
	 i. Defines the frequency of a wave as the number of cycles per second (Hz) ii. Associates the frequency of a sound wave with the pitch of the sound (e.g. a low-frequency wave produces a low-pitched sound) Wavelength i. Defines wavelength as the distance between two identical points on a wave at a given time (e.g. distance between crests) ii. Describes the relationship between wavelength and energy (e.g. high-energy X-rays have a short wavelength) 			*		
C.	i. Defines the frequency of a wave as the number of cycles per second (Hz) ii. Associates the frequency of a sound wave with the pitch of the sound (e.g. a low-frequency wave produces a low-pitched sound) Wavelength i. Defines wavelength as the distance between two identical points on a wave at a given time (e.g. distance between crests) ii. Describes the relationship between wavelength and energy (e.g. high-energy X-rays have a short wavelength) Amplitude			*		

e. Elec	stromagnetic spectrum					
i.	Locates different areas on the electromagnetic spectrum (e.g. radio waves, visible light, X-rays)			*		
ii.	Describes different applications of electromagnetic waves in the health care sector (e.g. X-rays, infrared optical imaging)			*		
f. Dev	iation of light waves					
į.	Describes how light rays are deviated by a plane reflective surface			*		
ii.	Determines the angle of reflection of a light ray on the surface of a plane mirror			*		
iii.	Describes how light rays are deviated when they pass through the surface of a translucent substance			*		
g. Foc	al point of a lens					
į.	Determines the focal point of concave and convex lenses			*		
ii.	Describes the relationship between the focal point of a lens and the degree of deviation of light rays in different situations (e.g. accommodation of the crystalline lens, choice of corrective lenses)			*		
F. Electric	ity and electromagnetism	1	2	3	4	4
	chool the components of a simple electrical circuit (wire, power source, bulb, switch) arecognize the effects of magnetism in magnets (attraction and repulsion).	nd de	escri	be th	eir	
Secondary sc	hool					
1. Electricity		S	Т	A	ST	SE
	ctrical charge	S	Т	AS	ST	SE
a. Elec	<u>′</u>	S	Т	AS	ST 🖈	SE
a. Elec	ctrical charge	S	Т	AS		SE
a. Elec i.	Associates elementary particles with their electrical charge Describes the behaviour of electrical charges of opposite signs or of the same	S	Т	AS	*	SE
a. Elec i. ii. b. Stat	Associates elementary particles with their electrical charge Describes the behaviour of electrical charges of opposite signs or of the same sign when close together	S	Т	A	*	SE
a. Elec i. ii. b. Stat	Associates elementary particles with their electrical charge Describes the behaviour of electrical charges of opposite signs or of the same sign when close together ic electricity Describes static electricity as the transfer of electrons from one body to another	S	Т	A	*	SE
a. Electing in the state of the	Associates elementary particles with their electrical charge Describes the behaviour of electrical charges of opposite signs or of the same sign when close together ic electricity Describes static electricity as the transfer of electrons from one body to another	S	Т	A	*	SE
a. Election i. b. State i. c. Ohr	Associates elementary particles with their electrical charge Describes the behaviour of electrical charges of opposite signs or of the same sign when close together ic electricity Describes static electricity as the transfer of electrons from one body to another n's law Qualitatively describes the relationship between voltage, resistance and	S	Т	A	*	SE
a. Election in the state of the	Associates elementary particles with their electrical charge Describes the behaviour of electrical charges of opposite signs or of the same sign when close together ic electricity Describes static electricity as the transfer of electrons from one body to another n's law Qualitatively describes the relationship between voltage, resistance and current intensity in an electrical circuit Applies the mathematical relationship between voltage, resistance and current	S	T	A	* *	SE
a. Election in the state of the	Associates elementary particles with their electrical charge Describes the behaviour of electrical charges of opposite signs or of the same sign when close together ic electricity Describes static electricity as the transfer of electrons from one body to another n's law Qualitatively describes the relationship between voltage, resistance and current intensity in an electrical circuit Applies the mathematical relationship between voltage, resistance and current intensity in an electrical circuit (V = RI)	S	Т	A	* *	SE
a. Election in the state of the	Associates elementary particles with their electrical charge Describes the behaviour of electrical charges of opposite signs or of the same sign when close together ic electricity Describes static electricity as the transfer of electrons from one body to another in an electrical circuit intensity in an electrical circuit Applies the mathematical relationship between voltage, resistance and current intensity in an electrical circuit (V = RI) Extrical circuits Describes the function of different elements of an electrical circuit (e.g. the wires transmit electrons along the circuit, resistors transform electrical energy)	S	T	A	* * *	SE
a. Election in the state of the	Associates elementary particles with their electrical charge Describes the behaviour of electrical charges of opposite signs or of the same sign when close together ic electricity Describes static electricity as the transfer of electrons from one body to another n's law Qualitatively describes the relationship between voltage, resistance and current intensity in an electrical circuit Applies the mathematical relationship between voltage, resistance and current intensity in an electrical circuit (V = RI) Etrical circuits Describes the function of different elements of an electrical circuit (e.g. the wires transmit electrons along the circuit, resistors transform electrical energy into another form of energy) Between voltage, resistance and current intensity in an electrical circuit (v = RI)	S	T	A	* * *	SE

e.	Relationship between power and electrical energy					
	i. (Applies the mathematical relationship between power, voltage and current intensity in an electrical circuit (P = VI)				*	
	ii. Describes qualitatively the relationship between the power of an electrical appliance, the electrical energy it consumes and the amount of time it is in operation				*	
	iii. (Applies the mathematical relationship between electrical energy consumed, the power of an electrical appliance and the amount of time it is in operation $(E = P\Delta t)$				*	
2. Elec	tromagnetism	S	т	A	ST	SE
a.	Magnetic field of a live wire					
	i. (Describes the magnetic field produced by a current-carrying wire (right-hand rule or left-hand rule)				*	
	ii. (Names ways of modifying the intensity of the magnetic field produced by a current-carrying wire (type of wire, current intensity)				*	
b.	Forces of attraction and repulsion					
	i. (Compares the behaviour of a compass in the magnetic field of a magnet with the magnetic field created by a current-carrying wire)				*	
C.	Magnetic field of a solenoid					
+	i. (Describes the magnetic field produced by a solenoid (right-hand rule or left-hand rule)				*	
*	ii. (Names ways of changing the intensity of the magnetic field produced by a solenoid (nature of the core, intensity of the current, number of turns)				*	
d.	Electromagnetic induction					
+	 i. (Names ways of inducing electrical current in a wire (e.g. movement of a magnet, changing the intensity of a magnetic field) 				*	
G. For	ce and motion ⁹	1	2	3	4	4
econdai	re					
a.	Force					
+	 i. (Describes the effects produced by a force (change in the state of motion of a body, distortion of a body) 				*	
b.	Types of forces					
+	i. (Recognizes different types of forces in technical objects or technological systems (e.g. gravitational force in a chute, magnetic force exerted by an electromagnet)				*	
C.	Equilibrium of two forces					
+	 i. (Describes the conditions under which a body subjected to two forces can be in equilibrium) 				*	
d.	Relationship between constant speed, distance and time					
+	i. Qualitatively describes the relationship between speed, distance and time				*	
+	ii. (Applies the mathematical relationship between constant speed, distance and time $(v = d/\Delta t)$)				*	
e.	Effective force					
	Defines effective force as the component of the applied force parallel to the direction of travel					*
	ii. Determines graphically the magnitude of the effective force in a given					

	f. Re	elationship between work, force and distance travelled
		i. Describes qualitatively the relationship between the work done, the force applied on a body and the distance travelled by the body
	i	ii. Applies the mathematical relationship between work, effective force and distance travelled (W = FΔd)
9	g. <mark>Dis</mark>	stinction between mass and weight
4	•	i. Qualitatively describes the relationship between mass and weight
4	•	ii. Applies the mathematical relationship between mass and weight (F _g = mg)

- 1. See Techniques, Science, Separating mixtures (Techniques-Science, b).
- 2. These concepts are presented under The Living World, Life-sustaining processes (LW, B, f).
- 3. See Technological World, Mechanical Engineering, Technological systems, Transformation of energy (TW, B, 2, c).
- 4. The definition of temperature was covered in Cycle One. See *Properties* above (MW, A, 1, c).
- 5. This concept, related to program section Forces and motion (TW, B, 1), is presented here.
- 6. See Law of the conservation of energy (MW, B, 4, b).
- 7. See The Material World, Changes, Changes in matter, Mixtures (MW, B, 1, b).
- 8. See The Technological World, Electrical engineering (TW, C).
- 9. For the concepts of *Force and motion* in Secondary Cycle One, as presented in the program, see *The Technological World*.

