

Québec Education Program Progression of Learning

Applied Science and Technology Secondary III 2021-2022 School Year

Learning to Be Prioritized for the 2021-2022 School Year in the Context of the Pandemic

This document is identical to the one produced for the 2020-2021 school year.





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Progression of Learning in Secondary School

The progression of learning in secondary school constitutes a complement to each school subject, providing further information on the knowledge that the students must acquire and be able to use in each year of secondary school. This tool is intended to assist teachers in planning both their teaching and the learning that their students are to acquire.

The role of knowledge in learning

The knowledge that young people acquire enables them to better understand the world in which they live. From a very early age, within their families and through contact with the media and with friends, they accumulate and learn to use an increasingly greater body of knowledge. The role of the school should be to progressively broaden, deepen and structure this knowledge.

Knowledge and competencies must mutually reinforce each other. On the one hand, knowledge becomes consolidated when it is used and, on the other hand, the exercise of competencies entails the acquisition of new knowledge. Helping young people acquire knowledge raises the challenging question of how to make this knowledge useful and durable, and thus evokes the notion of competency. For example, we can never be really assured that a grammar rule has been assimilated until it is used appropriately in a variety of texts and contexts that go beyond the confines of a repetitive, targeted exercise.

Intervention by the teacher

The role of the teacher in knowledge acquisition and competency development is essential, and he or she must intervene throughout the learning process. In effect, the *Education Act* confers on the teacher the right to "select methods of instruction corresponding to the requirements and objectives fixed for each group or for each student entrusted to his care." It is therefore the teacher's responsibility to adapt his or her instruction and to base it on a variety of strategies, whether this involves lecture-based teaching for the entire class, individualized instruction for a student or a small group of students, a series of exercises to be done, a team activity or a particular project to be carried out.

In order to meet the needs of students with learning difficulties, teachers should encourage their participation in the activities designed for the whole class, although support measures should also be provided, when necessary. These might involve more targeted teaching of certain key elements of knowledge, or they might take the form of other specialized interventions.

As for the evaluation of learning, it serves two essential functions. Firstly, it enables us to look at the students' learning in order to guide and support them effectively. Secondly, it enables us to verify the extent to which the students have acquired the expected learning. Whatever its function, in accordance with the *Policy on the Evaluation of Learning*, evaluation should focus on the acquisition of knowledge and the students' ability to use this knowledge effectively in contexts that draw upon their competencies.

Structure

The progression of learning is presented in the form of tables that organize the elements of knowledge similarly to the way they are organized in the subject-specific programs. In mathematics, for example, learning is presented in fields: arithmetic, geometry, etc. For subjects that continue on from elementary school, the *Progression of Learning in Secondary School* has been harmonized with the *Progression of Learning in Elementary School*. Every element of learning indicated is associated with one or more years of secondary school during which it is formally taught.

A uniform legend is used for all subjects. The legend employs three symbols: an arrow, a star and a shaded box. What is expected of the student is described as follows:



An **arrow** indicates that teaching must be planned in a way that enables students to begin acquiring knowledge during the school year and continue or conclude this process in the following year, with ongoing systematic intervention from the teacher.

A **star** indicates that the teacher must plan for the majority of students to have acquired this knowledge by the end of the school year.

A **shaded box** indicates that the teacher must plan to ensure that this knowledge will be applied during the school year.

Applied General Education Path

Introduction

This document provides additional information about the learning prescribed in the compulsory secondary-level Science and Technology programs and its progression from year to year and from cycle to cycle. This document is intended to help teachers with their lesson planning.

To progress in their learning, students need to do more than merely acquire knowledge. They must also learn to apply their knowledge in a variety of increasingly complex situations. By appropriately using the knowledge, techniques and strategies listed in this document, they will develop the competencies outlined in the Science and Technology programs. By applying these competencies, they will acquire new knowledge which, in turn, will help them further develop their competencies.

In order to seek answers or solutions to scientific and technological problems (Competency 1), students must become familiar with strategies and acquire conceptual and technical knowledge that will enable them to define a problem, explore it and then justify their methodological choices and results. Similarly, the appropriate scientific or technological concepts and principles can help them understand phenomena, explain the operation of objects or form an opinion and, consequently, make the most of their scientific and technological knowledge (Competency 2). Finally, in order to communicate in the languages used in science and technology (Competency 3), they must have knowledge that will enable them to interpret and convey messages using the languages and types of representation associated with science and technology.

In elementary school, students became familiar with science and technology and explored knowledge involving simple and usually observable phenomena in their immediate environment. In secondary school, they further develop their scientific and technological literacy and continue to do so throughout their lives. In Cycle One, students learn about natural phenomena and man-made objects that interest them. In Cycle Two, the concepts are organized around applications related to seven technological fields: medical, agricultural and agri-food, energy, information and communications, transportation, manufacturing and construction technologies. In the optional Science and the Environment program, the knowledge to be acquired is organized around two environmental issues about which the students will be required to form their own opinions, thus developing a new aspect of their subject-specific competencies. Successful completion of this program will make it easier to enroll in the optional Physics and Chemistry programs offered in Secondary V.

The tables in this document outline the knowledge related to each of the four areas of the programs: The Material World, The Living World, The Earth and Space, and The Technological World. Each table is preceded by a text explaining how this knowledge contributes to students' learning in science and technology. Each section begins with a short text describing the related knowledge that was acquired at the elementary level. Two other tables provide information about the appropriate techniques and strategies for students to use.

The concepts are further clarified by a list of statements indicating the degree of complexity of the subject matter targeted and explanations of the progression of learning from one year to the next. In some cases, specifications about the extent of the knowledge to be addressed appear in parentheses.

Elementary school teachers can choose themes from among those listed in the program. It is therefore possible that
some students may not have studied certain concepts, even though the concepts mentioned here should have been
addressed at the elementary level. The table of <u>strategies</u> includes a column devoted to learning acquired in elementary
school.

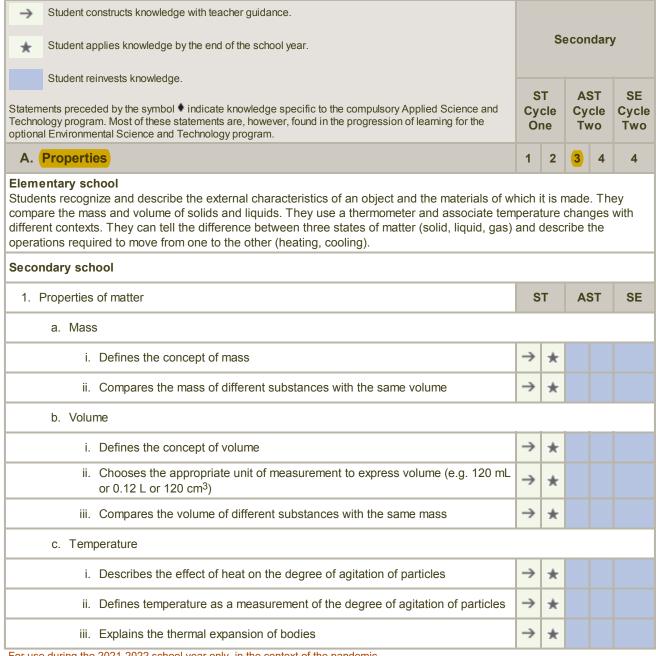
Applied General Education Path

The Material World

In The Material World, students acquire scientific and technological knowledge pertaining to the organization of the world around us, the elements that compose it and the forces that govern it.

In secondary school, students explore increasingly complex phenomena and technical objects and seek answers and solutions to a variety of problems. They acquire scientific knowledge about The Material World that helps them understand and explain the factors at play in different scientific issues and in the operation of technological objects, systems and processes. This knowledge, along with the knowledge they acquire in other areas of the program, enables them to understand scientific models, theories and laws. Students refine their understanding of the concepts related to The Material World by using the experimental method, technological analysis and the technological design process.

In Secondary III, students explore applications related to the seven technological fields, which enables them to make connections between human beings and The Material World, and provides them with an opportunity to integrate knowledge related to The Living World. In Secondary IV, they continue to construct and apply their knowledge about The Material World by analyzing and designing a variety of applications related to the same technological fields. Thus they acquire a better understanding of the omnipresence of science and technology in the world around us. In the optional Science and the Environment program, students consolidate their knowledge and form their own opinions regarding two environmental issues they will be asked to examine.



	i	Names the different phase changes of matter (vaporization, condensation,					
		freezing, melting, deposition, sublimation)	\rightarrow	*			
	ii.	Interprets the phase change diagram for a pure substance	\rightarrow	*			
e.	Acidit	y/alkalinity					
	i.	Determines the observable properties of acidic, basic or neutral solutions (e.g. reaction to litmus, reactivity with metals)	\rightarrow	*			
	ii.	Determines the acidity or alkalinity of common substances (e.g. water, lemon juice, vinegar, soft drinks, milk of magnesia, cleaners)	\rightarrow	*			
f.	Char	acteristic properties					
	i.	Defines a characteristic property as a property that aids in the identification of a substance or group of substances	\rightarrow	*			
	ii.	Identifies groups of substances based on their common characteristic properties (e.g. acids turn litmus red)	\rightarrow	*			
	iii.	Associates a characteristic property of a substance or material with its use (e.g. metal is used to make pots because it is a good conductor of heat)	\rightarrow	*			
2. Char	acteris	stic physical properties	S	Т	AS	ST	SE
a.	Meltin	ng point)					
	i.	Identifies a substance by its melting point using a reference document			*		
b.	Boilir	g point					
	i.	Identifies a substance by its boiling point using a reference document			*		
C.	Dens	ity)					
	i.	Explains the concept of density			*		
	ii.	Determines the density of different substances			*		
		Identifies liquid and solid substances by their density using a reference document)			*		
d.	Solub	pility					
	i.	Defines the concept of solubility					*
							-
	ii. 	Describes the effect of variations in temperature on the solubility of a substance					*
3. Prop		· · · · · · · · · · · · · · · · · · ·	S	т	AS	ЭТ	SE
		substance of solutions	S	т	AS	ВТ	
	erties Solut	substance of solutions	S	īΤ	AS	Т	
a.	Solut	substance of solutions ions	S	т		Т	
a.	Solut i.	substance of solutions ions Recognizes the solute and the solvent in a homogeneous mixture Describes the effect of variations in the quantity of solute or solvent on a	S	T .	*	т	
a.	Solut i. ii.	substance of solutions ions Recognizes the solute and the solvent in a homogeneous mixture Describes the effect of variations in the quantity of solute or solvent on a solution's concentration	S	ST .	*	ВТ	
a.	Solut i. ii. iii.	substance of solutions ions Recognizes the solute and the solvent in a homogeneous mixture Describes the effect of variations in the quantity of solute or solvent on a solution's concentration Determines the concentration of an aqueous solution (g/L or percentage) Determines the concentration of an aqueous solution (g/L, percentage, ppm,	S	т	*	ST	SE
a.	Solut i. ii. iii. iv.	substance of solutions ions Recognizes the solute and the solvent in a homogeneous mixture Describes the effect of variations in the quantity of solute or solvent on a solution's concentration Determines the concentration of an aqueous solution (g/L or percentage) Determines the concentration of an aqueous solution (g/L, percentage, ppm, mol/L)	S	T	*	ST	SE

c. pH scale					
 i. Describes the pH scale (acidity, alkalinity, neutrality, increasing and decreasing values) 					*
ii. Determines the pH of a few common substances (e.g. distilled water, rainwater, saliva, lemon juice, cleaners)					*
d. lons					
i. Defines the concept of ion					*
e. Electrical conductivity					
 Describes the mechanism that allows aqueous solutions to conduct electricity (electrolytic dissolution of a solute, formation of mobile ions) 					*
4. Characteristic chemical properties	S	Т	AS	ST	SE
a. Reaction to indicators					
 i. Recognizes a substance by its characteristic chemical properties (e.g. starch turns blue in the presence of an iodine solution, acidic solutions turn bromothymol blue yellow) 			*		
B. Changes	1	2	3	4	
Elementary school Students demonstrate that the properties of matter do not change during physical changes (e.g. crushing), but that they do change during chemical changes (e.g. cooking, combustion). They re of matter is conserved during physical changes (e.g. mass of a piece of chalk whether whole or familiar with how certain household products are made (e.g. soap, paper, maple syrup).	cogr	ize t	hat tl	he qu	uantity
Secondary school					
•					
1. Changes in matter	S	Т	AS	ST	SE
Changes in matter a. Conservation of matter	S	Т	AS	ST	SE
	s →	T ★	AS	ST	SE
a. Conservation of matter i. Demonstrates that matter is conserved during a chemical change			AS	ST	SE
a. Conservation of matter i. Demonstrates that matter is conserved during a chemical change (e.g. conservation of mass in a precipitation reaction)			AS	ST	SE
a. Conservation of matter i. Demonstrates that matter is conserved during a chemical change (e.g. conservation of mass in a precipitation reaction) b. Mixtures i. Describes the properties of a mixture (e.g. made up of several substances,	→	*	AS	ST	SE
a. Conservation of matter i. Demonstrates that matter is conserved during a chemical change (e.g. conservation of mass in a precipitation reaction) b. Mixtures i. Describes the properties of a mixture (e.g. made up of several substances, one or more phases) ii. Distinguishes between a solution or homogenous mixture (e.g. drinking water,	→	*	AS	ST	SE
a. Conservation of matter i. Demonstrates that matter is conserved during a chemical change (e.g. conservation of mass in a precipitation reaction) b. Mixtures i. Describes the properties of a mixture (e.g. made up of several substances, one or more phases) ii. Distinguishes between a solution or homogenous mixture (e.g. drinking water, air, alloy) and a heterogeneous mixture (e.g. tomato juice, smog, rock)	→	*	AS	ST	SE
a. Conservation of matter i. Demonstrates that matter is conserved during a chemical change (e.g. conservation of mass in a precipitation reaction) b. Mixtures i. Describes the properties of a mixture (e.g. made up of several substances, one or more phases) ii. Distinguishes between a solution or homogenous mixture (e.g. drinking water, air, alloy) and a heterogeneous mixture (e.g. tomato juice, smog, rock) c. Solutions i. Describes the properties of an aqueous solution (e.g. only one visible phase,	→	* *	AS	ST	SE
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a. Conservation of matter i. Demonstrates that matter is conserved during a chemical change (e.g. conservation of mass in a precipitation reaction) b. Mixtures i. Describes the properties of a mixture (e.g. made up of several substances, one or more phases) ii. Distinguishes between a solution or homogenous mixture (e.g. drinking water, air, alloy) and a heterogeneous mixture (e.g. tomato juice, smog, rock) c. Solutions i. Describes the properties of an aqueous solution (e.g. only one visible phase, translucent) d. Separation of mixtures i. Associates a separation technique with the type of mixture to be separated ii. Describes the steps involved in separating a complex mixture (e.g. sedimentation, decantation and evaporation to separate salt water and	 → → → 	* * *	AS	ST	SE
a. Conservation of matter i. Demonstrates that matter is conserved during a chemical change (e.g. conservation of mass in a precipitation reaction) b. Mixtures i. Describes the properties of a mixture (e.g. made up of several substances, one or more phases) ii. Distinguishes between a solution or homogenous mixture (e.g. drinking water, air, alloy) and a heterogeneous mixture (e.g. tomato juice, smog, rock) c. Solutions i. Describes the properties of an aqueous solution (e.g. only one visible phase, translucent) d. Separation of mixtures i. Associates a separation technique with the type of mixture to be separated 1 ii. Describes the steps involved in separating a complex mixture (e.g. sedimentation, decantation and evaporation to separate salt water and sand)	 → → → 	* * *	*	ST	SE

2. Physical changes	S	ST .	AS	ST	SE
a. (Physical changes)					
 Describes the characteristics of a physical change (e.g. substance retains its properties, molecules remain intact) 	\rightarrow	*			
 Recognizes different physical changes (e.g. phase changes, preparation or separation of a mixture) 	\rightarrow	*			
Describes a few physical changes (e.g. dissolution, dilution, phase changes)			*		
iv. Illustrates physical changes using the particle model			*		
b. Dissolution					
i. Explains dissolution using the particle model					*
c. Dilution					
i. Explains dilution in terms of concentration and volume					*
 Determines the final volume or concentration of an aqueous solution after dilution (e.g. the concentration of a solution decreases by half when the volume of solvent is doubled) 					*
3. Chemical changes	5	ST.	AS	ST	SE
a. Chemical changes					
 Describes the indicators of a chemical change (formation of a precipitate, effervescence, colour change, heat, light) 	\rightarrow	*			
 Explains a chemical change based on the changes in the properties of the substances involved 	\rightarrow	*			
iii. Names different types of chemical changes (e.g. decomposition, oxidation)	\rightarrow	*			
iv. Names chemical changes that occur in the human body (e.g. respiration, digestion)			*		
b. Decomposition and synthesis					
 i. Associates known chemical reactions with decomposition or synthesis reactions (e.g. respiration, photosynthesis, combustion, digestion) 					*
c. Oxidation					
i. Represents an oxidation reaction using the particle model				*	
Associates known chemical reactions with oxidation reactions (e.g. combustion, corrosion)				*	
iii. Associates a chemical equation in which oxygen is one of the reactants with one of the possible cases of an oxidation reaction					*
d. Precipitation					
 Describes the visible manifestation of precipitation (formation of a solid deposit after two aqueous solutions are mixed) 					*
ii. Represents a precipitation reaction using the particle model					*
e. Combustion					
i. Describes the perceivable manifestations of rapid combustion (e.g. heat, light)			*	
ii. Explains a combustion reaction using the fire triangle				*	
f. Photosynthesis and respiration ²					

g. Acid-base neutralization reaction					
 Gives examples of acid-base neutralization reactions (e.g. adding lime to neutralize the acidity of a lake) 			Ш		
ii. Names the products formed during acid-base neutralization (salt and water)					
iii. Recognizes an acid-base neutralization from its equation					
h. Salts					
 Determines the molecular formula of the salt produced by the neutralization of a given acid and a given base 					
i. Types of bonds					
i. Covalent					
 Defines a covalent bond as a bond resulting from a sharing of electrons 					
Makes a schematic representation of a covalent bond					
 Identifies molecules that feature a covalent bond (e.g. N₂, CO₂) 					
ii. Ionic					
 Defines an ionic bond as a bond resulting from the gain or loss of electrons 					
 Makes a schematic representation of an ionic bond 					
 Identifies molecules that present an ionic bond (e.g. NaCl, NH4OH) 					
 Associates an ionic bond with an electrolytic substance 					
j. Law of conservation of mass					
i. Explains the law of conservation of mass during a chemical reaction					
ii. Represents the conservation of mass using the particle model					
k. Balancing chemical equations					
i. Balances chemical equations					
I. Stoichiometry					
 Determines the quantities of reactants or products using stoichiometric calculations (gram or mole) 					
m. Endothermic and exothermic reactions					
 Distinguishes an endothermic reaction from an exothermic reaction according to perceptible signs (e.g. temperature variations, emission of light) 					
 Distinguishes an endothermic reaction from an exothermic reaction according to the position of the energy term in the chemical equation 					
Transformation of energy ³	5	ST .	AS	ST	ļ
a. Forms of energy					
i. Describes different forms of energy (chemical, thermal, mechanical, radiation)			*		
ii. (Identifies the forms of energy involved in a transformation (e.g. electrical to					

b.	Law of conservation of energy						
	i. Explain qualitatively the law of conse	rvation of energy				*	
	ii. Applies the law of conservation of en	ergy in different contexts				*	
C.	Energy efficiency						
	Defines the energy efficiency of a de- energy consumed that is transformed energy / amount of energy consumed	into effective work (amount of useful				*	
	ii. Explains how to improve the energy of	efficiency of an electrical appliance				*	
d.	Distinction between heat and temperature ⁴						
	i. Describes heat as a manifestation of	energy				*	
	ii. Describes the relationship between h	eat and temperature				*	
e.	Relationship between thermal energy, spec	cific heat capacity and temperature variation	on ⁵				
		p between the change in thermal energy mass, its specific heat capacity and the sexposed					*
	 ii. Applies the mathematical relationship heat capacity and temperature variati 	between thermal energy, mass, specific on ($\Delta E = Q = mc\Delta T$)					*
f.	Relationship between potential energy, ma	ss, acceleration and distance travelled					
	 Describes qualitatively the relationshi body, its mass, its gravitational acceleration 						*
	ii. Applies the mathematical relationship gravitational acceleration and the dis						*
g.	Relationship between kinetic energy, mass	and speed					
	 Describes qualitatively the relationshi its mass and its speed 	p between the kinetic energy of a body,					*
	ii. Applies the mathematical relationship speed ($E_k = \frac{1}{2}mv^2$)	between kinetic energy, mass and					*
h.	Relationship between work and energy ⁶						
	Describes qualitatively the relationshi the energy change within that body	p between the work done on a body and					*
	ii. Applies the mathematical relationship	between work and energy (W = ΔE)					*
C. Org	anization		1	2	3	4	4
characteri	ry school assify objects or substances based on theil tics. They use the common names for certa esis and respiration.				•		dying
Secondar	y school						
a.	Atom						
	i. Describes Dalton's atomic model		\rightarrow	*			
	ii. Defines the atom as the basic unit of	the molecule	\rightarrow	*			
b.	Molecule						
	Describes a molecule using Dalton's linked by chemical bonds)	atomic model (combination of atoms	\rightarrow	*			
	ii. Represents the formation of a molecu	ıle using Dalton's atomic model	\rightarrow	*			

c. Element					
 i. Defines an element as a pure substance made of a single type of atom (e.g. Fe, N₂) 	\rightarrow	*			
d. Periodic table					
i. Describes the periodic table as a structured classification of elements	\rightarrow	*			
e. Pure substance					
 Defines a pure substance as a substance made up of a single type of atom or molecule 			*		
ii. Distinguishes between elements (e.g. iron, dioxygen, sodium) and compounds (e.g. water, carbon dioxide, glucose)			*		
f. Homogeneous and heterogeneous mixtures?					
 i. Describes homogeneous and heterogeneous mixtures in the human body (e.g. lymph, blood, urine) 			*		
g. Elementary particles					
 Describes the position and electrical charge of the elementary particles in an atom (proton, electron, neutron) 					*
h. Simplified atomic model					
i. Represents an atom of a given element using the simplified atomic model					*
i. Lewis notation					
i. Determines the number of valence electrons in an element					*
ii. Represents atoms using Lewis notation					*
j. Nomenclature and notation rules					
 i. Applies nomenclature and notation rules to name the molecule or write the molecular formula of binary compounds 					*
k. Polyatomic ions					
i. Recognizes the common polyatomic ions (e.g. NH ₄ ⁺ , OH ⁻ , NO ₃ ⁻ , CO ₃ ² -, SO ₄ ² -, PO ₄ ³ -) by their name, their formula or their composition					*
I. Concept of the mole					
i. Defines the mole as the unit of measure of the amount of a substance					*
ii. Expresses an amount of a substance in moles					*
m. Relative atomic mass and isotopes					
 Defines isotopes as atoms of the same element whose nuclei have different numbers of neutrons and therefore different atomic masses 					*
ii. Explains qualitatively the concept of relative atomic mass					*
D. Fluids	1	2	3	4	4
Elementary school Students distinguish between three states of matter: solid, liquid and gas.					

a.	Pres	<mark>sure</mark>					
	i.	Defines pressure as the force exerted by particles when they collide with a constricting surface			*		
	ii.	Qualitatively describes the main factors that affect the pressure exerted by a fluid			*		
b.	Com	pressible and incompressible fluids					
	i.	Distinguishes between compressible and incompressible fluids			*		
	ii.	Names compressible fluids (e.g. air) and incompressible fluids (e.g. blood) in the human body			*		
	iii.	Explains how fluids move around in the human body, using the concept of pressure			*		
C.	Rela	cionship between pressure and volume					
	i.	Qualitatively describes the relationship between the pressure and volume of a gas (e.g. inhaling and exhaling, bicycle pump)			*		
d.	Archi	medes' principle					
+	i.	Describes the relationship between the weight of the water displaced by an immersed body and the upward acting force				*	
•	ii.	Explains the buoyancy of a body in terms of Archimedes' principle				*	
e.	Pasc	al's law					
*	i.	Recognizes technical objects or technological systems whose operation is based on Pascal's principle (e.g. hydraulic systems, pneumatic systems)				*	
f.	Bern	oulli's principle					
+	i.	Describes the relationship between the velocity of a fluid and its pressure				*	
•	ii.	Explains the concept of lift in terms of Bernoulli's principle				*	
E. Wa	ves		1	2	3	4	4
menta idents a		nool ate sunlight with a source of energy.					
condar	y sch	ool					
2	Гиол						
a.	Freq	uency					
<u>a.</u>		Defines the frequency of a wave as the number of cycles per second (Hz)			*		
а. 	i.	·			*		
	i. ii.	Defines the frequency of a wave as the number of cycles per second (Hz) Associates the frequency of a sound wave with the pitch of the sound					
	i. ii. Wave	Defines the frequency of a wave as the number of cycles per second (Hz) Associates the frequency of a sound wave with the pitch of the sound (e.g. a low-frequency wave produces a low-pitched sound)					
	i. ii. Wave	Defines the frequency of a wave as the number of cycles per second (Hz) Associates the frequency of a sound wave with the pitch of the sound (e.g. a low-frequency wave produces a low-pitched sound) elength Defines wavelength as the distance between two identical points on a wave			*		
b.	i. ii. Wave	Defines the frequency of a wave as the number of cycles per second (Hz) Associates the frequency of a sound wave with the pitch of the sound (e.g. a low-frequency wave produces a low-pitched sound) elength Defines wavelength as the distance between two identical points on a wave at a given time (e.g. distance between crests) Describes the relationship between wavelength and energy (e.g. high-energy X-rays have a short wavelength)			*		
b.	i. ii. Wave i. ii.	Defines the frequency of a wave as the number of cycles per second (Hz) Associates the frequency of a sound wave with the pitch of the sound (e.g. a low-frequency wave produces a low-pitched sound) elength Defines wavelength as the distance between two identical points on a wave at a given time (e.g. distance between crests) Describes the relationship between wavelength and energy (e.g. high-energy X-rays have a short wavelength)			*		
b.	i. ii. Wave i. ii. Ampl	Defines the frequency of a wave as the number of cycles per second (Hz) Associates the frequency of a sound wave with the pitch of the sound (e.g. a low-frequency wave produces a low-pitched sound) elength Defines wavelength as the distance between two identical points on a wave at a given time (e.g. distance between crests) Describes the relationship between wavelength and energy (e.g. high-energy X-rays have a short wavelength)			*		