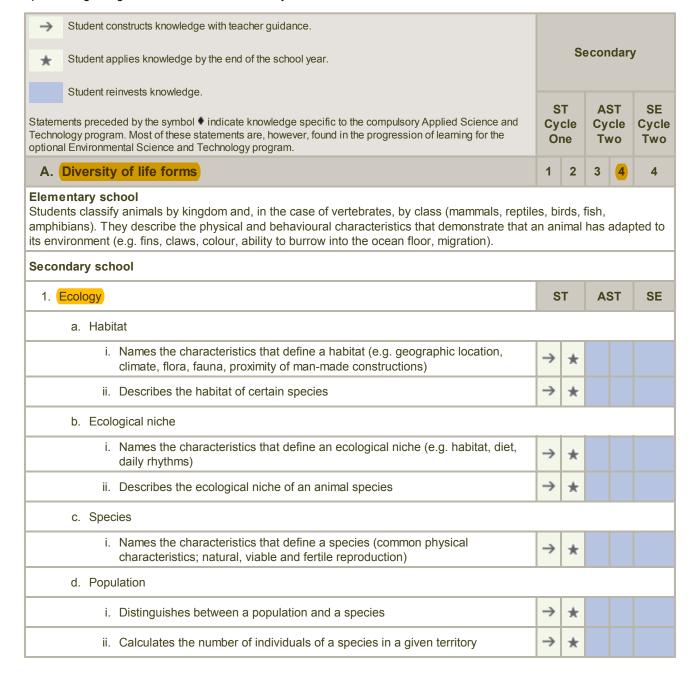
Applied General Education Path

The Living World

In The Living World, students acquire scientific and technological knowledge pertaining to life as it relates to molecules, cells, organisms and ecosystems.

In secondary school, students explore increasingly complex phenomena and technical objects and seek answers and solutions to a variety of problems. They acquire scientific knowledge about The Living World that helps them understand and explain the factors at play in different scientific issues. This knowledge, along with the knowledge they acquire in other areas of the program, particularly The Earth and Space, enables them to understand the complexity of the relationships between living organisms and their planet. Students refine their understanding of the concepts related to The Living World by using the experimental method, the observation method and modelling.

In Secondary III, students explore applications related to the seven technological fields, which enables them to make connections between technology and The Living World. In Secondary IV, they continue constructing and applying their knowledge about The Living World by analyzing and designing a variety of applications related to the same technological fields. Thus they acquire a better understanding of the contribution of science and technology to human and environmental health. In the optional Science and the Environment program, students consolidate their knowledge and form their own opinions regarding two environmental issues they will be asked to examine.



e. Dynamics of ecosystems	
i. (Ecosystems)	
 Defines an ecosystem as the relationships between the individuals in a community and abiotic factors in the environment 	
ii. Disturbances	
 Defines a disturbance in a community 	
 Explains the effects of certain factors that disturb the ecological balance (e.g. human activity, natural disasters) 	
iii. Trophic relationships	
 ─ Describes the trophic levels (producers, consumers, decomposers) 	
 Explains the relationships between the trophic levels of a food web 	
iv. Primary productivity	
 Defines primary productivity as the quantity of organic matter produced by plants in a given territory 	
 Explains the effects of certain factors on primary productivity (e.g. bees help pollinate fruit trees, pathogenic microorganisms hinder plant growth) 	
v. Material and energy flow	
 ─ Describes material and energy flow in an ecosystem 	
vi. Chemical recycling	
 Describes certain processes underlying chemical recycling (e.g. action of microorganisms and decomposers, erosion) 	
vii. Factors that influence the distribution of biomes	
 Describes the geographical and climatic factors that affect the distribution of biomes (e.g. latitude, humidity, temperature, salinity) 	
f. Ecotoxicology	
i. Contaminant ¹	
Defines a contaminant as an agent that causes changes in the physical, chemical or biological properties of an environment or an organism	*
ii. Bioaccumulation	
Defines bioaccumulation as the process by which a contaminant from the environment or food supply accumulates in an organism	*
Explains bioaccumulation in food chains (biomagnification)	*
iii. Bioconcentration	
Defines bioconcentration as a special case of bioaccumulation by which an organism accumulates a contaminant through direct contact with its environment (from sources other than food)	*
iv. Toxicity threshold	
Defines the toxicity threshold of a substance as the minimum concentration of a substance that produces a significant harmful effect in an organism (mg/kg of the organism's mass)	*
Describes factors that influence the toxicity of a contaminant (e.g. concentration, characteristics of the environment into which it is released, nature of the organisms with which it is in contact, duration of exposure)	*

2. Diversity of life forms		ST		AS	ST	SE
a. Physical and behavioural adaptation						
 Describes physical adaptations that enable animals and plants their chances of survival (e.g. coat colour matched to the envir of leaves) 		→	k			
 ii. Describes behavioural adaptations that enable animals and please their chances of survival (e.g. movement in groups, phototropic 		>	*			
b. Evolution						
i. Describes the stages in the evolution of living organisms	-	→	k			
ii. Explains the natural selection process	-	→	k			
c. Taxonomy						
 Defines taxonomy as a system for classifying living organisms most part on their anatomical and genetic characteristics 	based for the	>	*			
ii. Identifies a species using a taxonomic key	-	→	k			
d. Genes and chromosomes ²						
i. Locates chromosomes in the cell	-	→	k			
ii. Defines a gene as part of a chromosome	-	> 7	k			
iii. Describes the role of genes (transmission of hereditary charac	teristics) -	→ n	k			
B. Life-sustaining processes	1	1 :	2	3	4	4
Students explain the essential needs of living organisms (e.g. food, respiration) (transformation of energy, growth, maintenance of systems and body temperature photosynthesis, which they distinguish from respiration. Secondary school						
a. Characteristics of living things						
i. Describes certain characteristics common to all living things (note that it is not the common to all living things (not the common to all living things (not the common to all living things (not the common to all living things). I describes certain characteristics common to all living things (not the common to all living things).	utrition,	→	k			
b. Plant and animal cells						
i. Defines the cell as the structural unit of life	-	→	k			
ii. Names vital functions carried out by cells	_	> 7	*			
iii. Distinguishes between animal and plant cells		> 7	*			
c. Cellular components visible under a microscope						
i. Identifies the main cellular components visible under a microsomembrane, cytoplasm, nucleus, vacuoles)	cope (cell	→	k			
ii. Describes the role of the main cellular components visible und	er a microscope	> 7	k			
d. Inputs and outputs (energy, nutrients, waste)						
i. Names cellular inputs	-	→	k			
ii. Names cellular outputs	-	→ 7	k			
e. Osmosis and diffusion						
i. Distinguishes between osmosis and diffusion	_	>	*			

i. Names th	ne inputs and outputs involved in photosynthesis	\rightarrow	*			
ii Renrese	nts the photosynthesis reaction in a balanced equation					*
·	· · ·	\rightarrow				^
	ne inputs and outputs involved in respiration	7	*			
	nts the photosynthesis reaction in a balanced equation					*
C. Systems		1	2	3	4	4
Elementary school Students describe the fun and systems with their ma	ctions of certain parts of their anatomy (limbs, head). They associate p in functions.	arts	of ar	nimal	ana	tomy
Secondary school						
	Nutrition					
Digestive system		S	Т	A	ST	SE
a. Digestive tract						
	ne main parts of the digestive tract (mouth, esophagus, stomach, estine, large intestine, anus)			*		
	the role of the digestive tract (decomposition of food, absorption of and water, elimination of waste)			*		
	s the functions of the main organs that make up the digestive tract stomach, small intestine, large intestine)			*		
b. Digestive gland	ds					
	ne main digestive glands (salivary glands, gastric glands, pancreas, estinal glands)			*		
	s the function of the main digestive glands (e.g. secretion of saliva, nzymes, digestive juices, bile)			*		
c. Types of foods						
	s the main biological functions of the different food constituents roteins, carbohydrates, fats, vitamins, minerals)			*		
	es food constituents with their main sources (e.g. proteins with meat substitutes)			*		
d. Energy value of	of different foods					
i. Evaluate	s the energy and nutritional value of different foods			*		
e. Transformation	n of food					
	es the two types of transformation of food that take place in the system (mechanical, chemical)			*		
	es the organs in the digestive tract with the type of transformation form (e.g. mechanical action of teeth, chemical action of glands)			*		
2. Respiratory and circ	ulatory and systems	S	Т	A	ST	SE
a. Respiratory sy	stem					
	ne main parts of the respiratory system (nasal cavity, pharynx, bronchi, lungs)			*		
	the role of the respiratory system (gaseous exchanges between the d the surrounding air)			*		
iii. Describe	s the function of the nasal cavity and lungs			*		

1.	Describes the main function of plasma (transportation of the blood's soluble and formed elements)			*		
ii.	Names the formed elements of the blood (red blood cells, white blood cells, platelets)			*		
iii.	Describes the main function of the formed elements of the blood			*		
c. Con	npatibility of blood types					
i.	Determines the compatibility or incompatibility of blood types (e.g. an individual with type A blood can only receive type O- or type A- blood)			*		
d. Circ	ulatory system					
i.	Names the main parts of the circulatory system (heart, types of blood vessels, pulmonary and systemic circulation)			*		
ii.	Explains the role of the circulatory system (transportation and exchange of gases, nutrients and waste)			*		
iii.	Describes the function of the main parts of the circulatory system (heart, arteries and veins, capillaries)			*		
e. Lym	phatic system					
i.	Names the main parts of the lymphatic system (lymph, antibodies)			*		
ii.	Explains the role of the lymphatic system (circulation of antibodies outside the blood vessels)			*		
iii.	Describes two ways of acquiring active immunity (production of antibodies, vaccination)			*		
3. Excretory	system	S	Т	AS	Т	SE
a. Urin	ary system					
i.	Names the main parts of the urinary system (kidneys, ureters, bladder, urethra)			*		
ii.	Explains the role of the excretory system (filtration of the blood, elimination of cellular waste)			*		
iii.	Describes the function of the kidneys and bladder			*		
b. Con	nponents of urine					
i.	Names the main components of urine (water, mineral salts, urea)			*		
c. Maii	ntaining a balanced metabolism					
j.	Explains the role of the kidneys, lungs and sweat glands in maintaining a balanced metabolism			*		
	Relationships					
1. Nervous a	and musculoskeletal systems	S	Т	AS	Т	SE
a. Cen	tral nervous system					
	ldentifies the parts of the central nervous system (brain, spinal cord)			*		
i.				*		
i. ii.	Identifies the parts of the central nervous system (brain, spinal cord) Explains the role of the central nervous system (e.g. to manage complex					
i. ii. iii.	Identifies the parts of the central nervous system (brain, spinal cord) Explains the role of the central nervous system (e.g. to manage complex behaviours and process sensory information and the related responses)			*		
i. ii. b. Peri	ldentifies the parts of the central nervous system (brain, spinal cord) Explains the role of the central nervous system (e.g. to manage complex behaviours and process sensory information and the related responses) Describes the functions of the brain and the spinal cord			*		

 Explains the role of the peripheral nervous system (transportation of nerve impulses from the senses to the brain and from the brain to the muscles) 	te	*
ii. Neural inflow		
 Associates nerves with the transmission of nerve impulses 	1	•
Distinguishes between voluntary acts and reflex arcs	1	r
c. Sensory receptors		
i. Eye		
Names the parts of the eye involved in vision (iris, cornea, crystalline lens, retina)	4	r
 Describes the function of the main parts of the eye 	1	r
ii. Ear		
Names the main parts of the ear involved in hearing and balance (auditory canal, ear drum, ossicles, cochlea, semicircular canals)	3	+
Describes the function of the main parts of the ear involved in hearing	1	r
Describes the role of the semicircular canals in maintaining balance	4	r
iii. Tongue		
 Describes the function of the taste buds on the tongue (transformation and transmission of flavours: sweet, salty, sour, bitter, umami) 	4	r
iv. Nose		
 Names the parts of the nose involved in smelling (nasal cavity, olfactory bulb) 	1	t
Describes the function of the olfactory bulb	t d	t
v. Skin		
 Describes the function of the skin's sensory receptors (transformation and transmission of feelings of pressure, temperature and pain) 	1	+
d. Musculoskeletal system		
i. Function of bones, joints and muscles		
 Names the main parts of the skeleton (head, thorax, spinal column, upper and lower limbs) 	t e	r
Describes the functions of the main parts of the skeleton (e.g. the spinal column protects the spinal cord and allows the trunk to move)	1	F
Explains the role of the musculoskeletal system	1	t
Describes how pairs of antagonistic muscles work (e.g. biceps, triceps)	1	+
Describes how joints work (linking bone to bone, mobility)	7	*
ii. Types of muscles		
 Associates the different types of muscles (smooth, skeletal, heart) with the tissues in which they are found 	4	*
iii. Types of joint movement		
Describes types of joint movement (e.g. flexion, rotation)	7	F

D. Sui	vival of species	1	2	3	4	4
	ry school lescribe the growth and reproduction of flowering plants and different animals.					
Seconda	y school					
1. Rep	roduction	S	Т	AS	ST	SE
a.	Asexual and sexual reproduction					
	 Distinguishes between asexual and sexual reproduction (e.g. sexual reproduction involves gametes) 	\rightarrow	*			
b.	Reproductive mechanisms in plants					
	i. Describes asexual reproductive mechanisms in plants (e.g. cutting, layering)	\rightarrow	*			
	ii. Describes the sexual reproductive mechanism in plants (flowering plants)	\rightarrow	*			
C.	Reproductive mechanisms in animals					
	 Describes the roles of the male and female in the reproduction of certain types of animals (e.g. birds, fish, mammals) 	\rightarrow	*			
d.	Reproductive organs					
	 Names the main male and female reproductive organs (penis, testicles, vagina, ovaries, Fallopian tubes, uterus) 	\rightarrow	*			
e.	Gametes					
	i. Names the male and female gametes	\rightarrow	*			
	ii. Describes the role of gametes in reproduction	\rightarrow	*			
f.	Fertilization					
	i. Describes fertilization in humans	\rightarrow	*			
g.	Pregnancy					
	 Names the stages of human development during pregnancy (zygote, embryo, fetus) 	\rightarrow	*			
h.	Stages of human development					
	 Describes the stages of human development (childhood, adolescence, adulthood) 	\rightarrow	*			
i.	Contraception					
	 Describes contraceptive methods (e.g. condom, ovulation suppression agents) 	\rightarrow	*			
	 Describes the advantages and disadvantages of different contraceptive methods 	\rightarrow	*			
j.	Methods of preventing the implantation of the zygote in the uterus	_				
	 Names methods of preventing the implantation of the zygote in the uterus (intrauterine device, day-after pill) 	\rightarrow	*			
k.	Sexually transmitted and blood-borne diseases ⁴					
	i. Names sexually transmitted and blood-borne diseases	\rightarrow	*			
	 Describes behaviours to prevent contracting sexually transmitted and blood-borne diseases (e.g. wearing a condom) 	\rightarrow	*			
	iii. Describes responsible behaviours to adopt after being diagnosed with a sexually transmitted or blood-borne disease (e.g. informing one's partner)	\rightarrow	*			

2. Cell division ⁵	S	Т	AST	SE
a. Mitosis			_	
i. Describes the functions of mitosis (reproduction, growth, regeneration)			*	
b. Meiosis				
i. Describes the function of meiosis (production of gametes)			*	
c. Genetic diversity				
i. Associates genetic diversity with sexual reproduction			*	
Reproduction				
Reproductive system	S	Т	AST	SE
a. Puberty (male and female)				
 Describes physical and psychological changes that occur at puberty (e.g. appearance of body hair, voice change, ability to procreate, need for independence) 			*	
b. Hormone regulation in men				
i. Spermatogenesis				
 Names the hormones responsible for the formation of spermatozoa (follicle stimulating hormone [FSH], luteinizing hormone [LH], testosterone) 			*	
ii. Erection				
 Describes the physiology of erection 			*	
iii. Ejaculation				
 Explains the function of ejaculation in reproduction 			*	
c. Hormone regulation in women				
i. Oogenesis				
 Names the hormones responsible for the maturation of the ovarian follicle (FSH, LH, estrogen, progesterone) 			*	
ii. Ovarian cycle				
 Describes the hormone changes that occur during the menstrual cycle 			*	
iii. Menstrual cycle				
 Describes the main stages in the menstrual cycle (e.g. menstruation, endometrium development, ovulation) 			*	

- 1. See The Earth and Space, Characteristics of the Earth, Contamination (ES, A, 2, h; ES, A, 3, c; ES, A, 4, f).
- 2. See The Living World, Survival of species, Cell division (LW, D, 2).
- 3. The Secondary IV concepts related to photosynthesis and respiration are presented in the program under *Material World*, *Changes*, *Chemical changes*.
- 4. This replaces the term "sexually transmitted diseases" used in the program.
- 5. For concepts related to *Genes and chromosomes* addressed in Cycle One, see *The Living World, Diversity of life forms* above (<u>LW, A, 2, d</u>).

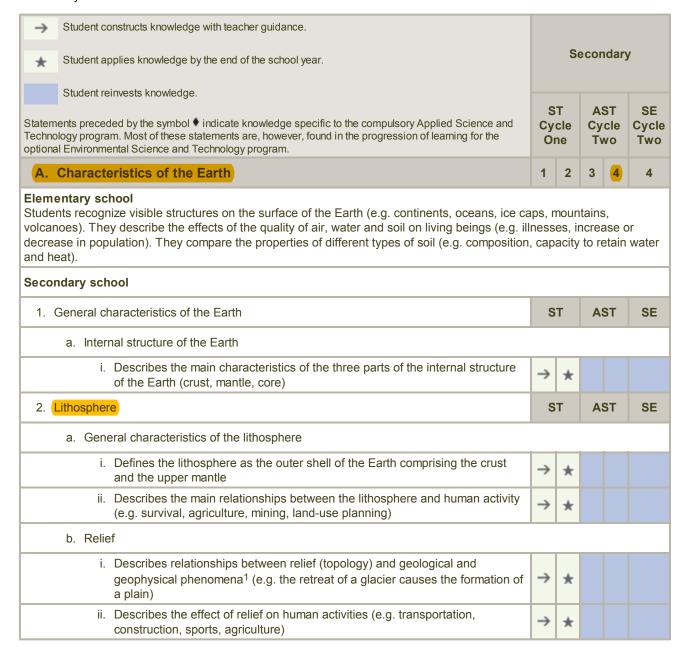
Applied General Education Path

The Earth and Space

In The Earth and Space, students acquire scientific and technological knowledge pertaining to space and interactions in the biosphere.

In secondary school, students explore different phenomena that occur on Earth and in space and related technical objects. They seek answers and solutions to a variety of problems. They acquire scientific knowledge about The Earth and Space that helps them explain the factors at play in different scientific issues. This knowledge, along with the knowledge they acquire in other areas of the program, enables them to understand scientific models, theories and laws. Students refine their understanding of the concepts related to The Earth and Space by using the experimental method, technological analysis and modelling.