e. Electromagnetic spectrum					
 i. Locates different areas on the electromagnetic spectrum (e.g. radio waves, visible light, X-rays) 			*		
 Describes different applications of electromagnetic waves in the health care sector (e.g. X-rays, infrared optical imaging) 			*		
f. Deviation of light waves					
i. Describes how light rays are deviated by a plane reflective surface			*		
ii. Determines the angle of reflection of a light ray on the surface of a plane mirror			*		
 Describes how light rays are deviated when they pass through the surface of a translucent substance 			*		
g. Focal point of a lens					
i. Determines the focal point of concave and convex lenses			*		
 Describes the relationship between the focal point of a lens and the degree of deviation of light rays in different situations (e.g. accommodation of the crystalline lens, choice of corrective lenses) 			*		
F. Electricity and electromagnetism	1	2	3	4	4
Elementary school Students name the components of a simple electrical circuit (wire, power source, bulb, switch) a function. They recognize the effects of magnetism in magnets (attraction and repulsion).	nd de	escri	be th	eir	
Secondary school					
1. Electricity	S	Т	A	ST	SE
Electricity a. Electrical charge	S	Т	AS	ST	SE
	S	Т	AS	ST ★	SE
a. Electrical charge	S	Т	AS		SE
a. Electrical charge i. Associates elementary particles with their electrical charge ii. Describes the behaviour of electrical charges of opposite signs or of the same	S	Т	AS	*	SE
a. Electrical charge i. Associates elementary particles with their electrical charge ii. Describes the behaviour of electrical charges of opposite signs or of the same sign when close together	S	Т	A	*	SE
a. Electrical charge i. Associates elementary particles with their electrical charge ii. Describes the behaviour of electrical charges of opposite signs or of the same sign when close together b. Static electricity i. Describes static electricity as the transfer of electrons from one body to	S	T	AS	*	SE
a. Electrical charge i. Associates elementary particles with their electrical charge ii. Describes the behaviour of electrical charges of opposite signs or of the same sign when close together b. Static electricity i. Describes static electricity as the transfer of electrons from one body to another	S	T	A	*	SE
a. Electrical charge i. Associates elementary particles with their electrical charge ii. Describes the behaviour of electrical charges of opposite signs or of the same sign when close together b. Static electricity i. Describes static electricity as the transfer of electrons from one body to another c. Ohm's law i. Qualitatively describes the relationship between voltage, resistance and	S	Т	A	*	SE
a. Electrical charge i. Associates elementary particles with their electrical charge ii. Describes the behaviour of electrical charges of opposite signs or of the same sign when close together b. Static electricity i. Describes static electricity as the transfer of electrons from one body to another c. Ohm's law i. Qualitatively describes the relationship between voltage, resistance and current intensity in an electrical circuit ii. Applies the mathematical relationship between voltage, resistance and current	S	T	A	* *	SE
 a. Electrical charge Associates elementary particles with their electrical charge Describes the behaviour of electrical charges of opposite signs or of the same sign when close together Static electricity Describes static electricity as the transfer of electrons from one body to another Ohm's law Qualitatively describes the relationship between voltage, resistance and current intensity in an electrical circuit Applies the mathematical relationship between voltage, resistance and current intensity in an electrical circuit (V = RI) 	S		A	* *	SE
a. Electrical charge i. Associates elementary particles with their electrical charge ii. Describes the behaviour of electrical charges of opposite signs or of the same sign when close together b. Static electricity i. Describes static electricity as the transfer of electrons from one body to another c. Ohm's law i. Qualitatively describes the relationship between voltage, resistance and current intensity in an electrical circuit ii. Applies the mathematical relationship between voltage, resistance and current intensity in an electrical circuit (V = RI) d. Electrical circuits i. Describes the function of different elements of an electrical circuit (e.g. the wires transmit electrons along the circuit, resistors transform electrical energy	S		A	* * *	SE
 a. Electrical charge i. Associates elementary particles with their electrical charge ii. Describes the behaviour of electrical charges of opposite signs or of the same sign when close together b. Static electricity i. Describes static electricity as the transfer of electrons from one body to another c. Ohm's law i. Qualitatively describes the relationship between voltage, resistance and current intensity in an electrical circuit ii. Applies the mathematical relationship between voltage, resistance and current intensity in an electrical circuit (V = RI) d. Electrical circuits i. Describes the function of different elements of an electrical circuit (e.g. the wires transmit electrons along the circuit, resistors transform electrical energy into another form of energy)⁸ 	S		A	* * *	SE

	i.	Applies the mathematical relationship between power, voltage and current intensity in an electrical circuit (P = VI)				*	
	ii.	Describes qualitatively the relationship between the power of an electrical appliance, the electrical energy it consumes and the amount of time it is in operation				*	
	iii.	Applies the mathematical relationship between electrical energy consumed, the power of an electrical appliance and the amount of time it is in operation (E = $P\Delta t$)				*	
Elec	troma	gnetism	S	Т	AS	ST	
a.	Mag	netic field of a live wire					
	i.	Describes the magnetic field produced by a current-carrying wire (right-hand rule or left-hand rule)				*	
	ii.	Names ways of modifying the intensity of the magnetic field produced by a current-carrying wire (type of wire, current intensity)				*	
b.	Forc	es of attraction and repulsion					
	i.	Compares the behaviour of a compass in the magnetic field of a magnet with the magnetic field created by a current-carrying wire				*	
C.	Mag	netic field of a solenoid					
*	i.	Describes the magnetic field produced by a solenoid (right-hand rule or left-hand rule)				*	
*	ii.	Names ways of changing the intensity of the magnetic field produced by a solenoid (nature of the core, intensity of the current, number of turns)				*	
d.	Elec	tromagnetic induction					
*	i.	Names ways of inducing electrical current in a wire (e.g. movement of a magnet, changing the intensity of a magnetic field)				*	
			1	2	3	*	
	rce ar	magnet, changing the intensity of a magnetic field)	1	2	3		
Foi	rce ar	magnet, changing the intensity of a magnetic field) nd motion ⁹	1	2	3		
Foi	rce ar	magnet, changing the intensity of a magnetic field) nd motion ⁹	1	2	3		
For onda	rce ar ire Forc	magnet, changing the intensity of a magnetic field) nd motion ⁹ e Describes the effects produced by a force (change in the state of motion of a	1	2	3	4	
For onda	rce arire Force	magnet, changing the intensity of a magnetic field) nd motion ⁹ e Describes the effects produced by a force (change in the state of motion of a body, distortion of a body)	1	2	3	4	
Foi onda a. b.	rce ar ire Forc i. Type i.	magnet, changing the intensity of a magnetic field) nd motion ⁹ e Describes the effects produced by a force (change in the state of motion of a body, distortion of a body) es of forces Recognizes different types of forces in technical objects or technological systems (e.g. gravitational force in a chute, magnetic force exerted by an	1	2	3	*	
Foi onda a. b.	rce ar ire Forc i. Type i.	magnet, changing the intensity of a magnetic field) de motion ⁹ Describes the effects produced by a force (change in the state of motion of a body, distortion of a body) es of forces Recognizes different types of forces in technical objects or technological systems (e.g. gravitational force in a chute, magnetic force exerted by an electromagnet)	1	2	3	*	
Foodaa.a.b.	rce ar ire Force i. Type i. Equi	magnet, changing the intensity of a magnetic field) nd motion ⁹ Describes the effects produced by a force (change in the state of motion of a body, distortion of a body) es of forces Recognizes different types of forces in technical objects or technological systems (e.g. gravitational force in a chute, magnetic force exerted by an electromagnet) librium of two forces Describes the conditions under which a body subjected to two forces can be	1	2	3	*	
Foodaa.a.b.	Force i. Type i. Equi i. Rela	magnet, changing the intensity of a magnetic field) nd motion ⁹ Describes the effects produced by a force (change in the state of motion of a body, distortion of a body) es of forces Recognizes different types of forces in technical objects or technological systems (e.g. gravitational force in a chute, magnetic force exerted by an electromagnet) librium of two forces Describes the conditions under which a body subjected to two forces can be in equilibrium	1	2	3	*	
Foodaa.a.b.	rce ar ire Forc i. Type i. Equi i. Rela	magnet, changing the intensity of a magnetic field) nd motion ⁹ e Describes the effects produced by a force (change in the state of motion of a body, distortion of a body) es of forces Recognizes different types of forces in technical objects or technological systems (e.g. gravitational force in a chute, magnetic force exerted by an electromagnet) librium of two forces Describes the conditions under which a body subjected to two forces can be in equilibrium tionship between constant speed, distance and time	1	2	3	*	
b.	rce ar ire Force i. Type i. Equi i. Rela i.	magnet, changing the intensity of a magnetic field) nd motion ⁹ Describes the effects produced by a force (change in the state of motion of a body, distortion of a body) es of forces Recognizes different types of forces in technical objects or technological systems (e.g. gravitational force in a chute, magnetic force exerted by an electromagnet) librium of two forces Describes the conditions under which a body subjected to two forces can be in equilibrium tionship between constant speed, distance and time Qualitatively describes the relationship between speed, distance and time Applies the mathematical relationship between constant speed, distance and	1	2	3	*	
b.	rce and inception inceptio	magnet, changing the intensity of a magnetic field) nd motion ⁹ Describes the effects produced by a force (change in the state of motion of a body, distortion of a body) as of forces Recognizes different types of forces in technical objects or technological systems (e.g. gravitational force in a chute, magnetic force exerted by an electromagnet) librium of two forces Describes the conditions under which a body subjected to two forces can be in equilibrium tionship between constant speed, distance and time Qualitatively describes the relationship between speed, distance and time Applies the mathematical relationship between constant speed, distance and time (v = d/Δt)	1	2	3	*	

f.	Relationship between work, force and distance travelled			
	 Describes qualitatively the relationship between the work done, the force applied on a body and the distance travelled by the body 			*
	ii. Applies the mathematical relationship between work, effective force and distance travelled (W = $F\Delta d$)			*
g.	Distinction between mass and weight			
+	i. Qualitatively describes the relationship between mass and weight		*	
•	ii. Applies the mathematical relationship between mass and weight ($F_g = mg$)		*	

- 1. See Techniques, Science, Separating mixtures (Techniques-Science, b).
- 2. These concepts are presented under The Living World, Life-sustaining processes (LW, B, f).
- 3. See Technological World, Mechanical Engineering, Technological systems, Transformation of energy (TW, B, 2, c).
- 4. The definition of temperature was covered in Cycle One. See *Properties* above (MW, A, 1, c).
- 5. This concept, related to program section Forces and motion (TW, B, 1), is presented here.
- 6. See Law of the conservation of energy (MW, B, 4, b).
- 7. See The Material World, Changes, Changes in matter, Mixtures (MW, B, 1, b).
- 8. See The Technological World, Electrical engineering (TW, C).
- 9. For the concepts of *Force and motion* in Secondary Cycle One, as presented in the program, see *The Technological World*.

Applied General Education Path

The Living World

In The Living World, students acquire scientific and technological knowledge pertaining to life as it relates to molecules, cells, organisms and ecosystems.

In secondary school, students explore increasingly complex phenomena and technical objects and seek answers and solutions to a variety of problems. They acquire scientific knowledge about The Living World that helps them understand and explain the factors at play in different scientific issues. This knowledge, along with the knowledge they acquire in other areas of the program, particularly The Earth and Space, enables them to understand the complexity of the relationships between living organisms and their planet. Students refine their understanding of the concepts related to The Living World by using the experimental method, the observation method and modelling.

In Secondary III, students explore applications related to the seven technological fields, which enables them to make connections between technology and The Living World. In Secondary IV, they continue constructing and applying their knowledge about The Living World by analyzing and designing a variety of applications related to the same technological fields. Thus they acquire a better understanding of the contribution of science and technology to human and environmental health. In the optional Science and the Environment program, students consolidate their knowledge and form their own opinions regarding two environmental issues they will be asked to examine.

Student constructs knowledge with teacher guidance.					
★ Student applies knowledge by the end of the school year.		S	econ	dary	/
Student reinvests knowledge.		.		. T	0.5
Statements preceded by the symbol • indicate knowledge specific to the compulsory Applied Science and Technology program. Most of these statements are, however, found in the progression of learning for the optional Environmental Science and Technology program.	Су	cle ne	Cy Tv	cle	SE Cycle Two
A. Diversity of life forms	1	2	3	4	4
Elementary school Students classify animals by kingdom and, in the case of vertebrates, by class (mammals, reptile amphibians). They describe the physical and behavioural characteristics that demonstrate that a its environment (e.g. fins, claws, colour, ability to burrow into the ocean floor, migration).				adap	ted to
Secondary school					
1. Ecology	S	Т	AS	ST.	SE
a. Habitat					
 i. Names the characteristics that define a habitat (e.g. geographic location, climate, flora, fauna, proximity of man-made constructions) 	\rightarrow	*			
ii. Describes the habitat of certain species	\rightarrow	*			
b. Ecological niche					
 Names the characteristics that define an ecological niche (e.g. habitat, diet, daily rhythms) 	\rightarrow	*			
ii. Describes the ecological niche of an animal species	\rightarrow	*			
c. Species					
 Names the characteristics that define a species (common physical characteristics; natural, viable and fertile reproduction) 	\rightarrow	*			
d. Population					
i. Distinguishes between a population and a species	\rightarrow	*			

i.	Ecosystems	
	Defines an ecosystem as the relationships between the individuals in a community and abiotic factors in the environment	*
ii.	Disturbances	
	Defines a disturbance in a community	*
	Explains the effects of certain factors that disturb the ecological balance (e.g. human activity, natural disasters)	×
iii.	Trophic relationships	
	Describes the trophic levels (producers, consumers, decomposers)	*
	Explains the relationships between the trophic levels of a food web	*
iv.	Primary productivity	
	Defines primary productivity as the quantity of organic matter produced by plants in a given territory	*
	Explains the effects of certain factors on primary productivity (e.g. bees help pollinate fruit trees, pathogenic microorganisms hinder plant growth)	rk.
V.	Material and energy flow	
	Describes material and energy flow in an ecosystem	*
vi.	Chemical recycling	
	Describes certain processes underlying chemical recycling (e.g. action of microorganisms and decomposers, erosion)	*
vii.	Factors that influence the distribution of biomes	
	Describes the geographical and climatic factors that affect the distribution of biomes (e.g. latitude, humidity, temperature, salinity)	*
f. Ecot	toxicology	
i.	Contaminant ¹	
	Defines a contaminant as an agent that causes changes in the physical, chemical or biological properties of an environment or an organism	
ii.	Bioaccumulation	
	Defines bioaccumulation as the process by which a contaminant from the environment or food supply accumulates in an organism	
	Explains bioaccumulation in food chains (biomagnification)	
iii.	Bioconcentration	
	Defines bioconcentration as a special case of bioaccumulation by which an organism accumulates a contaminant through direct contact with its environment (from sources other than food)	
iv.	Toxicity threshold	
	Defines the toxicity threshold of a substance as the minimum concentration of a substance that produces a significant harmful effect in an organism (mg/kg of the organism's mass)	
	Describes factors that influence the toxicity of a contaminant (e.g. concentration, characteristics of the environment into which it is released, nature of the organisms with which it is in contact, duration of	