In Secondary IV, students explore applications related to the seven technological fields, which enables them to make connections between technology and The Earth and Space. Thus they continue constructing and applying their scientific and technological knowledge and further explore the impact of technology on the biosphere. In the optional Science and the Environment program, students consolidate their knowledge and form their own opinions regarding two environmental issues they will be asked to examine.



c. Types of rocks				
 Describes the formation of three types of rock: igneous, metamorphic, sedimentary 	\rightarrow	*		
 ii. Classifies rocks by method of formation (e.g. granite is an igneous rock, lime is a sedimentary rock, slate is a metamorphic rock) 	\rightarrow	*		
iii. Distinguishes between rocks and minerals	\rightarrow	*		
d. (Minerals)				
 Names basic minerals based on their properties (e.g. colour, hardness, magnetism) 	\rightarrow	*		
ii. Distinguishes between minerals and ore			4	
iii. Describes some of the environmental impacts of mining and of the transformation of minerals			3	
e. Types of soil				
i. Classifies soils based on their composition (e.g. sand, clay, organic material)	\rightarrow	*		
f. Soil horizons (profile)				
 Describes the structure of a soil (superimposition of layers of various compositions and thicknesses) 				*
 ii. Explains the chemical and biological reactivity of a soil it based on its composition (e.g. oxidation, acid-base neutralization, decomposition) 				*
g. Buffering capacity of the soil				
i. Defines the buffering capacity of a soil as its ability to limit pH variations				*
ii. Explains the advantages of a good soil buffering capacity				*
h. Contamination				
i. Names soil contaminants ²				*
3. (Hydrosphere)	S	Т	AST	SE
a. General characteristics of the hydrosphere				
 Describes the distribution of fresh water and salt water on the Earth's surface (e.g. glaciers contain inaccessible fresh water) 	\rightarrow	*		
ii. Describes the main interactions between the hydrosphere and the	\rightarrow	*		
atmosphere (e.g. heat exchanges, climate regulation, meteorological phenomena)	7			
phenomena)			4	•
b. Catchment area			9	
b. Catchment area i. Defines a catchment area as a territory surrounding a waterway ii. Describes some of the impacts of human activity on the waterways in a			9	
b. Catchment area i. Defines a catchment area as a territory surrounding a waterway ii. Describes some of the impacts of human activity on the waterways in a catchment area			4	*
b. Catchment area i. Defines a catchment area as a territory surrounding a waterway ii. Describes some of the impacts of human activity on the waterways in a catchment area c. Contamination			9	
phenomena) b. Catchment area i. Defines a catchment area as a territory surrounding a waterway ii. Describes some of the impacts of human activity on the waterways in a catchment area c. Contamination i. Names water contaminants ³				

4. (Atmosphere)	S	T	AS	ST	SE
a. General characteristics of the atmosphere					
 Locates the main layers of the atmosphere (troposphere, stratosphere, mesosphere, thermosphere) 	\rightarrow	*			
 ii. Describes the composition of pure air at sea level (nitrogen, oxygen, carbo dioxide, water vapour) 	on →	*			
iii. Describes the relationships between the atmosphere and certain human activities (e.g. recreation, transportation, energy consumption)	\rightarrow	*			
b. Greenhouse effect					
i. Describes the greenhouse effect					*
 ii. Explains some consequences of a higher concentration of greenhouse gas (e.g. global warming that could result in higher sea levels, disturbances in ecosystems or the melting of glaciers) 	ses				*
c. (Air mass)					
i. Describes the properties of an air mass (temperature, humidity, pressure)				*	
ii. Explains the formation of clouds when two different air masses meet				*	
d. Atmospheric circulation					
Describes the main factors responsible for atmospheric circulation (e.g. pressure variations, uneven heating of the Earth's surface)					*
 Describes the effect of prevailing winds on the dispersal of air pollutants in given region 	ı a				*
e. Cyclone and anticyclone					
i. Explains the formation of cyclones (low-pressure areas) and anticyclones (high-pressure areas)				*	
f. Contamination					
i. Names air contaminants ⁴					*
B. Geological and geophysical phenomena	1	2	3	4	4
Elementary school Students explain the water cycle (evaporation, condensation, precipitation, runoff, infiltration of precipitation (rain, snow, hail, freezing rain). Concepts related to energy play an important school program. Students explain that sunlight, moving water and wind are renewable energy differentiate them from nonrenewable energy resources such as fossil fuels (e.g. gasoline, pas). They describe technologies used to convert renewable energy into electricity (hydroel solar panels).	nt role in to gy resour propane,	he el ces. butar	emei They ne, o	ntary / il, na	tural
Secondary school					
a. Tectonic plate					
 Describes the main elements of the theory of tectonic plates (e.g. plate, subduction zone, mid-oceanic ridge) 	\rightarrow	*			
b. Orogenesis					
 Describes the formation of mountains, folding and breaks (tectonic plate movements) 	\rightarrow	*			
c. Volcano					
i. Describes a volcanic eruption	\rightarrow	*			
ii. Describes the geographical distribution of volcanoes	\rightarrow	*			

d.	Earthquake					
	 Describes the processes that cause earthquakes (e.g. tectonic plate movements, slides) 	\rightarrow	*			
e.	Erosion					
	 Describes different types of erosion (e.g. soils dried by the wind, fragmentation of rocks caused by water freezing and thawing) 	\rightarrow	*			
f.	Winds					
	 Names the main factors responsible for wind (e.g. convection movements, movement of air masses) 	\rightarrow	*			
g.	Water cycle					
	i. Explains the water cycle (phase changes, energy exchanges)	\rightarrow	*			
h.	Natural energy sources					
	 Describes the role of solar energy as a natural energy source (e.g. wind, tornadoes, hurricanes, storms) 	\rightarrow	*			
i.	Renewable and nonrenewable energy resources					
	Distinguishes between renewable and nonrenewable energy resources (e.g. Sun, molten rock, moving water, oil)	\rightarrow	*			
	ii. Describes technologies used to produce electricity using the energy resources in the lithosphere, hydrosphere and atmosphere				*	
	Describes the main impact of the use of energy resources in the lithosphere, hydrosphere and atmosphere				*	
C. Ast	onomical phenomena	1	2	3	4	4
and moons	ry school earn that the cycle of day and night is related to the Earth's rotation. They distinguish in our solar system. They describe seasonal changes (e.g. temperature variations, len), in particular the apparent position of the Sun and its influence on the length of sha	umino	osity			
Secondar	y school					
1. Cond	epts related to astronomy	S	Т	Α	ST	SE
a.	Universal Gravitation					
	i. Defines gravitation as a force of mutual attraction between bodies	\rightarrow	*			
b.	Earth-Moon system					
	 Describes the tides in terms of the gravitational effect of the Earth-Moon system 				*	
C.	Light					
	i. Defines light as a form of radiant energy ⁵	\rightarrow	*			
	 Describes properties of light (propagation in a straight line, diffuse reflection by surfaces) 	\rightarrow	*			
	iii. Explains different phenomena using the properties of light (cycles of day and night, seasons, phases of the Moon, eclipses)	\rightarrow	*			
d.	Solar energy flow					
	 Describes the main factors that affect the quantity of solar energy that reaches the Earth's surface (e.g. reflection and absorption of solar energy by the atmosphere or surfaces) 				*	

2. Sola	system	S	Т	AST	SE
a.	Characteristics of the solar system				
	 i. Compares some of the characteristics of the planets in our solar system (e.g. distances, relative size, composition) 	\rightarrow	*		
b.	Cycles of day and night				
	i. Explains the alternation of day and night in terms of the Earth's rotation	\rightarrow	*		
C.	Phases of the Moon				
	i. Explains the phases of the lunar cycle	\rightarrow	*		
d.	Eclipses				
	i. Explains a lunar or solar eclipse	\rightarrow	*		
e.	Seasons				
	 i. Explains the phenomenon of seasons in terms of the position of the Earth with respect to the Sun (tilt, revolution) 	\rightarrow	*		
f.	Comets				
	 Describes the main parts of a comet (core of ice and rock, tail of gas, and tail of dust) 	\rightarrow	*		
g.	Aurora borealis (northern lights)				
	 Locates the geographic regions where the aurora borealis occurs (polar regions) 	\rightarrow	*		
	ii. Identifies the atmospheric layer in which the aurora borealis occurs	\rightarrow	*		
h.	Meteoroid impact				
	 i. Identifies traces left by meteoroid impacts in Québec (e.g. craters, astroblemes) 	\rightarrow	*		

- 1. See The Earth and Space, Geological and geophysical phenomena below (ES, B).
- 2. See The Living World, Ecotoxicology, Contaminant (LW, A, 1, f, i).
- 3. See The Living World, Ecotoxicology, Contaminant (LW, A, 1, f, i).
- 4. See The Living World, Ecotoxicology, Contaminant (LW, A, 1, f, i).
- 5. See The Material World, Changes, Transformation of energy, Forms of energy (MW, B, 4, a).

Applied General Education Path

The Technological World

In The Technological World, students acquire and apply scientific and technological knowledge.

In Secondary school, students analyze and design increasingly complex technical objects and seek solutions to increasingly sophisticated technological problems. The technical and technological knowledge they acquire helps them understand the objects and factors at play in different scientific issues as well as evaluate possible technological solutions. It also helps them apply knowledge they acquire in other areas of the program, in particular The Material World.

In Secondary III, students analyze and design technical objects, processes or systems related to the seven technological fields, which enables them to make connections between human beings and technology and, consequently, to integrate their knowledge of The Living World. In Secondary IV, they continue constructing their scientific and technological knowledge and examine the influence of technology on the world around us by exploring a variety of applications related to the same technological fields. Thus they discover how technology helps us understand and improve our world. In the optional Science and the Environment program, students consolidate their knowledge and form their own opinions regarding two environmental issues they will be asked to examine. They can thus apply their knowledge of the Technological World in new contexts.

Student	constructs knowledge with teacher guidance.					
- Student	constructs knowledge with teacher guidance.					
★ Student	applies knowledge by the end of the school year.		Secondary			/
Student	reinvests knowledge.					
Technology prog	eded by the symbol • indicate knowledge specific to the compulsory Applied Science and ram. Most of these statements are, however, found in the progression of learning for the nental Science and Technology program.	Су	cle ne	Су	ST cle vo	SE Cycl Two
A. Graphi	cal language ¹	1	2	3	4	4
Elementary s Students learn	chool symbols associated with motion and parts and use them to produce or interpret of	liagra	ams	or dr	awin	gs.
Secondary so	hool					
a. Dia	gram of principles (design plan)					
	i. Defines a diagram of principles as a representation used to effectively explain the operation of a technical object	\rightarrow	*			
i	i. Associates the functional elements of a technical object with the appropriate diagram of principles	\rightarrow	*			
ii	Explains the operation of a simple technical object by drawing a diagram illustrating the active forces and the resulting motion	\rightarrow	*			
iv	 Names the subassemblies and parts essential to the operation of a technical object 	\rightarrow	*			
\	. Indicates certain principles of simple machines illustrated in a technical object (e.g. a lever in a wheelbarrow, a wedge in an axe)	\rightarrow	*			
b. Co	nstruction diagram (technical diagram)					
	Defines a construction diagram as a representation used to effectively explain the construction and assembly of a technical object	\rightarrow	*			
i	 Associates the shape and arrangement of parts of technical objects with the appropriate construction diagram 	\rightarrow	*			
ii	Explains the construction of a simple technical object by drawing a diagram illustrating the assembly and arrangement of parts	\rightarrow	*			
iv	. Names the components of a simple technical object	\rightarrow	*			
\	. Indicates the links and guiding controls on a construction diagram	\rightarrow	*			

	i. Chooses the appropriate type of diagram for a given representation (e.g. uses a construction diagram to represent assembly solutions, a diagram of principles to represent the operation of an object)	\rightarrow	*
	ii. Represents different types of motion related to the operation of an object using the appropriate symbols (rectilinear translation, rotation, helical)	\rightarrow	*
d.	Geometric lines		
	Associates a drawing with a combination of geometric lines (e.g. the drawing of a rounded corner of a table is an arc joined to two sides of a right angle)	*	
e.	Basic lines		
	Names basic lines in a drawing (visible contour, hidden contour, centre, extension, dimension lines)	*	
	ii. Associates the basic lines in a drawing with the contours and details of a simple part	*	
f.	Orthogonal projections		
	Associates the types of projection with their use (multiview and isometric projections)	*	
	ii. Interprets drawings representing parts in multiview orthogonal projection	*	
	iii. Represents simple shapes in multiview orthogonal projection	*	
	iv. Represents simple shapes in isometric projection	*	
+	v. Interprets assembly drawings of technical objects consisting of a small number of parts		*
g.	Scales ³		
	Associates scales with their use (actual-size representation, reduction or enlargement of an object)	*	
	ii. Chooses a simple scale for a drawing (e.g. 1 : 1, 1 : 2, 5 : 1)	*	
	iii. Takes the scale into account when interpreting drawings	*	
h.	Forms of representation		
	i. Defines perspective drawing, oblique projection and axonometric projection	*	
	ii. Sketches simple objects freehand using different forms of representation	*	
i.	Axonometric projection: exploded view (reading)		
+	i. Names the characteristics of an exploded view	*	
+	ii. Explains the purpose of exploded views (projection accompanying the assembly instructions or specifications for an object)	*	
j.	Cross-sectional views and sections		
	i. Cross-sectional views		
	Describes the purpose of cross-sectional views in technical drafting	*	
	Interprets a technical drawing with cross-sectional views	*	
	Represents a simple shape in a cross-sectional view	*	

	 Describes the purpose of removed sections and revolved sections 			*		
k.	Dimensioning and tolerances					
	i. Dimensioning					
	 Describes the main dimensioning rules (e.g. to make a drawing easy to read, avoid crossing dimensioning lines) 			*		
	 Interprets technical drawings including the dimensions required for manufacturing purposes 			*		
	ii. Tolerances					
+	 Defines tolerance as the required manufacturing precision (dimensions indicated on the drawing, along with allowances) 			*		
	iii. (Functional dimensioning)					
*	 Defines functional dimensioning as the set of specific tolerances related to certain parts responsible for the smooth operation of an object (e.g. the distance between two axes is a determining factor in the operation of sprocket wheels in a gear assembly) 				*	
I.	Developments (prism, cylinder, pyramid, cone)					
*	i. Associates the development of three-dimensional shapes with the construction of objects from sheet stock (e.g. cardboard boxes, metal air ducts)				*	
•	ii. Draws developments of simple solids (e.g. pyramid, cylinder, cube)				*	
B. Med	chanical engineering	1	2	3	4	4
	s, cams, springs), distinguish between translation and rotation and describe a simple stion (e.g. in a door lock, the lever rotates and the motion of the bolt is rectilinear trans			of m	necha	inical
1. Forc	es and motion ⁴	S	Т	A	ST	SE
a.	Types of motion					
	 i. Identifies parts that move in a specific way in a technical object (rectilinear translation, rotation, helical) 	\rightarrow	*			
b.	Effects of a force					
	 Explains the effects of a force in a technical object (change in the motion of an object, distortion of a material) 	\rightarrow	*			
c.	Simple machines					
	 i. Identifies wheels, inclined planes and levers in simple technical objects (e.g. a wheelbarrow is made up of a second-class lever and a wheel) 	\rightarrow	*			
	ii. Describes qualitatively the mechanical advantages of different types of levers (first-class, second-class, third-class) in different applications	\rightarrow	*			
2. Tech	nological systems	S	Т	A	ST	SE
a.	System					
	 i. Identifies a system (set of connected elements that interact with each other) in a technical object or technological application 	\rightarrow	*			
	a technical object of technological application					
	ii. Describes the overall function of a technological system	\rightarrow	*			
	, , , , , , , , , , , , , , , , , , , ,	→	*			

b.	Components of a system					
	 Describes the role of the components of a technological system (e.g. explains the role of the parts of a lighting system) 	\rightarrow	*			
C.	Energy transformations ⁵					
	i. Associates energy with radiation, heat or motion	\rightarrow	*			
	ii. Defines energy transformations	\rightarrow	*			
	iii. Identifies energy transformations in a technical object or technological system	\rightarrow	*			
3. Engi	neering	S	Т	A	ST	SE
a.	Basic mechanical functions (links, guiding control)					
	i. Describes the role of links and guiding controls in a technical object	\rightarrow	*			
	ii. Identifies a guiding control in a technical object, as well as the related links (e.g. a pizza wheel is guided by a pivot, which links it to the handle)	\rightarrow	*			
b.	Typical mechanical links					
	i. Describes the advantages and disadvantages of different types of links			*		
	ii. Names the types of links used in a technical object (e.g. the spiral link between a jar and its lid)			*		
C.	Linking of mechanical parts					
	i. Describes the characteristics of the links in a technical object (direct or indirect, rigid or flexible, removable or permanent, partial or complete)				*	
	ii. Determines the desirable characteristics of links in the design of a technical object				*	
	iii. Judges the choice of assembly solutions in a technical object				*	
•	iv. Explains the purpose of limiting motion (degree of freedom) in a technical object (e.g. some hinges limit how far a cupboard door can open, preventing it from hitting the wall)				*	
d.	Typical functions					
	i. Defines the typical functions (linking, guiding, sealing, lubricating)			*		
	ii. Associates a typical function with certain parts of a technical object			*		
	Explains the choice of a type of link in a technical object (e.g. using a screw makes it possible to attach and remove a battery case)			\rightarrow	*	
e.	Guiding controls					
	i. Explains the choice of a type of guiding control in a technical object (e.g. the slide guides a drawer and reduces friction)				*	
f.	Adhesion and friction of parts					
+	 Describes the advantages and disadvantages of the adhesion and friction of parts in a technical object 				*	
g.	Motion transmission systems					
	i. Identifies motion transmission systems in technical objects	\rightarrow	*			
h.	Function, components and use of motion transmission systems					
	 Names motion transmission systems in technical objects (friction gears, pulleys and belt, gear assembly, sprocket wheels and chain, wheel and worm gear) 			*		