2. Diversity of life forms		ST		AS	ST	SE
a. Physical and behavioural adaptation						
 Describes physical adaptations that enable animals and plants their chances of survival (e.g. coat colour matched to the envir of leaves) 		→	k			
 ii. Describes behavioural adaptations that enable animals and please their chances of survival (e.g. movement in groups, phototropic 		>	*			
b. Evolution						
i. Describes the stages in the evolution of living organisms	-	→	k			
ii. Explains the natural selection process	-	>	k			
c. Taxonomy						
 Defines taxonomy as a system for classifying living organisms most part on their anatomical and genetic characteristics 	based for the	>	*			
ii. Identifies a species using a taxonomic key	-	→	k			
d. Genes and chromosomes ²						
i. Locates chromosomes in the cell	-	→	k			
ii. Defines a gene as part of a chromosome	-	> 7	k			
iii. Describes the role of genes (transmission of hereditary charac	teristics) -	→ n	k			
B. Life-sustaining processes	1	1 :	2	3	4	4
Students explain the essential needs of living organisms (e.g. food, respiration) (transformation of energy, growth, maintenance of systems and body temperature photosynthesis, which they distinguish from respiration. Secondary school						
a. Characteristics of living things						
i. Describes certain characteristics common to all living things (not relationships, adaptation, reproduction)	utrition,	→	k			
b. Plant and animal cells						
i. Defines the cell as the structural unit of life	-	→	k			
ii. Names vital functions carried out by cells	_	> 7	*			
iii. Distinguishes between animal and plant cells	_	> 7	*			
c. Cellular components visible under a microscope						
i. Identifies the main cellular components visible under a microsomembrane, cytoplasm, nucleus, vacuoles)	cope (cell	→	k			
ii. Describes the role of the main cellular components visible und	er a microscope	> 7	k			
d. Inputs and outputs (energy, nutrients, waste)						
i. Names cellular inputs	-	→	k			
ii. Names cellular outputs	-	→ 7	k			
e. Osmosis and diffusion						
i. Distinguishes between osmosis and diffusion	_	>	*			

i	Names the inputs and outputs involved in photosynthesis	\rightarrow	*		
		-	^		
	Represents the photosynthesis reaction in a balanced equation				7
	Names the inputs and outputs involved in respiration	\rightarrow	*		
iv.	Represents the photosynthesis reaction in a balanced equation				7
C. Systems	<u>-</u>	1	2	3	4 4
	chool be the functions of certain parts of their anatomy (limbs, head). They associate the their main functions.	oarts	of ar	nimal a	ınatom
Secondary sch	nool				
	Nutrition				
1. Digestive	system	S	т	AST	гв
a. Dige	estive tract				
i.	Names the main parts of the digestive tract (mouth, esophagus, stomach, small intestine, large intestine, anus)			*	
ii.	Explains the role of the digestive tract (decomposition of food, absorption of nutrients and water, elimination of waste)			*	
iii.	(mouth, stomach, small intestine, large intestine)			*	
b. Dige	estive glands				
i.	Names the main digestive glands (salivary glands, gastric glands, pancreas, liver, intestinal glands)			*	
ii.	Describes the function of the main digestive glands (e.g. secretion of saliva, gastric enzymes, digestive juices, bile)			*	
c. Type	es of foods				
i.	Describes the main biological functions of the different food constituents (water, proteins, carbohydrates, fats, vitamins, minerals)			*	
ii.	Associates food constituents with their main sources (e.g. proteins with meat and meat substitutes)			*	
d. Ene	rgy value of different foods				
i.	Evaluates the energy and nutritional value of different foods			*	
e. Trar	sformation of food				
i.	(Describes the two types of transformation of food that take place in the digestive system (mechanical, chemical)			*	
ii.	Associates the organs in the digestive tract with the type of transformation they perform (e.g. mechanical action of teeth, chemical action of glands)			*	
2. Respirator	ry and circulatory and systems	S	Т	AST	Γ S
a. Res	piratory system)				
i.	(Names the main parts of the respiratory system (nasal cavity, pharynx, trachea, bronchi, lungs)			*	
ii.	Explains the role of the respiratory system (gaseous exchanges between the blood and the surrounding air)			*	
iii.	Describes the function of the nasal cavity and lungs			*	

i. Describes the main function of plasma (transportation of the blood's soluble and formed elements)			*	
ii. Names the formed elements of the blood (red blood cells, white blood cells, platelets)			*	
iii. Describes the main function of the formed elements of the blood			*	
c. Compatibility of blood types				
i. Determines the compatibility or incompatibility of blood types (e.g. an individual with type A blood can only receive type O- or type A- blood)			*	
d. Circulatory system				
i. Names the main parts of the circulatory system (heart, types of blood vessels, pulmonary and systemic circulation)			*	
ii. Explains the role of the circulatory system (transportation and exchange of gases, nutrients and waste)			*	
iii. Describes the function of the main parts of the circulatory system (heart, arteries and veins, capillaries)			*	
e. (Lymphatic system)				
i. Names the main parts of the lymphatic system (lymph, antibodies)			*	
ii. Explains the role of the lymphatic system (circulation of antibodies outside the blood vessels)			*	
iii. Describes two ways of acquiring active immunity (production of antibodies, vaccination)			*	
3. (Excretory system)	S	Т	AST	SE
a. (Urinary system)			_	_
i. Names the main parts of the urinary system (kidneys, ureters, bladder, urethra)			*	
ii. Explains the role of the excretory system (filtration of the blood, elimination of cellular waste)			*	
iii. Describes the function of the kidneys and bladder			*	
b. Components of urine				
i. Names the main components of urine (water, mineral salts, urea)			*	
c. (Maintaining a balanced metabolism)				
c. (Maintaining a balanced metabolism) i. (Explains the role of the kidneys, lungs and sweat glands in maintaining a balanced metabolism)			*	
i. Explains the role of the kidneys, lungs and sweat glands in maintaining a			*	
i. Explains the role of the kidneys, lungs and sweat glands in maintaining a balanced metabolism	S		* AST	SE
i. Explains the role of the kidneys, lungs and sweat glands in maintaining a balanced metabolism Relationships	S			SE
i. Explains the role of the kidneys, lungs and sweat glands in maintaining a balanced metabolism Relationships 1. Nervous and musculoskeletal systems a. Central nervous system i. Identifies the parts of the central nervous system (brain, spinal cord)	s	т		SE
i. Explains the role of the kidneys, lungs and sweat glands in maintaining a balanced metabolism (Relationships) 1. (Nervous and musculoskeletal systems) a. (Central nervous system)	S	Т	AST	SE
i. Explains the role of the kidneys, lungs and sweat glands in maintaining a balanced metabolism Relationships 1. Nervous and musculoskeletal systems a. Central nervous system i. Identifies the parts of the central nervous system (brain, spinal cord) ii. Explains the role of the central nervous system (e.g. to manage complex)	S	Т	AST	SE
i. Explains the role of the kidneys, lungs and sweat glands in maintaining a balanced metabolism Relationships 1. Nervous and musculoskeletal systems a. Central nervous system i. Identifies the parts of the central nervous system (brain, spinal cord) ii. Explains the role of the central nervous system (e.g. to manage complex behaviours and process sensory information and the related responses)	S	Т	★ ★	SE
i. Explains the role of the kidneys, lungs and sweat glands in maintaining a balanced metabolism Relationships 1. Nervous and musculoskeletal systems a. Central nervous system i. Identifies the parts of the central nervous system (brain, spinal cord) ii. Explains the role of the central nervous system (e.g. to manage complex behaviours and process sensory information and the related responses) iii. Describes the functions of the brain and the spinal cord	S	Т	★ ★	SE

 Explains the role of the peripheral nervous system (transportation of nerve impulses from the senses to the brain and from the brain to the muscles)
ii. Neural inflow
- Associates nerves with the transmission of nerve impulses
 ─ Distinguishes between voluntary acts and reflex arcs
c. (Sensory receptors)
i. Eye
 Names the parts of the eye involved in vision (iris, cornea, crystalline lens, retina)
 Describes the function of the main parts of the eye
ii. <mark>Ear</mark>
- Names the main parts of the ear involved in hearing and balance (auditory canal, ear drum, ossicles, cochlea, semicircular canals)
 Describes the function of the main parts of the ear involved in hearing
Describes the role of the semicircular canals in maintaining balance
iii. Tongue
 Describes the function of the taste buds on the tongue (transformation and transmission of flavours: sweet, salty, sour, bitter, umami)
iv. (Nose)
Names the parts of the nose involved in smelling (nasal cavity, olfactory bulb)
 Describes the function of the olfactory bulb
v. Skin
 Describes the function of the skin's sensory receptors (transformation) and transmission of feelings of pressure, temperature and pain)
d. Musculoskeletal system
i. Function of bones, joints and muscles
 Names the main parts of the skeleton (head, thorax, spinal column,) upper and lower limbs)
 Describes the functions of the main parts of the skeleton (e.g. the spinal) column protects the spinal cord and allows the trunk to move)
Explains the role of the musculoskeletal system
Describes how pairs of antagonistic muscles work (e.g. biceps, triceps) ★
Describes how joints work (linking bone to bone, mobility)
ii. (Types of muscles)
- Associates the different types of muscles (smooth, skeletal, heart) with the tissues in which they are found
iii. (Types of joint movement)
 Describes types of joint movement (e.g. flexion, rotation)

D. Sui	vival of species	1	2	3	4	4		
	ry school lescribe the growth and reproduction of flowering plants and different animals.							
Seconda	y school							
1. Rep	roduction	S	ST		ST		ST	SE
a.	Asexual and sexual reproduction							
	 Distinguishes between asexual and sexual reproduction (e.g. sexual reproduction involves gametes) 	\rightarrow	*					
b.	Reproductive mechanisms in plants							
	i. Describes asexual reproductive mechanisms in plants (e.g. cutting, layering)	\rightarrow	*					
	ii. Describes the sexual reproductive mechanism in plants (flowering plants)	\rightarrow	*					
C.	Reproductive mechanisms in animals							
	 Describes the roles of the male and female in the reproduction of certain types of animals (e.g. birds, fish, mammals) 	\rightarrow	*					
d.	Reproductive organs							
	 Names the main male and female reproductive organs (penis, testicles, vagina, ovaries, Fallopian tubes, uterus) 	\rightarrow	*					
e.	Gametes							
	i. Names the male and female gametes	\rightarrow	*					
	ii. Describes the role of gametes in reproduction	\rightarrow	*					
f.	Fertilization							
	i. Describes fertilization in humans	\rightarrow	*					
g.	Pregnancy							
	 Names the stages of human development during pregnancy (zygote, embryo, fetus) 	\rightarrow	*					
h.	Stages of human development							
	 Describes the stages of human development (childhood, adolescence, adulthood) 	\rightarrow	*					
i.	Contraception							
	 Describes contraceptive methods (e.g. condom, ovulation suppression agents) 	\rightarrow	*					
	 Describes the advantages and disadvantages of different contraceptive methods 	\rightarrow	*					
j.	Methods of preventing the implantation of the zygote in the uterus	_						
	 Names methods of preventing the implantation of the zygote in the uterus (intrauterine device, day-after pill) 	\rightarrow	*					
k.	Sexually transmitted and blood-borne diseases ⁴							
	i. Names sexually transmitted and blood-borne diseases	\rightarrow	*					
	 Describes behaviours to prevent contracting sexually transmitted and blood-borne diseases (e.g. wearing a condom) 	\rightarrow	*					
	iii. Describes responsible behaviours to adopt after being diagnosed with a sexually transmitted or blood-borne disease (e.g. informing one's partner)	\rightarrow	*					

2. Cell division ⁵	S	Т	AST	SE						
a. Mitosis										
i. Describes the functions of mitosis (reproduction, growth, regeneration)			*							
b. Meiosis										
i. Describes the function of meiosis (production of gametes)			*							
c. Genetic diversity										
i. Associates genetic diversity with sexual reproduction			*							
Reproduction										
1. Reproductive system	S	Т	AST	SE						
a. (Puberty (male and female)										
 i. Describes physical and psychological changes that occur at puberty (e.g. appearance of body hair, voice change, ability to procreate, need for independence) 			*							
b. (Hormone regulation in men)	b. Hormone regulation in men									
i. <mark>Spermatogenesis</mark>										
 Names the hormones responsible for the formation of spermatozoa (follicle stimulating hormone [FSH], luteinizing hormone [LH], (testosterone) 			*							
ii. Erection										
 Describes the physiology of erection 			*							
iii. <mark>Ejaculation</mark>										
 Explains the function of ejaculation in reproduction 			*							
c. (Hormone regulation in women)										
i. <mark>Oogenesis</mark>										
 Names the hormones responsible for the maturation of the ovarian follicle (FSH, LH, estrogen, progesterone) 			*							
ii. <mark>Ovarian cycle</mark>										
 Describes the hormone changes that occur during the menstrual cycle 			*							
iii. Menstrual cycle										
 Describes the main stages in the menstrual cycle (e.g. menstruation, endometrium development, ovulation) 			*							

- 1. See The Earth and Space, Characteristics of the Earth, Contamination (ES, A, 2, h; ES, A, 3, c; ES, A, 4, f).
- 2. See The Living World, Survival of species, Cell division (LW, D, 2).
- 3. The Secondary IV concepts related to photosynthesis and respiration are presented in the program under *Material World*, *Changes*, *Chemical changes*.
- 4. This replaces the term "sexually transmitted diseases" used in the program.
- 5. For concepts related to *Genes and chromosomes* addressed in Cycle One, see *The Living World, Diversity of life forms* above (<u>LW, A, 2, d</u>).

Applied General Education Path

The Earth and Space

In The Earth and Space, students acquire scientific and technological knowledge pertaining to space and interactions in the biosphere.

In secondary school, students explore different phenomena that occur on Earth and in space and related technical objects. They seek answers and solutions to a variety of problems. They acquire scientific knowledge about The Earth and Space that helps them explain the factors at play in different scientific issues. This knowledge, along with the knowledge they acquire in other areas of the program, enables them to understand scientific models, theories and laws. Students refine their understanding of the concepts related to The Earth and Space by using the experimental method, technological analysis and modelling.

In Secondary IV, students explore applications related to the seven technological fields, which enables them to make connections between technology and The Earth and Space. Thus they continue constructing and applying their scientific and technological knowledge and further explore the impact of technology on the biosphere. In the optional Science and the Environment program, students consolidate their knowledge and form their own opinions regarding two environmental issues they will be asked to examine.