(e.g. in a bicycle, the gear assembly on the crankset is the driving unit, sprocket wheel on the rear wheel is the receiving unit, and the chain is intermediate unit)				*		
 iii. Describes the speed changes or reversibility of a motion transmission s (e.g. a sprocket wheel that is replaced by a smaller wheel or a wheel wi fewer teeth increases rotation speed) 				*		
i. Construction and characteristics of motion transmission systems						
 i. Explains the choice of a motion transmission system in a technical objection (e.g. using a gear assembly rather than friction gears to get better enging torque and avoid slipping) 					*	
j. Motion transformation systems						
i. Identifies motion transformation systems in technical objects	-	>	*			
k. Function, components and use of motion transformation systems						
 Names motion transformation systems in technical objects (e.g. screw of system, cam and roller, connecting rod and crank, rack and pinion) 	gear			*		
 Describes the functions of the components of a motion transformation s (e.g. in a double-lever corkscrew, the pinion is the driving unit and the r the receiving unit) 				*		
 Describes speed changes or the reversibility of a motion transformation system (e.g. the cam and roller is a nonreversible motion transformation system) 				*		
I. Construction and characteristics of motion transformation systems						
i. Explains the choice of a motion transformation system (screw gear syst cams, connecting rods, cranks, slides, rotating slider crank mechanisms rack-and-pinion drive) in a technical object (e.g. most car jacks use a segurar system rather than a rack-and-pinion system, because the force of arm on the small crank provides more thrust and because, given that it nonreversible, the system is safer)	s, crew f the				*	
ii. Distinguishes between cams and eccentrics					*	
m Chand shanger						
m. (Speed changes)						
i. Uses systems that allow for speed changes in the design of technical o	bjects			*		
				*	*	
 i. Uses systems that allow for speed changes in the design of technical o ii. Explains speed changes in a technical object using the concepts of res 	isting	1	2	*	*	4
 i. Uses systems that allow for speed changes in the design of technical o ii. Explains speed changes in a technical object using the concepts of restorque and engine torque 	isting describe			3	4	
 i. Uses systems that allow for speed changes in the design of technical of ii. Explains speed changes in a technical object using the concepts of resistorque and engine torque. C. Electrical engineering Elementary school Students describe energy transformations and recognize them in different devices. They 	isting describe			3	4	
 i. Uses systems that allow for speed changes in the design of technical of it. Explains speed changes in a technical object using the concepts of restorque and engine torque. C. Electrical engineering Elementary school Students describe energy transformations and recognize them in different devices. They energy resources into electricity (e.g. wind turbines transform wind energy into electricity) 	isting describe			3	4	
 i. Uses systems that allow for speed changes in the design of technical of it. Explains speed changes in a technical object using the concepts of resistorque and engine torque. C. Electrical engineering Elementary school Students describe energy transformations and recognize them in different devices. They energy resources into electricity (e.g. wind turbines transform wind energy into electricity Secondary school 	isting describe			3	4	
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 i. Uses systems that allow for speed changes in the design of technical of it. Explains speed changes in a technical object using the concepts of rest torque and engine torque. C. Electrical engineering Elementary school Students describe energy transformations and recognize them in different devices. They energy resources into electricity (e.g. wind turbines transform wind energy into electricity Secondary school a. Power supply i. Defines power supply as the ability to generate electrical current ii. Determines the source of current in technical objects with an electrical of the concepts of restorated electrical current iii. 	describe ').			3 f tran	4 nsforr	
i. Uses systems that allow for speed changes in the design of technical of ii. Explains speed changes in a technical object using the concepts of resistorque and engine torque. C. Electrical engineering Elementary school Students describe energy transformations and recognize them in different devices. They energy resources into electricity (e.g. wind turbines transform wind energy into electricity secondary school a. Power supply i. Defines power supply as the ability to generate electrical current ii. Determines the source of current in technical objects with an electrical (e.g. chemical battery, solar cell, alternator, thermocouple, piezoelectrical current)	describe ').			3 f tran	4 nsforr	
i. Uses systems that allow for speed changes in the design of technical of ii. Explains speed changes in a technical object using the concepts of resistorque and engine torque. C. Electrical engineering Elementary school Students describe energy transformations and recognize them in different devices. They energy resources into electricity (e.g. wind turbines transform wind energy into electricity Secondary school a. Power supply i. Defines power supply as the ability to generate electrical current ii. Determines the source of current in technical objects with an electrical (e.g. chemical battery, solar cell, alternator, thermocouple, piezoelectrical b. Conduction, insulation and protection	describe //).			3 f tran	4 nsforr	

		IV.	Analyzes the factors that affect electrical conductivity (section, length, nature, temperature of conductor)				*	
	+	V.	Uses the colour code to determine the electrical resistance of a resistor				*	
	+	vi.	Describes the operation of a printed circuit				*	
	C.	Conf	trol					
		i.	Defines control as the ability to control the travel of electrical current			*		
		ii.	Describes different types of switches (lever, pushbutton, flip-flop, magnetic control)			*		
	+	iii.	Distinguishes between unipolar and bipolar switches				*	
	+	iv.	Distinguishes between unidirectional and bidirectional switches				*	
	d.	Tran	sformation of energy (electricity and light, heat, vibration, magnetism)					
		i.	Associates the transformation of energy with different components of a circuit (e.g. bulbs transform electrical energy into light and heat)				*	
		ii.	Describes the energy transformations that take place in electrical or electronic appliances (e.g. in a cell phone, electricity is transformed into light for the display and vibrations for the sound)				*	
	e.	Othe	er functions					
	*	i.	Describes the function of certain electronic components (condenser, diode, transistor, relay)				*	
D.	Mat	orial		1	2	3	4	4
leme	enta	ry sc	<mark>-</mark>					
leme tudei	enta nts d	ry sc lescri y sch	hool be the physical properties of certain materials. nool					QI
leme tudei	enta nts d ndar Mate	ry sc lescril y sch	hool be the physical properties of certain materials. nool esources		T		ST	SE
leme tudei	enta nts d ndar Mate	ry sc lescril y sch rial re	hool be the physical properties of certain materials. nool	S	т			SE
leme tudei	enta nts d ndar Mate a.	ry sc lescri y sch erial re Raw i.	hool be the physical properties of certain materials. hool esources materials Associates raw materials with the unprocessed materials used in an industry (e.g. bauxite is the raw material used in aluminum smelters)					SE
leme tudei	enta nts d ndar Mate a.	ry scherial re Raw i.	hool be the physical properties of certain materials. pool esources materials Associates raw materials with the unprocessed materials used in an industry (e.g. bauxite is the raw material used in aluminum smelters) erials Names the materials present in a technical object (e.g. a cooking pot is composed of two materials: a metal used to make the container and plastic	S	т			SI
leme tudei	enta nts d ndar Mate a.	ry sc lescrii y sch erial re Raw i. Mate	hool be the physical properties of certain materials. nool esources materials Associates raw materials with the unprocessed materials used in an industry (e.g. bauxite is the raw material used in aluminum smelters) erials Names the materials present in a technical object (e.g. a cooking pot is composed of two materials: a metal used to make the container and plastic used to coat the handle) Determines the origins of the materials present in a technical object (animal,	S	±			SE
leme tudei	enta nts d ndar Mate a.	ry scherial rename i. Mate	hool be the physical properties of certain materials. nool esources materials Associates raw materials with the unprocessed materials used in an industry (e.g. bauxite is the raw material used in aluminum smelters) erials Names the materials present in a technical object (e.g. a cooking pot is composed of two materials: a metal used to make the container and plastic used to coat the handle)	S →	*			SE
leme tudei	enta nts d ndar Mate a.	ry sc lescril y sch erial re Raw i. Mate i.	be the physical properties of certain materials. pool esources materials Associates raw materials with the unprocessed materials used in an industry (e.g. bauxite is the raw material used in aluminum smelters) erials Names the materials present in a technical object (e.g. a cooking pot is composed of two materials: a metal used to make the container and plastic used to coat the handle) Determines the origins of the materials present in a technical object (animal, plant, mineral, wood)	S →	*			SE
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b.	Mec	hanical properties					
	i.	Describes the mechanical properties of different materials (hardness, ductility, elasticity, malleability, corrosion resistance)			*		
C.	Cha	racteristics of mechanical properties					
	i.	Explains the choice of a material based on its properties (e.g. the malleability of aluminum makes it useful for making thin-walled containers)				*	
d.	Туре	es and properties					
	i.	Associates the use of different types of materials with their respective properties	:S				
		Ferrous alloys (cast iron is harder than steel)			*		
		 Nonferrous metals and alloys (the wire used in a dental appliance can be made of a nickel and titanium alloy, which has shape memory) 			*		
		 Wood and modified wood (e.g. oak is used for flooring because it is a hard wood that isis shock- and wear-resistant) 			*		
		 Plastics: thermoplastics (e.g. thermoplastics are used for prostheses because of their corrosion resistance and lightness) 			*		
		 Plastics: thermosetting plastics (e.g. Bakelite is used to mould electrical parts because it is a good electrical insulator) 				*	
		 Ceramics (e.g. ceramics are used in ovens because they are very hard and heat- and wear-resistant) 				*	
		 Composites (e.g. carbon fibre is used for hockey sticks because of its hardness, resilience and lightness) 				*	
e.	Cell						
•	i.	Describes how a living cell can be considered a material (e.g. artificial skin is manufactured from human tissue to treat burns)			*		
*	ii.	Compares a cell to a technological system (overall function, inputs, outputs, processes, control)			*		
f.	Mod	ification of properties					
	i.	Describes different treatments to prevent degradation of materials (e.g. metal) plating, antirust treatments, painting)				*	
g.	Heat	t treatments					
*	i.	Defines heat treatments as ways of changing the properties of materials (quenching increases hardness but fragility as well)				*	
E. Mai	nufac	cturing	1	2	3	4	4
systems (ed)	are int e.g. wa ateria	thool roduced to the design and construction of instruments, tools, machines, structurents ater filtration), models (e.g. glider) and simple electrical circuits. They trace parts als using the appropriate tools. They use a variety of assembly methods (e.g. screeners, nuts) and tools to obtain an aesthetic finish.	and	cut	them	out	
Secondar	y sch	nool					
a.	Spec	cifications					
	i.	Defines specifications as a set of constraints associated with the design of a technical object	\rightarrow	*			
	ii.	Evaluates a prototype or technical object based on the environments described in the specifications (human, technical, industrial, economic, physical, environmental)	\rightarrow	*			
b.	Man	ufacturing process sheet					
	i.	Defines a manufacturing process sheet as a set of steps to follow to machine the parts that make up a technical object	\rightarrow	*			