Student constructs knowledge with teacher guidance.									
★ Student applies knowledge by the end of the school year.		′							
Student reinvests knowledge.	9	ST AST			SE				
Statements preceded by the symbol • indicate knowledge specific to the compulsory Applied Science and Technology program. Most of these statements are, however, found in the progression of learning for the optional Environmental Science and Technology program.	ST Cycle One		Cycle Cy		Cycle Cy		Cycle Cy		Cycle Two
A. Characteristics of the Earth	1	2	3	4	4				
Elementary school Students recognize visible structures on the surface of the Earth (e.g. continents, oceans, ice ca volcanoes). They describe the effects of the quality of air, water and soil on living beings (e.g. ill decrease in population). They compare the properties of different types of soil (e.g. composition and heat).	Iness	es, ir	ncrea	ase c					
Secondary school									
General characteristics of the Earth	S	Т	AS	ST	SE				
a. Internal structure of the Earth									
 Describes the main characteristics of the three parts of the internal structure of the Earth (crust, mantle, core) 	\rightarrow	*							
2. Lithosphere	ST		AS	ST	SE				
a. General characteristics of the lithosphere									
 Defines the lithosphere as the outer shell of the Earth comprising the crust and the upper mantle 	\rightarrow	*							
ii. Describes the main relationships between the lithosphere and human activity (e.g. survival, agriculture, mining, land-use planning)	\rightarrow	*							
b. Relief									
 Describes relationships between relief (topology) and geological and geophysical phenomena¹ (e.g. the retreat of a glacier causes the formation of a plain) 	\rightarrow	*							
ii. Describes the effect of relief on human activities (e.g. transportation, construction, sports, agriculture)	\rightarrow	*							

	Types of						
	sed	scribes the formation of three types of rock: igneous, metamorphic, limentary	\rightarrow	*			
		ssifies rocks by method of formation (e.g. granite is an igneous rock, lime sedimentary rock, slate is a metamorphic rock)	\rightarrow	*			
	iii. Dis	tinguishes between rocks and minerals	\rightarrow	*			
d.	Minerals		_			_	
		mes basic minerals based on their properties (e.g. colour, hardness, gnetism)	\rightarrow	*			
	ii. Dis	tinguishes between minerals and ore				*	
		scribes some of the environmental impacts of mining and of the nsformation of minerals				*	
e.	Types of	soil					
	i. Cla	ssifies soils based on their composition (e.g. sand, clay, organic material)	\rightarrow	*			
f.	Soil horiz	cons (profile)					
		scribes the structure of a soil (superimposition of layers of various npositions and thicknesses)					7
		plains the chemical and biological reactivity of a soil it based on its nposition (e.g. oxidation, acid-base neutralization, decomposition)					7
g.	Buffering	capacity of the soil					
	i. Def	fines the buffering capacity of a soil as its ability to limit pH variations					7
	ii. Exp	plains the advantages of a good soil buffering capacity					7
h.	Contamir	nation					
	i. Naı	mes soil contaminants ²					7
3. Hyd	osphere		S	Т	AS	т	S
а.	General	characteristics of the hydrosphere					
٠.	i De	scribes the distribution of fresh water and salt water on the Earth's surface	\rightarrow	*			
<u></u>		g. glaciers contain inaccessible fresh water)	7	^			
J.	ii. Des		→	*			
	ii. Des	g. glaciers contain inaccessible fresh water) scribes the main interactions between the hydrosphere and the nosphere (e.g. heat exchanges, climate regulation, meteorological enomena)					
	ii. Des atm phe	g. glaciers contain inaccessible fresh water) scribes the main interactions between the hydrosphere and the nosphere (e.g. heat exchanges, climate regulation, meteorological enomena)				*	
	ii. Des atm phe Catchme i. Def ii. Des	g. glaciers contain inaccessible fresh water) scribes the main interactions between the hydrosphere and the nosphere (e.g. heat exchanges, climate regulation, meteorological enomena) nt area				*	
b.	ii. Des atm phe Catchme i. Def ii. Des	g. glaciers contain inaccessible fresh water) scribes the main interactions between the hydrosphere and the hosphere (e.g. heat exchanges, climate regulation, meteorological enomena) Int area fines a catchment area as a territory surrounding a waterway scribes some of the impacts of human activity on the waterways in a chment area					
b.	ii. Des atm phe Catchme i. Def ii. Des catc	g. glaciers contain inaccessible fresh water) scribes the main interactions between the hydrosphere and the hosphere (e.g. heat exchanges, climate regulation, meteorological enomena) Int area fines a catchment area as a territory surrounding a waterway scribes some of the impacts of human activity on the waterways in a chment area					1
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4. Atmosphere	ST		ST AST		SE
a. General characteristics of the atmosphere					
 i. Locates the main layers of the atmosphere (troposphere, stratosphere, mesosphere, thermosphere) 	\rightarrow	*			
 ii. Describes the composition of pure air at sea level (nitrogen, oxygen, carbon dioxide, water vapour) 	\rightarrow	*			
iii. Describes the relationships between the atmosphere and certain human activities (e.g. recreation, transportation, energy consumption)	\rightarrow	*			
b. Greenhouse effect					
i. Describes the greenhouse effect					*
 ii. Explains some consequences of a higher concentration of greenhouse gases (e.g. global warming that could result in higher sea levels, disturbances in ecosystems or the melting of glaciers) 					*
c. Air mass					
i. Describes the properties of an air mass (temperature, humidity, pressure)				*	
ii. Explains the formation of clouds when two different air masses meet				*	
d. Atmospheric circulation					
Describes the main factors responsible for atmospheric circulation (e.g. pressure variations, uneven heating of the Earth's surface)					*
ii. Describes the effect of prevailing winds on the dispersal of air pollutants in a given region					*
e. Cyclone and anticyclone					
i. Explains the formation of cyclones (low-pressure areas) and anticyclones (high-pressure areas)				*	
f. Contamination					
i. Names air contaminants ⁴					*
B. Geological and geophysical phenomena	1	2	3	4	4
Elementary school Students explain the water cycle (evaporation, condensation, precipitation, runoff, infiltration) an of precipitation (rain, snow, hail, freezing rain). Concepts related to energy play an important role school program. Students explain that sunlight, moving water and wind are renewable energy redifferentiate them from nonrenewable energy resources such as fossil fuels (e.g. gasoline, propagas). They describe technologies used to convert renewable energy into electricity (hydroelectric solar panels).	e in tl sour ane, l	he el ces. butai	eme They ne, o	ntary / oil, na	tural
Secondary school					
a. Tectonic plate					
 Describes the main elements of the theory of tectonic plates (e.g. plate, subduction zone, mid-oceanic ridge) 	\rightarrow	*			
b. Orogenesis					
Describes the formation of mountains, folding and breaks (tectonic plate movements)	\rightarrow	*			
c. Volcano					
i. Describes a volcanic eruption	\rightarrow	*			
ii. Describes the geographical distribution of volcanoes	\rightarrow	*			

d.	Earthquake					
	 Describes the processes that cause earthquakes (e.g. tectonic plate movements, slides) 	\rightarrow	*			
e.	Erosion					
	 Describes different types of erosion (e.g. soils dried by the wind, fragmentation of rocks caused by water freezing and thawing) 	\rightarrow	*			
f.	Winds					
	 Names the main factors responsible for wind (e.g. convection movements, movement of air masses) 	\rightarrow	*			
g.	Water cycle					
	i. Explains the water cycle (phase changes, energy exchanges)	\rightarrow	*			
h.	Natural energy sources					
	 Describes the role of solar energy as a natural energy source (e.g. wind, tornadoes, hurricanes, storms) 	\rightarrow	*			
i.	Renewable and nonrenewable energy resources					
	Distinguishes between renewable and nonrenewable energy resources (e.g. Sun, molten rock, moving water, oil)	\rightarrow	*			
	 Describes technologies used to produce electricity using the energy resources in the lithosphere, hydrosphere and atmosphere 				*	
	 Describes the main impact of the use of energy resources in the lithosphere, hydrosphere and atmosphere 				*	
C. Ast	ronomical phenomena	1	2	3	4	4
and moon	ry school earn that the cycle of day and night is related to the Earth's rotation. They distinguish is in our solar system. They describe seasonal changes (e.g. temperature variations, l in), in particular the apparent position of the Sun and its influence on the length of sha	umino	osity			
Secondar	y school					
1. Con	cepts related to astronomy	S	т	A	ST	SE
a.	Universal Gravitation					
	i. Defines gravitation as a force of mutual attraction between bodies	\rightarrow	*			
b.	Earth-Moon system					
	 Describes the tides in terms of the gravitational effect of the Earth-Moon system 				*	
C.	Light					
	i. Defines light as a form of radiant energy ⁵	\rightarrow	*			
	 ii. Describes properties of light (propagation in a straight line, diffuse reflection by surfaces) 	\rightarrow	*			
	 Explains different phenomena using the properties of light (cycles of day and night, seasons, phases of the Moon, eclipses) 	\rightarrow	*			
d.	Solar energy flow					
	 Describes the main factors that affect the quantity of solar energy that reaches the Earth's surface (e.g. reflection and absorption of solar energy by the atmosphere or surfaces) 				*	

2. Sola	system	S	Т	AST	SE
a.	Characteristics of the solar system				
	 i. Compares some of the characteristics of the planets in our solar system (e.g. distances, relative size, composition) 	\rightarrow	*		
b.	Cycles of day and night				
	i. Explains the alternation of day and night in terms of the Earth's rotation	\rightarrow	*		
C.	Phases of the Moon				
	i. Explains the phases of the lunar cycle	\rightarrow	*		
d.	Eclipses				
	i. Explains a lunar or solar eclipse	\rightarrow	*		
e.	Seasons				
	 i. Explains the phenomenon of seasons in terms of the position of the Earth with respect to the Sun (tilt, revolution) 	\rightarrow	*		
f.	Comets				
	 Describes the main parts of a comet (core of ice and rock, tail of gas, and tail of dust) 	\rightarrow	*		
g.	Aurora borealis (northern lights)				
	 Locates the geographic regions where the aurora borealis occurs (polar regions) 	\rightarrow	*		
	ii. Identifies the atmospheric layer in which the aurora borealis occurs	\rightarrow	*		
h.	Meteoroid impact				
	 i. Identifies traces left by meteoroid impacts in Québec (e.g. craters, astroblemes) 	\rightarrow	*		

- 1. See The Earth and Space, Geological and geophysical phenomena below (ES, B).
- 2. See The Living World, Ecotoxicology, Contaminant (LW, A, 1, f, i).
- 3. See The Living World, Ecotoxicology, Contaminant (LW, A, 1, f, i).
- 4. See The Living World, Ecotoxicology, Contaminant (LW, A, 1, f, i).
- 5. See The Material World, Changes, Transformation of energy, Forms of energy (MW, B, 4, a).

Applied General Education Path

The Technological World

In The Technological World, students acquire and apply scientific and technological knowledge.

In Secondary school, students analyze and design increasingly complex technical objects and seek solutions to increasingly sophisticated technological problems. The technical and technological knowledge they acquire helps them understand the objects and factors at play in different scientific issues as well as evaluate possible technological solutions. It also helps them apply knowledge they acquire in other areas of the program, in particular The Material World.

In Secondary III, students analyze and design technical objects, processes or systems related to the seven technological fields, which enables them to make connections between human beings and technology and, consequently, to integrate their knowledge of The Living World. In Secondary IV, they continue constructing their scientific and technological knowledge and examine the influence of technology on the world around us by exploring a variety of applications related to the same technological fields. Thus they discover how technology helps us understand and improve our world. In the optional Science and the Environment program, students consolidate their knowledge and form their own opinions regarding two environmental issues they will be asked to examine. They can thus apply their knowledge of the Technological World in new contexts.

→ Stude	t constructs knowledge with teacher guidance.					
★ Stude	at applies knowledge by the end of the school year.		S	econ	dary	/
Stude	t reinvests knowledge.		_	Δ.	`_	0.5
Technology pro	ceded by the symbol • indicate knowledge specific to the compulsory Applied Science and gram. Most of these statements are, however, found in the progression of learning for the mental Science and Technology program.	Су	T cle ne	Cy Tv	cle	SE Cycle Two
A. Grap	ical language ¹	1	2	3	4	4
Elementary Students lea	school n symbols associated with motion and parts and use them to produce or interpret o	liagra	ams	or dr	awin	gs.
Secondary	chool					
а. С	agram of principles (design plan)					
	i. Defines a diagram of principles as a representation used to effectively explain the operation of a technical object	\rightarrow	*			
	ii. Associates the functional elements of a technical object with the appropriate diagram of principles	\rightarrow	*			
	iii. Explains the operation of a simple technical object by drawing a diagram illustrating the active forces and the resulting motion	\rightarrow	*			
	 Names the subassemblies and parts essential to the operation of a technical object 	\rightarrow	*			
	v. Indicates certain principles of simple machines illustrated in a technical object (e.g. a lever in a wheelbarrow, a wedge in an axe)	\rightarrow	*			
b. C	onstruction diagram (technical diagram)					
	i. Defines a construction diagram as a representation used to effectively explain the construction and assembly of a technical object	\rightarrow	*			
	ii. Associates the shape and arrangement of parts of technical objects with the appropriate construction diagram	\rightarrow	*			
	iii. Explains the construction of a simple technical object by drawing a diagram illustrating the assembly and arrangement of parts	\rightarrow	*			
	iv. Names the components of a simple technical object	\rightarrow	*			
	v. Indicates the links and guiding controls on a construction diagram	\rightarrow	*			

	i. Chooses the appropriate type of diagram for a given representation (e.g. uses		
	a construction diagram to represent assembly solutions, a diagram of principles to represent the operation of an object)	\rightarrow	*
	 Represents different types of motion related to the operation of an object using the appropriate symbols (rectilinear translation, rotation, helical) 	\rightarrow	*
d.	Geometric lines		
	i. Associates a drawing with a combination of geometric lines (e.g. the drawing of a rounded corner of a table is an arc joined to two sides of a right angle)	*	
e.	Basic lines	 	
	 i. Names basic lines in a drawing (visible contour, hidden contour, centre, extension, dimension lines) 	*	
	ii. Associates the basic lines in a drawing with the contours and details of a simple part	*	
f.	Orthogonal projections		
	 Associates the types of projection with their use (multiview and isometric projections) 	*	
	ii. Interprets drawings representing parts in multiview orthogonal projection	*	
	iii. Represents simple shapes in multiview orthogonal projection	*	
	iv. Represents simple shapes in isometric projection	*	
+	Interprets assembly drawings of technical objects consisting of a small number of parts		*
g.	Scales ³		
	i. Associates scales with their use (actual-size representation, reduction or enlargement of an object)	*	
	ii. Chooses a simple scale for a drawing (e.g. 1:1, 1:2, 5:1)	*	
	iii. Takes the scale into account when interpreting drawings	*	
h.	Forms of representation		
	i. Defines perspective drawing, oblique projection and axonometric projection	*	
	ii. Sketches simple objects freehand using different forms of representation	*	
i.	Axonometric projection: exploded view (reading)		
+	i. Names the characteristics of an exploded view	*	
+	ii. Explains the purpose of exploded views (projection accompanying the assembly instructions or specifications for an object)	*	
j.	Cross-sectional views and sections		
	i. Cross-sectional views		
	 Describes the purpose of cross-sectional views in technical drafting 	*	
	 Interprets a technical drawing with cross-sectional views 	*	
	Represents a simple shape in a cross-sectional view	*	

	 Describes the purpose of removed sections and revolved sections 			*		
k.	Dimensioning and tolerances					
	i. Dimensioning					
	 Describes the main dimensioning rules (e.g. to make a drawing easy to read, avoid crossing dimensioning lines) 			*		
	 Interprets technical drawings including the dimensions required for manufacturing purposes) 			*		
	ii. Tolerances					
+	 Defines tolerance as the required manufacturing precision (dimensions) indicated on the drawing, along with allowances) 			*		
	iii. Functional dimensioning					
*	 Defines functional dimensioning as the set of specific tolerances related to certain parts responsible for the smooth operation of an object (e.g. the distance between two axes is a determining factor in the operation of sprocket wheels in a gear assembly) 				*	
l.	Developments (prism, cylinder, pyramid, cone)					
*	 Associates the development of three-dimensional shapes with the construction of objects from sheet stock (e.g. cardboard boxes, metal air ducts) 				*	
+	ii. Draws developments of simple solids (e.g. pyramid, cylinder, cube)				*	
B. Med	chanical engineering	1	2	3	4	4
arte in mo	s, cams, springs), distinguish between translation and rotation and describe a simple		ence			ar inical
	otion (e.g. in a door lock, the lever rotates and the motion of the bolt is rectilinear trans ry school		ence			
Secondar	otion (e.g. in a door lock, the lever rotates and the motion of the bolt is rectilinear trans	slatio	ence	of m		
Secondar 1. Forc	otion (e.g. in a door lock, the lever rotates and the motion of the bolt is rectilinear trans	slatio	ence n).	of m	necha	inical
Secondar 1. Forc	otion (e.g. in a door lock, the lever rotates and the motion of the bolt is rectilinear trans ry school ees and motion ⁴	slatio	ence n).	of m	necha	inical
1. Forc	otion (e.g. in a door lock, the lever rotates and the motion of the bolt is rectilinear transvery school ees and motion ⁴ Types of motion i. Identifies parts that move in a specific way in a technical object (rectilinear	slatio	ence n).	of m	necha	inical
1. Forc	otion (e.g. in a door lock, the lever rotates and the motion of the bolt is rectilinear transvery school ees and motion ⁴ Types of motion i. Identifies parts that move in a specific way in a technical object (rectilinear translation, rotation, helical)	slatio	ence n).	of m	necha	inical
1. Forc a. b.	otion (e.g. in a door lock, the lever rotates and the motion of the bolt is rectilinear transvery school es and motion ⁴ Types of motion i. Identifies parts that move in a specific way in a technical object (rectilinear translation, rotation, helical) Effects of a force i. Explains the effects of a force in a technical object (change in the motion of	slatio	ence n).	of m	necha	inical
1. Forc a. b.	otion (e.g. in a door lock, the lever rotates and the motion of the bolt is rectilinear transfery school es and motion ⁴ Types of motion i. Identifies parts that move in a specific way in a technical object (rectilinear translation, rotation, helical) Effects of a force i. Explains the effects of a force in a technical object (change in the motion of an object, distortion of a material)	slatio	ence n).	of m	necha	inical
1. Forc a. b.	otion (e.g. in a door lock, the lever rotates and the motion of the bolt is rectilinear transity school des and motion ⁴ Types of motion i. Identifies parts that move in a specific way in a technical object (rectilinear translation, rotation, helical) Effects of a force i. Explains the effects of a force in a technical object (change in the motion of an object, distortion of a material) Simple machines i. Identifies wheels, inclined planes and levers in simple technical objects	S →	ence n).	of m	necha	inical
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1. Forc a. b.	otion (e.g. in a door lock, the lever rotates and the motion of the bolt is rectilinear transity school es and motion ⁴ Types of motion i. Identifies parts that move in a specific way in a technical object (rectilinear translation, rotation, helical) Effects of a force i. Explains the effects of a force in a technical object (change in the motion of an object, distortion of a material) Simple machines i. Identifies wheels, inclined planes and levers in simple technical objects (e.g. a wheelbarrow is made up of a second-class lever and a wheel) ii. Describes qualitatively the mechanical advantages of different types of levers (first-class, second-class, third-class) in different applications	→ → → → →	* * *	A:	3T	SE
1. Forc a. b.	otion (e.g. in a door lock, the lever rotates and the motion of the bolt is rectilinear transity school des and motion ⁴ Types of motion i. Identifies parts that move in a specific way in a technical object (rectilinear translation, rotation, helical) Effects of a force i. Explains the effects of a force in a technical object (change in the motion of an object, distortion of a material) Simple machines i. Identifies wheels, inclined planes and levers in simple technical objects (e.g. a wheelbarrow is made up of a second-class lever and a wheel) ii. Describes qualitatively the mechanical advantages of different types of levers (first-class, second-class, third-class) in different applications	→ → → → →	* * *	A:	3T	SE
1. Forc a. b. c.	otion (e.g. in a door lock, the lever rotates and the motion of the bolt is rectilinear transfer school des and motion ⁴ Types of motion i. Identifies parts that move in a specific way in a technical object (rectilinear translation, rotation, helical) Effects of a force i. Explains the effects of a force in a technical object (change in the motion of an object, distortion of a material) Simple machines i. Identifies wheels, inclined planes and levers in simple technical objects (e.g. a wheelbarrow is made up of a second-class lever and a wheel) ii. Describes qualitatively the mechanical advantages of different types of levers (first-class, second-class, third-class) in different applications nnological systems System i. Identifies a system (set of connected elements that interact with each other) in	♦	* * *	A:	3T	SE
1. Forc a. b. c.	otion (e.g. in a door lock, the lever rotates and the motion of the bolt is rectilinear transported by school des and motion ⁴ Types of motion i. Identifies parts that move in a specific way in a technical object (rectilinear translation, rotation, helical) Effects of a force i. Explains the effects of a force in a technical object (change in the motion of an object, distortion of a material) Simple machines i. Identifies wheels, inclined planes and levers in simple technical objects (e.g. a wheelbarrow is made up of a second-class lever and a wheel) ii. Describes qualitatively the mechanical advantages of different types of levers (first-class, second-class, third-class) in different applications mological systems System i. Identifies a system (set of connected elements that interact with each other) in a technical object or technological application	→ → → → → → →	* * * * * * * * * * * * * * * * * * *	A:	3T	SE

b. (Components of a system					
	 Describes the role of the components of a technological system (e.g. explains the role of the parts of a lighting system) 	\rightarrow	*			
c. I	Energy transformations ⁵					
	i. Associates energy with radiation, heat or motion	\rightarrow	*			
	ii. Defines energy transformations	\rightarrow	*			
	iii. Identifies energy transformations in a technical object or technological system	\rightarrow	*			
3. Engine	e <mark>ering</mark>	S	T	A	ST	SE
a. I	Basic mechanical functions (links, guiding control)					
	i. Describes the role of links and guiding controls in a technical object	\rightarrow	*			
	ii. Identifies a guiding control in a technical object, as well as the related links (e.g. a pizza wheel is guided by a pivot, which links it to the handle)	\rightarrow	*			
b. (ypical mechanical links					
	i. Describes the advantages and disadvantages of different types of links			*		
	ii. Names the types of links used in a technical object (e.g. the spiral link between a jar and its lid)			*		
c. l	inking of mechanical parts					
	Describes the characteristics of the links in a technical object (direct or indirect, rigid or flexible, removable or permanent, partial or complete)				*	
	ii. Determines the desirable characteristics of links in the design of a technical object				*	
	iii. Judges the choice of assembly solutions in a technical object				*	
+	iv. Explains the purpose of limiting motion (degree of freedom) in a technical object (e.g. some hinges limit how far a cupboard door can open, preventing it from hitting the wall)				*	
d. (Typical functions					
	i. Defines the typical functions (linking, guiding, sealing, lubricating)			*		
	ii. Associates a typical function with certain parts of a technical object			*		
	iii. Explains the choice of a type of link in a technical object (e.g. using a screw makes it possible to attach and remove a battery case)			\rightarrow	*	
e. (Guiding controls					
	i. Explains the choice of a type of guiding control in a technical object (e.g. the slide guides a drawer and reduces friction)				*	
f. /	Adhesion and friction of parts					
*	 Describes the advantages and disadvantages of the adhesion and friction of parts in a technical object 				*	
g. I	Notion transmission systems					
	i. Identifies motion transmission systems in technical objects	\rightarrow	*			
h. (Function, components and use of motion transmission systems					
	i. Names motion transmission systems in technical objects (friction gears, pulleys and belt, gear assembly, sprocket wheels and chain, wheel and worm gear)			*		

	ii.	Describes the functions of the components of a motion transmission system (e.g. in a bicycle, the gear assembly on the crankset is the driving unit, the sprocket wheel on the rear wheel is the receiving unit, and the chain is the intermediate unit)			*		
	iii.	Describes the speed changes or reversibility of a motion transmission system (e.g. a sprocket wheel that is replaced by a smaller wheel or a wheel with fewer teeth increases rotation speed)			*		
i.	Con	struction and characteristics of motion transmission systems					
	i.	Explains the choice of a motion transmission system in a technical object (e.g. using a gear assembly rather than friction gears to get better engine torque and avoid slipping)				*	
j.	Moti	on transformation systems					
	i.	Identifies motion transformation systems in technical objects	\rightarrow	*			
k.	Fund	ction, components and use of motion transformation systems					
	i.	Names motion transformation systems in technical objects (e.g. screw gear system, cam and roller, connecting rod and crank, rack and pinion)			*		
	ii.	Describes the functions of the components of a motion transformation system (e.g. in a double-lever corkscrew, the pinion is the driving unit and the rack is the receiving unit)			*		
	iii.	Describes speed changes or the reversibility of a motion transformation system (e.g. the cam and roller is a nonreversible motion transformation system)			*		
l.	Con	struction and characteristics of motion transformation systems					
	i.	Explains the choice of a motion transformation system (screw gear system, cams, connecting rods, cranks, slides, rotating slider crank mechanisms, rack-and-pinion drive) in a technical object (e.g. most car jacks use a screw gear system rather than a rack-and-pinion system, because the force of the arm on the small crank provides more thrust and because, given that it is nonreversible, the system is safer)				*	
•	ii.	Distinguishes between cams and eccentrics				*	
m.	Spec	ed changes)					
	i.	Uses systems that allow for speed changes in the design of technical objects			*		
+	ii.	Explains speed changes in a technical object using the concepts of resisting torque and engine torque				*	
C. Elec	ctrica	al engineering	1	2	3	4	4
	descri	hool be energy transformations and recognize them in different devices. They describes into electricity (e.g. wind turbines transform wind energy into electricity).	e wa	ays o	of trar	nsforr	ming
	y sch	nool					
econdar							
	Pow	er supply					
		Defines power supply as the ability to generate electrical current			*		
	i.				★	*	
a.	i. ii.	Defines power supply as the ability to generate electrical current Determines the source of current in technical objects with an electrical circuit				*	
a.	i. ii.	Defines power supply as the ability to generate electrical current Determines the source of current in technical objects with an electrical circuit (e.g. chemical battery, solar cell, alternator, thermocouple, piezoelectric) ⁶				*	
	i. ii. Con	Defines power supply as the ability to generate electrical current Determines the source of current in technical objects with an electrical circuit (e.g. chemical battery, solar cell, alternator, thermocouple, piezoelectric) ⁶ duction, insulation and protection			→	*	