	 Follows a process and assembly sheet to construct an object consisting of few components or to construct part of that object 	\rightarrow	*			
C.	Shaping					
	i. Machines and tools					
*	 Associates shaping processes with the types of materials used (e.g. injection blow moulding is used to shape plastics) 			*		
*	 Determines the appropriate shaping techniques based on the direct observation of technical objects (e.g. some table legs are turned on a lathe) 			*		
d.	Manufacturing					
	i. Roughing					
+	 Defines roughing as one of the first steps in the manufacturing process 			*		
	ii. Characteristics of laying out					
*	 Associates laying out with saving materials, shaping techniques and the types of materials used 			*		
	iii. Characteristics of drilling, tapping, threading and bending					
*	 Describes the characteristics of the tools needed to shape a material (e.g. the tip of a metal drill is conical, while that of a wood drill is double fluted) 				*	
e.	Measurement and inspection					
	i. Direct measurement					
*	 Explains the purpose of direct measurement (using a ruler) to control the machining of a part 			*		
*	 Explains the choice of the direct measurement instrument used (a vernier calliper is more precise than a ruler) 				*	
	ii. Control, shape and position (plane, section, angle)					
*	 Associates quality control techniques (indirect measurement) for materials and technical objects with the desired degree of precision (e.g. the shape of a musical instrument is validated using a three- dimensional digitizer to ensure the proper sound) 				*	
F. Biot	echnology	1	2	3	4	4
Elementa Students d	ry school o not address any concepts associated with biotechnology.					
Secondar	y school					
a.	Processes					
	i. Pasteurization					
	Describes the pasteurization process			*		
	 Describes the purpose of pasteurization (preservation of food and its nutritional properties) 			*		
	ii. Manufacture of vaccines					
	Describes the process for manufacturing vaccines			*		
	iii. Assisted reproduction					
	Describes different assisted-reproduction processes			*		

*	 Describes the purpose of artificial insemination (animal reproduction, an answer to human infertility, preservation of the gene pool, food self-sufficiency) 		*	
	iv. Cell cultures			
	 Names parameters to be controlled in the case of cultured cells (sources of mother cells, growth, preservation, characteristics of cell media, ethical standards) 		*	

- 1. See Techniques, Technology, Graphic communication (Techniques Technology, 1).
- 2. The progression of learning associated with these concepts is characterized by the increasing complexity of the objects to be represented.
- 3. See Techniques, Technology, Graphic communication, Using scales (Techniques Technology, 1, d).
- 4. This section is continued in *The Material World*, Cycle Two (MW, G).
- 5. For Secondary Cycle Two concepts related to *Energy transformations*, see *The Material World, Changes*, *Transformation of energy* (MW, B, 4).
- 6. The progression of learning associated with this concept is characterized by the increasing complexity of the objects to be studied.

Applied General Education Path

Techniques

The techniques listed below are divided into three categories, depending on whether they apply to science or technology or both. Many of them require the use of instruments and tools or chemicals. Safety in the workshop and laboratory should be a constant concern.

be a constant cond	cern.					
→ Student con	nstructs knowledge with teacher guidance.					
★ Student app	olies knowledge by the end of the school year.		Se	econ	dary	,
Student rein	nvests knowledge.		_			
Technology program	ed by the symbol • indicate knowledge specific to the compulsory Applied Science and n. Most of these statements are, however, found in the progression of learning for the ntal Science and Technology program.	Су	cle ne	Cy Tv	cle	SE Cycle Two
A. Technolog	gy	1	2	3	4	4
simple drawings of machines (e.g. levintroduced to the (e.g. screws, glue	ne symbols associated with motion and electrical and mechanical parts. They in containing symbols. By designing technical objects, they become familiar with the ver, inclined plane, pulley, wheel). They trace parts and cut them out of different safe use of tools (e.g. pliers, screwdriver, hammer, wrench, template) and different e, nails, round-head fasteners, nuts). They pay attention to finishing.	he us	se of ateria	simp ls. T	ole hey a	are
Secondary scho	ool					
1. Graphic con	nmunication ¹	S	Т	AS	ST	SE
a. Doing	a technical drawing					
i. (Chooses the best view for an elevation drawing of a technical object	\rightarrow	*			
ii. F	Represents the visible edges using solid lines	\rightarrow	*			
iii. F	Represents the hidden edges using dotted lines	\rightarrow	*			
iv. I	Indicates the overall external dimensions of an object on a drawing	\rightarrow	*			
b. Readii	ng plans					
i. <i>i</i>	Associates views with the sides of a technical object	\rightarrow	*			
ii. <i>I</i>	Associates lines with the edges of a technical object	\rightarrow	*			
c. Drawir	ng diagrams ²					
i. (Chooses the best view to describe a technical object	\rightarrow	\rightarrow	\rightarrow	*	
ii. U	Uses different colours for each part of a technical object	\rightarrow	*			
	Indicates all the information needed to explain the operation or construction of an object	\rightarrow	\rightarrow	\rightarrow	*	
d. Using	scales ³					
i. /	Associates real measurements with each of the dimensions in a drawing	\rightarrow	*			
ii. F	Reduces or multiplies the dimensions of a technical object based on the scale	\rightarrow	*			
	Dimensions multiview orthogonal projections in accordance with the main dimensioning rules			*		

e.	Using drawing instruments					
	i. Uses drawing instruments (e.g. ruler, square) to make diagrams	\rightarrow	*			
f.	Constructing a graph using instruments					
	 Uses instruments to construct a graph (e.g. multiview orthogonal projection, isometric representation, perspective drawing) 			\rightarrow	*	
g.	Using vector graphic software					
*	 Uses vector graphic software to draw different diagrams in two and three dimensions (e.g. drawing toolbar in Word) 			\rightarrow	*	
2. Manı	ıfacturing ⁴	S	Т	A	ST	SE
a.	Safely using machines and tools ⁵					
	i. Uses tools sadelys (edgring trac table-20 titizys knidel, yleannneyr, is one wolniver options) and e	mic	*			
	ii. Uses machine tools safely (band saw, drill, sander)			\rightarrow	*	
b.	Measuring and laying out					
	i. Identifies the unit of measurement on the instrument	\rightarrow	*			
	ii. Positions the measuring instrument to obtain reliable reference points	\rightarrow	*			
	iii. Adopts the appropriate position for reading an instrument	\rightarrow	*			
	iv. Marks the materials to be shaped using a pencil or punch	\rightarrow	*			
C.	Machining and forming					
	i. Chooses the appropriate materials, tools, techniques and processes	\rightarrow	*			
	ii. Draws the necessary reference lines	\rightarrow	*			
	iii. Immobilizes the part to be formed	\rightarrow	*			
	iv. Forms the part in accordance with the steps in the following machining processes: sawing, drilling, sanding, filing	\rightarrow	*			
	v. Forms the part in accordance with the steps in the following machining processes: stripping, splicing, soldering			\rightarrow	*	
d.	Finishing					
	i. Sands the sides or deburrs the edges of each part after forming	\rightarrow	*			
	ii. Uses the appropriate finish (stain, paint)	\rightarrow	*			
	iii. Grinds, polishes, hammers or chisels metal parts			\rightarrow	*	
e.	Assembling					
	i. Marks the references (holes, points or guidelines)	\rightarrow	*			
	ii. Immobilizes parts during gluing	\rightarrow	*			
	iii. Drills to the diameter of the screws, nails or rivets used	\rightarrow	*			
	iv. Countersinks the openings for countersunk screws	\rightarrow	*			
f.	Assembling and disassembling					
	i. Identifies and gathers the parts and hardware	\rightarrow	*			
	ii. Chooses the appropriate tools	\rightarrow	*			