	iv.	Analyzes the factors that affect electrical conductivity (section, length, nature, temperature of conductor)				*	
+	V.	Uses the colour code to determine the electrical resistance of a resistor				*	
+	vi.	Describes the operation of a printed circuit				*	
C.	Cont	trol					
	i.	Defines control as the ability to control the travel of electrical current			*		
	ii.	Describes different types of switches (lever, pushbutton, flip-flop, magnetic control)			*		
+	iii.	Distinguishes between unipolar and bipolar switches				*	
+	iv.	Distinguishes between unidirectional and bidirectional switches				*	
d.	Tran	sformation of energy (electricity and light, heat, vibration, magnetism)					
	i.	Associates the transformation of energy with different components of a circuit (e.g. bulbs transform electrical energy into light and heat)				*	
	ii.	Describes the energy transformations that take place in electrical or electronic appliances (e.g. in a cell phone, electricity is transformed into light for the display and vibrations for the sound)				*	
e.	Othe	er functions					
*	i.	Describes the function of certain electronic components (condenser, diode, transistor, relay)				*	
D. Ma	terial	s	1	2	3	4	4
	descri	hool be the physical properties of certain materials.					
tudents (descri	hool be the physical properties of certain materials.	S	T.	AS	ST	SE
econda 1. Mate	descri	hool be the physical properties of certain materials. nool	S	ŝΤ	AS	ВТ	SE
econda 1. Mate	ry sch erial re	hool be the physical properties of certain materials. nool esources materials Associates raw materials with the unprocessed materials used in an industry	s	T *	AS	ST .	SE
tudents of econdarian decondarian decondar	ry sch erial re Raw i.	hool be the physical properties of certain materials. hool esources materials Associates raw materials with the unprocessed materials used in an industry (e.g. bauxite is the raw material used in aluminum smelters)		4	AS	ВТ	SE
tudents of econdarian decondarian decondar	ry sch erial re Raw i.	hool be the physical properties of certain materials. cool esources materials Associates raw materials with the unprocessed materials used in an industry (e.g. bauxite is the raw material used in aluminum smelters) erials Names the materials present in a technical object (e.g. a cooking pot is composed of two materials: a metal used to make the container and plastic		4	AS	ST	SE
tudents of econdarian decondarian decondar	ry scherial reRaw i. Mate	hool be the physical properties of certain materials. hool esources materials Associates raw materials with the unprocessed materials used in an industry (e.g. bauxite is the raw material used in aluminum smelters) erials Names the materials present in a technical object (e.g. a cooking pot is	→	*	AS	TE	SE
1. Mate	ry scherial reRaw i. Mate i.	hool be the physical properties of certain materials. nool esources materials Associates raw materials with the unprocessed materials used in an industry (e.g. bauxite is the raw material used in aluminum smelters) erials Names the materials present in a technical object (e.g. a cooking pot is composed of two materials: a metal used to make the container and plastic used to coat the handle) Determines the origins of the materials present in a technical object (animal,	→	*	AS	ST	SE
1. Mate	ry scherial reRaw i. Mate i.	be the physical properties of certain materials. cool esources materials Associates raw materials with the unprocessed materials used in an industry (e.g. bauxite is the raw material used in aluminum smelters) erials Names the materials present in a technical object (e.g. a cooking pot is composed of two materials: a metal used to make the container and plastic used to coat the handle) Determines the origins of the materials present in a technical object (animal, plant, mineral, wood)	→	*	ASS	ST	SE
1. Mate b.	ry scherial reRaw i. Mate i. Equi	be the physical properties of certain materials. Incol Esources materials Associates raw materials with the unprocessed materials used in an industry (e.g. bauxite is the raw material used in aluminum smelters) erials Names the materials present in a technical object (e.g. a cooking pot is composed of two materials: a metal used to make the container and plastic used to coat the handle) Determines the origins of the materials present in a technical object (animal, plant, mineral, wood) pment Defines tools and equipment as the elements needed to manufacture an	 → → → 	*	AS		SE
1. Mate a. b.	ry scherial reRaw i. Mate i. Equi	be the physical properties of certain materials. pool esources materials Associates raw materials with the unprocessed materials used in an industry (e.g. bauxite is the raw material used in aluminum smelters) erials Names the materials present in a technical object (e.g. a cooking pot is composed of two materials: a metal used to make the container and plastic used to coat the handle) Determines the origins of the materials present in a technical object (animal, plant, mineral, wood) pment Defines tools and equipment as the elements needed to manufacture an object (machining, control, assembly)	 → → → 	* *			
1. Mate a. b.	ry scherial reRaw i. Mate i. Equi	be the physical properties of certain materials. pool esources materials Associates raw materials with the unprocessed materials used in an industry (e.g. bauxite is the raw material used in aluminum smelters) erials Names the materials present in a technical object (e.g. a cooking pot is composed of two materials: a metal used to make the container and plastic used to coat the handle) Determines the origins of the materials present in a technical object (animal, plant, mineral, wood) pment Defines tools and equipment as the elements needed to manufacture an object (machining, control, assembly) all properties of materials	 → → → 	* *			

b. (Mechanical properties)					
 Describes the mechanical properties of different materials (hardness, ductility, elasticity, malleability, corrosion resistance) 			*		
c. Characteristics of mechanical properties					
 i. Explains the choice of a material based on its properties (e.g. the malleability of aluminum makes it useful for making thin-walled containers) 				*	
d. Types and properties					
i. Associates the use of different types of materials with their respective properties	es				
 Ferrous alloys (cast iron is harder than steel) 	Г		*		
 Nonferrous metals and alloys (the wire used in a dental appliance can be made of a nickel and titanium alloy, which has shape memory) 			*		
 Wood and modified wood (e.g. oak is used for flooring because it is a hard wood that isis shock- and wear-resistant) 			*		
 Plastics: thermoplastics (e.g. thermoplastics are used for prostheses because of their corrosion resistance and lightness) 			*		
 Plastics: thermosetting plastics (e.g. Bakelite is used to mould electrical parts because it is a good electrical insulator) 				*	
 Ceramics (e.g. ceramics are used in ovens because they are very hard and heat- and wear-resistant) 				*	
 Composites (e.g. carbon fibre is used for hockey sticks because of its hardness, resilience and lightness) 				*	
e. Cell					
i. Describes how a living cell can be considered a material (e.g. artificial skin is manufactured from human tissue to treat burns)			*		
ii. (Compares a cell to a technological system (overall function, inputs, outputs, processes, control)			*		
f. Modification of properties					
 Describes different treatments to prevent degradation of materials (e.g. metal plating, antirust treatments, painting) 				*	
g. Heat treatments					
 i. Defines heat treatments as ways of changing the properties of materials (quenching increases hardness but fragility as well) 				*	
E. Manufacturing	1	2	3	4	4
Elementary school Students are introduced to the design and construction of instruments, tools, machines, structur systems (e.g. water filtration), models (e.g. glider) and simple electrical circuits. They trace parts different materials using the appropriate tools. They use a variety of assembly methods (e.g. sc round-head fasteners, nuts) and tools to obtain an aesthetic finish.	s and	cut	them	out	
Secondary school					
a. Specifications					
 Defines specifications as a set of constraints associated with the design of a technical object 	\rightarrow	*			
 ii. Evaluates a prototype or technical object based on the environments described in the specifications (human, technical, industrial, economic, physical, environmental) 	\rightarrow	*			
b. Manufacturing process sheet					
Defines a manufacturing process sheet as a set of steps to follow to machine the parts that make up a technical object	\rightarrow	*			

	 Follows a process and assembly sheet to construct an object consisting of few components or to construct part of that object 	\rightarrow	*			
C.	Shaping					
	i. Machines and tools					
+	 Associates shaping processes with the types of materials used (e.g. injection blow moulding is used to shape plastics) 			*		
*	 Determines the appropriate shaping techniques based on the direct observation of technical objects (e.g. some table legs are turned on a lathe) 			*		
d.	Manufacturing					
	i. Roughing					
•	 Defines roughing as one of the first steps in the manufacturing process 			*		
	ii. Characteristics of laying out					
*	 Associates laying out with saving materials, shaping techniques and the types of materials used 			*		
	iii. Characteristics of drilling, tapping, threading and bending					
*	 Describes the characteristics of the tools needed to shape a material (e.g. the tip of a metal drill is conical, while that of a wood drill is double fluted) 				*	
e.	Measurement and inspection					
	i. Direct measurement					
*	 Explains the purpose of direct measurement (using a ruler) to control the machining of a part 			*		
*	 Explains the choice of the direct measurement instrument used (a vernier calliper is more precise than a ruler) 				*	
	ii. Control, shape and position (plane, section, angle)					
•	 Associates quality control techniques (indirect measurement) for materials and technical objects with the desired degree of precision (e.g. the shape of a musical instrument is validated using a three- dimensional digitizer to ensure the proper sound) 				*	
F. Bio	echnology	1	2	3	4	4
Elementa Students o	ry school o not address any concepts associated with biotechnology.					
Secondar	y school					
a.	Processes					
	i. Pasteurization			_	_	
	 Describes the pasteurization process 			*		
	 Describes the purpose of pasteurization (preservation of food and its nutritional properties) 			*		
	ii. Manufacture of vaccines					
	Describes the process for manufacturing vaccines			*		
	iii. Assisted reproduction					
	Describes different assisted-reproduction processes			*		

*	 Describes the purpose of artificial insemination (animal reproduction, an answer to human infertility, preservation of the gene pool, food self-sufficiency) 		*	
	iv. Cell cultures			
	 Names parameters to be controlled in the case of cultured cells (sources of mother cells, growth, preservation, characteristics of cell media, ethical standards) 		*	

- 1. See Techniques, Technology, Graphic communication (Techniques Technology, 1).
- 2. The progression of learning associated with these concepts is characterized by the increasing complexity of the objects to be represented.
- 3. See Techniques, Technology, Graphic communication, Using scales (Techniques Technology, 1, d).
- 4. This section is continued in *The Material World*, Cycle Two (MW, G).
- 5. For Secondary Cycle Two concepts related to *Energy transformations*, see *The Material World, Changes*, *Transformation of energy* (MW, B, 4).
- 6. The progression of learning associated with this concept is characterized by the increasing complexity of the objects to be studied.

Applied General Education Path

Techniques

The techniques listed below are divided into three categories, depending on whether they apply to science or technology or both. Many of them require the use of instruments and tools or chemicals. Safety in the workshop and laboratory should be a constant concern.

→ Student co	onstructs knowledge with teacher guidance.					
★ Student a	pplies knowledge by the end of the school year.		S	econ	dary	1
Student re	einvests knowledge.	_	_	Α.	ST	SE
echnology progra	ded by the symbol • indicate knowledge specific to the compulsory Applied Science and am. Most of these statements are, however, found in the progression of learning for the ental Science and Technology program.	Су	T cle ne	Су	cle vo	Cycle Two
A. Technol	ogy	1	2	3	4	4
simple drawings machines (e.g. I ntroduced to the e.g. screws, glu	ome symbols associated with motion and electrical and mechanical parts. They in a containing symbols. By designing technical objects, they become familiar with the ever, inclined plane, pulley, wheel). They trace parts and cut them out of difference safe use of tools (e.g. pliers, screwdriver, hammer, wrench, template) and difference, nails, round-head fasteners, nuts). They pay attention to finishing.	he us	se of ateria	simp lls. T	ole hey a	are
Secondary sch	ommunication ¹	S	Т	AS	ST	SE
·	g a technical drawing					0_
	Chooses the best view for an elevation drawing of a technical object	\rightarrow	*			
	Represents the visible edges using solid lines	\rightarrow	*			
iii.	Represents the hidden edges using dotted lines	\rightarrow	*			
iv.	Indicates the overall external dimensions of an object on a drawing	\rightarrow	*			
b. Rea	ding plans					
i.	Associates views with the sides of a technical object	\rightarrow	*			
ii.	Associates lines with the edges of a technical object	\rightarrow	*			
c. Drav	ving diagrams ²					
i.	Chooses the best view to describe a technical object	\rightarrow	\rightarrow	\rightarrow	*	
ii.	Uses different colours for each part of a technical object	\rightarrow	*			
iii.	Indicates all the information needed to explain the operation or construction of an object	\rightarrow	\rightarrow	\rightarrow	*	
d. Usin	g scales ³					
i.	Associates real measurements with each of the dimensions in a drawing	\rightarrow	*			
ii.	Reduces or multiplies the dimensions of a technical object based on the scale	\rightarrow	*			
iii.	Dimensions multiview orthogonal projections in accordance with the main			*		

dimensioning rules

e. Us	ing drawing instruments					
	i. Uses drawing instruments (e.g. ruler, square) to make diagrams	\rightarrow	*			
f. Co	nstructing a graph using instruments					
	 Uses instruments to construct a graph (e.g. multiview orthogonal projection, isometric representation, perspective drawing) 			\rightarrow	*	
g. Us	ing vector graphic software					
*	Uses vector graphic software to draw different diagrams in two and three dimensions (e.g. drawing toolbar in Word)			\rightarrow	*	
2. Manufa	cturing ⁴	S	Т	AS	ST	SE
a. Sa	fely using machines and tools ⁵					
	i. Uses tools safely (e.g. retractable utility knife, hammer, screwdriver, pliers)	\rightarrow	*			
	ii. Uses machine tools safely (band saw, drill, sander)			\rightarrow	*	
b. Me	easuring and laying out					
	i. Identifies the unit of measurement on the instrument	\rightarrow	*			
	ii. Positions the measuring instrument to obtain reliable reference points	\rightarrow	*			
i	ii. Adopts the appropriate position for reading an instrument	\rightarrow	*			
i	v. Marks the materials to be shaped using a pencil or punch	\rightarrow	*			
c. Ma	achining and forming					
	i. Chooses the appropriate materials, tools, techniques and processes	\rightarrow	*			
	ii. Draws the necessary reference lines	\rightarrow	*			
i	ii. Immobilizes the part to be formed	\rightarrow	*			
i	v. Forms the part in accordance with the steps in the following machining processes: sawing, drilling, sanding, filing	\rightarrow	*			
	v. Forms the part in accordance with the steps in the following machining processes: stripping, splicing, soldering			\rightarrow	*	
d. Fir	nishing					
	i. Sands the sides or deburrs the edges of each part after forming	\rightarrow	*			
	ii. Uses the appropriate finish (stain, paint)	\rightarrow	*			
i	ii. Grinds, polishes, hammers or chisels metal parts			\rightarrow	*	
e. As	sembling					
	i. Marks the references (holes, points or guidelines)	\rightarrow	*			
	ii. Immobilizes parts during gluing	\rightarrow	*			
i	ii. Drills to the diameter of the screws, nails or rivets used	\rightarrow	*			
i	v. Countersinks the openings for countersunk screws	\rightarrow	*			
f. As	sembling and disassembling					
	i. Identifies and gathers the parts and hardware	\rightarrow	*			
	ii. Chooses the appropriate tools	\rightarrow	*			

	iii.	For disassembly, numbers and records the location of the parts	\rightarrow	*			
	iv.	In the case of electrical circuits, identifies and gathers the electrical components			\rightarrow	*	
*	V.	In the case of electronic circuits, identifies and gathers the electronic components			\rightarrow	*	
	vi.	Chooses and places the electrical components in sequence based on the circuit diagram			\rightarrow	*	
+	vii.	Chooses and places the electronic components in sequence based on the circuit diagram			\rightarrow	*	
	viii.	Connects the components using wire, connectors or solders			\rightarrow	*	
+	ix.	Connects the components on a printed circuit board			\rightarrow	*	
+	X.	Uses a desoldering bulb to remove a solder			\rightarrow	*	
g.	Perf	orming verification and control tasks					
	i.	Evaluates the dimensions of a part during and after construction using a ruler			\rightarrow	*	
	ii.	Compares the real dimensions of a part with the specifications (e.g. draft, drawing, technical sheet)			\rightarrow	*	
	iii.	Uses a template to verify the conformity of a part			\rightarrow	*	
	iv.	Evaluates the dimensions of a part during and after construction using vernier callipers				*	
h.	Maki	ing a part					
	i.	Makes a part using the appropriate techniques			\rightarrow	*	
B. Sci	ence		1	2	3	*	4
lementa tudents l	ence ary sc pecom ts (rule	hool ne familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduated cylinder, balance, thermometer, chronometer).			3	4	
lementa itudents l nstrumen	ence ary sc pecom ts (rule ry sch	hool ne familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduated cylinder, balance, thermometer, chronometer).			3	4	
lementa itudents l nstrumen	ence pecom ts (rule ry sch	hool ne familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduated cylinder, balance, thermometer, chronometer).			3	4	
lementa itudents l nstrumen	ence ary sc pecom ts (rule ry sch Safe i.	hool ne familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduated cylinder, balance, thermometer, chronometer). hool ly using laboratory materials and equipment ⁶ Uses laboratory materials and equipment safely (e.g. allows hotplate to cool,	rs) ar		3	4	
lementa tudents l estrumen econdai	ence ary sc pecom ts (rule ry sch Safe i.	hool ne familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduated cylinder, balance, thermometer, chronometer). hool ly using laboratory materials and equipment ⁶ Uses laboratory materials and equipment safely (e.g. allows hotplate to cool, uses beaker tongs)	rs) ar		3	4	
lementa tudents l estrumen econdai	ence ary sc pecom ts (rule ry sch Safe i. Sepa	hool ne familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduated cylinder, balance, thermometer, chronometer). hool ly using laboratory materials and equipment ⁶ Uses laboratory materials and equipment safely (e.g. allows hotplate to cool, uses beaker tongs) Handles chemicals safely (e.g. uses a spatula and pipette filler)	rs) ar		3	4	
lementa tudents l estrumen econdai	ence ary sc pecom ts (rule ry sch Safe i. Sepa i.	hool ne familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduated cylinder, balance, thermometer, chronometer). hool ly using laboratory materials and equipment ⁶ Uses laboratory materials and equipment safely (e.g. allows hotplate to cool, uses beaker tongs) Handles chemicals safely (e.g. uses a spatula and pipette filler) arating mixtures	⇒ →	→ →	3	4	
lementa tudents l estrumen econdai	ence ary sc pecom ts (rule ry sch ii. Sepa i. iii.	hool ne familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduated cylinder, balance, thermometer, chronometer). nool lly using laboratory materials and equipment ⁶ Uses laboratory materials and equipment safely (e.g. allows hotplate to cool, uses beaker tongs) Handles chemicals safely (e.g. uses a spatula and pipette filler) arating mixtures Separates heterogeneous mixtures using sedimentation and decantation	\Rightarrow \Rightarrow	→ →	3	4	
lementa itudents l nstrumen econdai a.	ence ary sc pecom ts (rule ry sch Safe i. Sepa i. ii.	hool ne familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduated cylinder, balance, thermometer, chronometer). nool lly using laboratory materials and equipment ⁶ Uses laboratory materials and equipment safely (e.g. allows hotplate to cool, uses beaker tongs) Handles chemicals safely (e.g. uses a spatula and pipette filler) arating mixtures Separates heterogeneous mixtures using sedimentation and decantation Separates heterogeneous mixtures using filtration	\Rightarrow \Rightarrow \Rightarrow	→ → ★	3	4	
lementa itudents l nstrumen econdai a.	ence ary sc pecom ts (rule ry sch ii. Sepa ii. iii. Desi	hool ne familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduated cylinder, balance, thermometer, chronometer). nool lly using laboratory materials and equipment ⁶ Uses laboratory materials and equipment safely (e.g. allows hotplate to cool, uses beaker tongs) Handles chemicals safely (e.g. uses a spatula and pipette filler) arating mixtures Separates heterogeneous mixtures using sedimentation and decantation Separates heterogeneous mixtures using filtration Separates different aqueous solutions using evaporation or distillation	\Rightarrow \Rightarrow \Rightarrow	→ → ★	3	4	
econdar a. b.	ence ary sc pecom ts (rule ry sch ii. Sepa ii. iii. Desi i.	hool ne familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduated cylinder, balance, thermometer, chronometer). In using laboratory materials and equipment (e.g. allows hotplate to cool, uses laboratory materials and equipment safely (e.g. allows hotplate to cool, uses beaker tongs) Handles chemicals safely (e.g. uses a spatula and pipette filler) arating mixtures Separates heterogeneous mixtures using sedimentation and decantation Separates heterogeneous mixtures using filtration Separates different aqueous solutions using evaporation or distillation gning and creating an environment Uses environmental design and construction techniques that respect the	$\begin{array}{c} \Rightarrow \\ \Rightarrow \\ \Rightarrow \\ \end{array}$	→ → ★ ★	3	4	
econdar a. b.	ence ary sc Decompts (rule ry sch ii. Sepa ii. Desi i. Usin	hool ne familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduated cylinder, balance, thermometer, chronometer). hool Ity using laboratory materials and equipment ⁶ Uses laboratory materials and equipment safely (e.g. allows hotplate to cool, uses beaker tongs) Handles chemicals safely (e.g. uses a spatula and pipette filler) arating mixtures Separates heterogeneous mixtures using sedimentation and decantation Separates heterogeneous mixtures using filtration Separates different aqueous solutions using evaporation or distillation gning and creating an environment Uses environmental design and construction techniques that respect the characteristics of the habitat (e.g. terrarium, aquarium, composting medium)	$\begin{array}{c} \Rightarrow \\ \Rightarrow \\ \Rightarrow \\ \end{array}$	→ → ★ ★	3	4	
econdar a. b.	ence ary sc Decompts (rule ry sch Safe i. Sepa i. iii. Desi i. Usin i.	hool ne familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduated cylinder, balance, thermometer, chronometer). Inool If using laboratory materials and equipment ⁶ Uses laboratory materials and equipment safely (e.g. allows hotplate to cool, uses beaker tongs) Handles chemicals safely (e.g. uses a spatula and pipette filler) arating mixtures Separates heterogeneous mixtures using sedimentation and decantation Separates heterogeneous mixtures using filtration Separates different aqueous solutions using evaporation or distillation gning and creating an environment Uses environmental design and construction techniques that respect the characteristics of the habitat (e.g. terrarium, aquarium, composting medium) g measuring instruments	$\begin{array}{c} \Rightarrow \\ \Rightarrow \\ \Rightarrow \\ \end{array}$	→ → ★ ★ ★	3	4	
econdar a. b.	ence ary sc pecom ts (rule ry sch Safe i. Sepa i. Usin i.	hool le familiar with the use of observational instruments (magnifying glass, binocular er, eyedropper, graduated cylinder, balance, thermometer, chronometer). ly using laboratory materials and equipment ⁶ Uses laboratory materials and equipment safely (e.g. allows hotplate to cool, uses beaker tongs) Handles chemicals safely (e.g. uses a spatula and pipette filler) arating mixtures Separates heterogeneous mixtures using sedimentation and decantation Separates heterogeneous mixtures using filtration Separates different aqueous solutions using evaporation or distillation gning and creating an environment Uses environmental design and construction techniques that respect the characteristics of the habitat (e.g. terrarium, aquarium, composting medium) g measuring instruments Adopts the appropriate position for reading an instrument	$\begin{array}{c} \Rightarrow \\ \Rightarrow \\ \Rightarrow \\ \end{array}$	→ → ★ ★ ★	3	4	

	iv. Measures the volume of an insoluble solid using water displacement	\rightarrow	*			
	v. Measures temperature using a graduated thermometer	\rightarrow	*			
	vi. Uses measuring instruments appropriately (e.g. ammeter, volumetric flask)			\rightarrow	*	
	vii. Uses measuring instruments appropriately (e.g. ammeter, volumetric flask)					*
e. l	Using observational instruments					
	 Uses observational instruments appropriately (e.g. magnifying glass, stereomicroscope, microscope) 	\rightarrow	*			
f. F	Preparing solutions					
	i. Prepares an aqueous solution of a specific concentration given a solid solu	te		*		
	 ii. Prepares an aqueous solution of a specific concentration given a concentrated aqueous solution 			*		
g. C	Collecting samples					
	 Collects samples appropriately (e.g. sterilizes the container, uses a spatula refrigerates the sample) 	,		\rightarrow	*	
Tech		, 1	2	→ 3	4	4
	refrigerates the sample)		2			4
	refrigerates the sample) nniques common to Science and Technology	1	2			4
	refrigerates the sample) nniques common to Science and Technology Verifying the repeatability, accuracy and sensitivity of measuring instruments i. Takes the same measurement several times to check the repeatability of the	1	2			
	refrigerates the sample) nniques common to Science and Technology Verifying the repeatability, accuracy and sensitivity of measuring instruments i. Takes the same measurement several times to check the repeatability of the instrument used ii. Carries out the required operations to ensure the accuracy of a measuring instrument (e.g. cleans and calibrates a balance, dries out a graduated)	e e	2			*
a. \	refrigerates the sample) nniques common to Science and Technology Verifying the repeatability, accuracy and sensitivity of measuring instruments i. Takes the same measurement several times to check the repeatability of the instrument used ii. Carries out the required operations to ensure the accuracy of a measuring instrument (e.g. cleans and calibrates a balance, dries out a graduated cylinder, rinses and calibrates a pH-meter) iii. Chooses a measuring instrument by taking into account the sensitivity of the instrument (e.g. uses a 25-mL graduated cylinder rather than a 100-mL one	e e	2			*
a. \	refrigerates the sample) nniques common to Science and Technology Verifying the repeatability, accuracy and sensitivity of measuring instruments i. Takes the same measurement several times to check the repeatability of the instrument used ii. Carries out the required operations to ensure the accuracy of a measuring instrument (e.g. cleans and calibrates a balance, dries out a graduated cylinder, rinses and calibrates a pH-meter) iii. Chooses a measuring instrument by taking into account the sensitivity of the instrument (e.g. uses a 25-mL graduated cylinder rather than a 100-mL one measure out 18 mL of water	e e e e to	2			*
a. \	refrigerates the sample) Iniques common to Science and Technology Verifying the repeatability, accuracy and sensitivity of measuring instruments i. Takes the same measurement several times to check the repeatability of the instrument used ii. Carries out the required operations to ensure the accuracy of a measuring instrument (e.g. cleans and calibrates a balance, dries out a graduated cylinder, rinses and calibrates a pH-meter) iii. Chooses a measuring instrument by taking into account the sensitivity of the instrument (e.g. uses a 25-mL graduated cylinder rather than a 100-mL one measure out 18 mL of water Interpreting the results of measurement i. Determines the error attributable to a measuring instrument (e.g. the error in measurement made using a graduated cylinder is provided by the	e e e e to	2			*
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- 1. See The Technological World, Graphical language (TW, A).
- 2. The progression of learning associated with these techniques is characterized by the increasing complexity of the objects to be represented.
- 3. See The Technological World, Graphical language, Scales (TW, A, g).
- 4. See The Technological World, Manufacturing, Shaping, Manifacturing, Measurement and inspection (TW, E, c-d-e).
- 5. When the teacher introduces a new technique, he or she should explain the related safety rules and repeat them often. After several practice sessions, students should apply the rules without being reminded.
- 6. When the teacher introduces a new technique, he or she should explain the related safety rules and repeat them often. After several practice sessions, students should apply the rules without being reminded.

Applied General Education Path

Strategies

The strategies listed below are fundamental to the approaches used in science and technology. They can be applied in a variety of increasingly complex contexts and are therefore inclusive. Thus, students build on the strategies they learned in elementary school. New strategies are added, including analytical strategies, which are adapted to students' level of cognitive development.

Student applies knowledge by the end of the school year. Student reinvests knowledge. E: The letter "E" indicates that students were introduced to this strategy in elementary school. A. Exploration strategies 1 2 3 4 1. Studying a problem or a phenomenon from different points of view (e.g. social, environmental, historical, economic) 2. Distinguishing between the different types of information useful for solving the problem 3. Referring to similar problems that have already been solved 4. Becoming aware of his or her previous representations E 5. Drawing a diagram for the problem or illustrating it E ST Cycle One AST Cycle One F Cycle One Two AST Cycle One Two AST Cycle One Two	SE Cycle Two
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5. Drawing a diagram for the problem or illustrating it	
6. Formulating questions	
7. Putting forward hypotheses (e.g. individually, in teams, as a class)	
8. Exploring various ways of solving the problem	
9. Anticipating the results of his or her approach	
10. Imagining solutions to a problem in light of his or her explanations	
11. Taking into account the constraints involved in solving a problem or making an object (e.g. specifications, available resources, time allotted)	
12. Examining his or her mistakes in order to identify their source	
13. Using different types of reasoning (e.g. induction, deduction, inference, comparison, classification)	
14. Using empirical approaches (e.g. trial and error, analysis, exploration using one's senses)	
15. Ensuring that the procedure is appropriate and safe and making the necessary adjustments	
16. Collecting as much scientific, technological and contextual information as possible to define a problem or predict patterns	
17. Generalizing on the basis of several structurally similar cases → ★	
18. Developing various scenarios → ★	
19. Considering various points of view on scientific or technological issues → ★	

B. Instrumentation strategies		1	2	3	4	4
 Using different sources of information (e.g. books, newspapers, Web sites, magazines, experts) 	Е					
2. Validating sources of information	Е					
 Using technical design to illustrate a solution (e.g. diagrams, sketches, technical drawings) 	Е					
 Using different tools for recording information (e.g. diagrams, notes, graphs, procedures, logbook) 	Е					
5. Using a variety of observational techniques and tools	Е					
6. Selecting suitable techniques or tools for observation		\rightarrow	*			
C. Analytical strategies		1	2	3	4	4
 Identifying the constraints and important elements related to the problem-solving situation 		\rightarrow	*			
2. Dividing a complex problem into simpler subproblems		\rightarrow	*			
 Using different types of reasoning (e.g. inductive and deductive reasoning, comparison, classification, prioritization) in order to process information 		\rightarrow	*			
Reasoning by analogy in order to process information and adapt scientific and technological knowledge				\rightarrow	*	
 Selecting relevant criteria to help him or her determine where he or she stands on a scientific or technological issue 				\rightarrow	*	
D. Communication strategies		1	2	3	4	4
 Using different means of communication to propose explanations or solutions (e.g. oral presentation, written presentation, procedure) 	Е					
2. Organizing information for a presentation (e.g. tables, diagrams, graphs)	Е					
Exchanging information	Е					
 Comparing different possible explanations for or solutions to a problem in order to assess their relevance (e.g. full-group discussion) 	Е					
Using tools to display information in various formats (e.g. data tables, graphs, diagrams)		\rightarrow	*			