	iii.	For disassembly, numbers and records the location of the parts	\rightarrow	*			
	iv.	In the case of electrical circuits, identifies and gathers the electrical components			\rightarrow	*	
+	V.	In the case of electronic circuits, identifies and gathers the electronic components			\rightarrow	*	
	vi.	Chooses and places the electrical components in sequence based on the circuit diagram			\rightarrow	*	
+	vii.	Chooses and places the electronic components in sequence based on the circuit diagram			\rightarrow	*	
	viii.	Connects the components using wire, connectors or solders			\rightarrow	*	
•	ix.	Connects the components on a printed circuit board			\rightarrow	*	
+	X.	Uses a desoldering bulb to remove a solder			\rightarrow	*	
g.	Perfo	orming verification and control tasks					
	i.	Evaluates the dimensions of a part during and after construction using a ruler			\rightarrow	*	
	ii.	Compares the real dimensions of a part with the specifications (e.g. draft, drawing, technical sheet)			\rightarrow	*	
	iii.	Uses a template to verify the conformity of a part			\rightarrow	*	
	iv.	Evaluates the dimensions of a part during and after construction using vernier callipers				*	
	Maki	ing a part					
h.							
h.		Makes a part using the appropriate techniques			\rightarrow	*	
B. Sci	i. ence		1	2	→	*	4
B. Scienenta	i. ence ary sc become ts (rule	hool ne familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduated cylinder, balance, thermometer, chronometer).			3	4	
B. Scienenta Elementa Etudents b Instrument	i. ence ary sc becom ts (rule	hool ne familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduated cylinder, balance, thermometer, chronometer).			3	4	
B. Scienenta Elementa Etudents b Instrument	i. ence ary sc becom ts (rule ry sch	hool ne familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduated cylinder, balance, thermometer, chronometer).			3	4	
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B. Sciellementa students to astrumenta a. b.	i. ence ary sc becom ts (rule ry sch Safe i. Sepa i. iii. Desi i. Usin i.	chool The familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduated cylinder, balance, thermometer, chronometer). The service of the familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduated cylinder, balance, thermometer, chronometer). The service of the familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduater). The service of the familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduater). The service of the use of observational instruments (magnifying glass, binocular er, eyedropper, eyedropper, chronometer). The service of the use of observational instruments (magnifying glass, binocular er, eyedropper, chronometer). The service of the use of observational instruments (magnifying glass, binocular er, eyedropper, chronometer). The service of the use of observational instruments (magnifying glass, binocular er, eyedropper, chronometer). The service of the use of observational instruments (magnifying glass, binocular er, eyedropper, chronometer). The service of the use of observational instruments (magnifying glass, binocular er, eyedropper, chronometer). The service of the use of observational instruments (magnifying glass, binocular er, eyedropper, chronometer). The service of the use of observational instruments (magnifying glass, binocular er, eyedropper, chronometer). The service of the use of observation er, eyedropper, chronometer). The service of the use of observation er, eyedropper, chronometer). The service of the use of observation er, eyedropper, chronometer). The service of the use of observation er, eyedropper, chronometer). The service of the use of observation er, eyedropper, chronometer). The service of the use of observation er, eyedropper, chronometer).	$\begin{array}{c} \Rightarrow \\ \Rightarrow \\ \Rightarrow \\ \end{array}$	→ → ★ ★ ★	3	4	
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	iv. Measures the volume of an insoluble solid using water displacement	\rightarrow	*			
	v. Measures temperature using a graduated thermometer	\rightarrow	*			
	vi. Uses measuring instruments appropriately (e.g. ammeter, volumetric flask)			\rightarrow	*	
	vii. Uses measuring instruments appropriately (e.g. ammeter, volumetric flask)					*
e. l	Using observational instruments					
	Uses observational instruments appropriately (e.g. magnifying glass, stereomicroscope, microscope)	\rightarrow	*			
f. F	Preparing solutions					
	i. Prepares an aqueous solution of a specific concentration given a solid solute			*		
	ii. Prepares an aqueous solution of a specific concentration given a concentrated aqueous solution			*		
g. C	Collecting samples					
	i Collecte complex approprietaly (a.g. starilizes the container uses a contrile					
	 i. Collects samples appropriately (e.g. sterilizes the container, uses a spatula, refrigerates the sample) 			\rightarrow	*	
Tech		1	2	3	4	4
	refrigerates the sample)	1	2			4
	refrigerates the sample) nniques common to Science and Technology	1	2			4
	refrigerates the sample) nniques common to Science and Technology Verifying the repeatability, accuracy and sensitivity of measuring instruments i. Takes the same measurement several times to check the repeatability of the	1	2			
	refrigerates the sample) Inniques common to Science and Technology Verifying the repeatability, accuracy and sensitivity of measuring instruments i. Takes the same measurement several times to check the repeatability of the instrument used ii. Carries out the required operations to ensure the accuracy of a measuring instrument (e.g. cleans and calibrates a balance, dries out a graduated	1	2			*
a. V	refrigerates the sample) Iniques common to Science and Technology Verifying the repeatability, accuracy and sensitivity of measuring instruments i. Takes the same measurement several times to check the repeatability of the instrument used ii. Carries out the required operations to ensure the accuracy of a measuring instrument (e.g. cleans and calibrates a balance, dries out a graduated cylinder, rinses and calibrates a pH-meter) iii. Chooses a measuring instrument by taking into account the sensitivity of the instrument (e.g. uses a 25-mL graduated cylinder rather than a 100-mL one to	1	2			*
a. V	refrigerates the sample) Iniques common to Science and Technology Verifying the repeatability, accuracy and sensitivity of measuring instruments i. Takes the same measurement several times to check the repeatability of the instrument used ii. Carries out the required operations to ensure the accuracy of a measuring instrument (e.g. cleans and calibrates a balance, dries out a graduated cylinder, rinses and calibrates a pH-meter) iii. Chooses a measuring instrument by taking into account the sensitivity of the instrument (e.g. uses a 25-mL graduated cylinder rather than a 100-mL one to measure out 18 mL of water	1	2			*
a. V	refrigerates the sample) Iniques common to Science and Technology Verifying the repeatability, accuracy and sensitivity of measuring instruments i. Takes the same measurement several times to check the repeatability of the instrument used ii. Carries out the required operations to ensure the accuracy of a measuring instrument (e.g. cleans and calibrates a balance, dries out a graduated cylinder, rinses and calibrates a pH-meter) iii. Chooses a measuring instrument by taking into account the sensitivity of the instrument (e.g. uses a 25-mL graduated cylinder rather than a 100-mL one to measure out 18 mL of water Interpreting the results of measurement i. Determines the error attributable to a measuring instrument (e.g. the error in a measurement made using a graduated cylinder is provided by the	1	2			*
a. V	refrigerates the sample) Iniques common to Science and Technology Verifying the repeatability, accuracy and sensitivity of measuring instruments i. Takes the same measurement several times to check the repeatability of the instrument used ii. Carries out the required operations to ensure the accuracy of a measuring instrument (e.g. cleans and calibrates a balance, dries out a graduated cylinder, rinses and calibrates a pH-meter) iii. Chooses a measuring instrument by taking into account the sensitivity of the instrument (e.g. uses a 25-mL graduated cylinder rather than a 100-mL one to measure out 18 mL of water Interpreting the results of measurement i. Determines the error attributable to a measuring instrument (e.g. the error in a measurement made using a graduated cylinder is provided by the manufacturer or corresponds to half of the smallest division on the scale) ii. Estimates the errors associated with the user and the environment when	1	2			*

- 1. See The Technological World, Graphical language (TW, A).
- 2. The progression of learning associated with these techniques is characterized by the increasing complexity of the objects to be represented.
- 3. See The Technological World, Graphical language, Scales (TW, A, g).
- 4. See The Technological World, Manufacturing, Shaping, Manifacturing, Measurement and inspection (TW, E, c-d-e).
- 5. When the teacher introduces a new technique, he or she should explain the related safety rules and repeat them often. After several practice sessions, students should apply the rules without being reminded.
- 6. When the teacher introduces a new technique, he or she should explain the related safety rules and repeat them often. After several practice sessions, students should apply the rules without being reminded.

Applied General Education Path

Strategies

The strategies listed below are fundamental to the approaches used in science and technology. They can be applied in a variety of increasingly complex contexts and are therefore inclusive. Thus, students build on the strategies they learned in elementary school. New strategies are added, including analytical strategies, which are adapted to students' level of cognitive development.

Student applies knowledge by the end of the school year. Student reinvests knowledge. E: The letter "E" indicates that students were introduced to this strategy in elementary school. A. Exploration strategies 1 2 1. Studying a problem or a phenomenon from different points of view (e.g. social, environmental, historical, economic)	AST SE Cycle Two Two
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Studying a problem or a phenomenon from different points of view (e.g. social, environmental, historical, economic)	
environmental, historical, economic)	3 4 4
Distinguishing between the different types of information useful for solving the problem	
3. Referring to similar problems that have already been solved	
4. Becoming aware of his or her previous representations	
5. Drawing a diagram for the problem or illustrating it	
6. Formulating questions	
7. Putting forward hypotheses (e.g. individually, in teams, as a class)	
Exploring various ways of solving the problem	
Anticipating the results of his or her approach	
10. Imagining solutions to a problem in light of his or her explanations	
11. Taking into account the constraints involved in solving a problem or making an object (e.g. specifications, available resources, time allotted)	
12. Examining his or her mistakes in order to identify their source	
13. Using different types of reasoning (e.g. induction, deduction, inference, comparison, classification)	
14. Using empirical approaches (e.g. trial and error, analysis, exploration using one's senses)	
15. Ensuring that the procedure is appropriate and safe and making the necessary adjustments	
Collecting as much scientific, technological and contextual information as possible to define a problem or predict patterns	→ *
17. Generalizing on the basis of several structurally similar cases	→ ★
18. Developing various scenarios	→ *
19. Considering various points of view on scientific or technological issues	→ *

B. Instrumentation strategies		1	2	3	4	4
 Using different sources of information (e.g. books, newspapers, Web sites, magazines, experts) 	Е					
2. Validating sources of information	Е					
 Using technical design to illustrate a solution (e.g. diagrams, sketches, technical drawings) 	Е					
 Using different tools for recording information (e.g. diagrams, notes, graphs, procedures, logbook) 	Е					
5. Using a variety of observational techniques and tools	Е					
6. Selecting suitable techniques or tools for observation		\rightarrow	*			
C. Analytical strategies		1	2	3	4	4
 Identifying the constraints and important elements related to the problem-solving situation 		\rightarrow	*			
2. Dividing a complex problem into simpler subproblems		\rightarrow	*			
 Using different types of reasoning (e.g. inductive and deductive reasoning, comparison, classification, prioritization) in order to process information 		\rightarrow	*			
Reasoning by analogy in order to process information and adapt scientific and technological knowledge				\rightarrow	*	
 Selecting relevant criteria to help him or her determine where he or she stands on a scientific or technological issue 				\rightarrow	*	
D. Communication strategies		1	2	3	4	4
 Using different means of communication to propose explanations or solutions (e.g. oral presentation, written presentation, procedure) 	Е					
2. Organizing information for a presentation (e.g. tables, diagrams, graphs)	Е					
Exchanging information	Е					
 Comparing different possible explanations for or solutions to a problem in order to assess their relevance (e.g. full-group discussion) 	Е					
Using tools to display information in various formats (e.g. data tables, graphs, diagrams)		\rightarrow	*			